

Chemical Propulsion Systems

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Space for Education, Education for Space
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Specialized lectures

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2. Chemical Rocket Propulsion
3. Performance of Rocket Vehicle

1. Fluid Flow and Thermodynamics

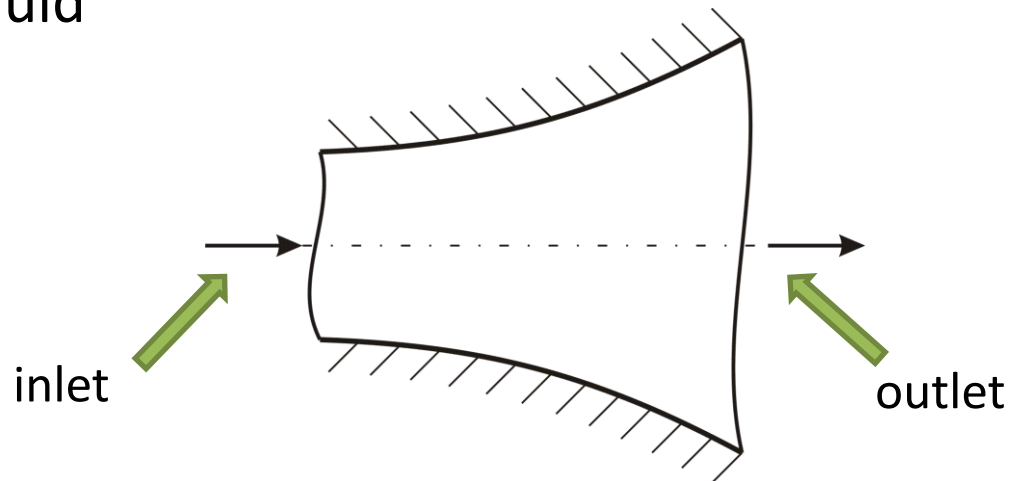
- Introduction
- Fundamental equations
- Thermodynamics of gases
- Speed of sound
- Isentropic flow
- Nozzle fluid flow

1. Fluid Flow and Thermodynamics

Introduction

- Fluid flow:
 - naturally three dimensional, but in some special cases can be considered as one dimensional or quasi-one dimension
 - fluid can be considered according to:
 - steady-state VS transient
 - turbulent VS laminar
 - inviscid VS viscous fluid

quasi-one dimension
fluid flow

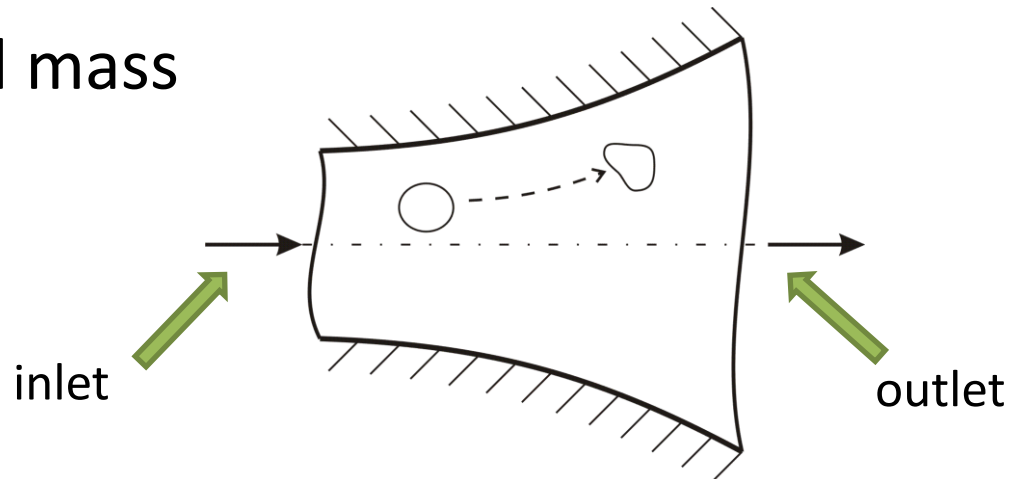


1. Fluid Flow and Thermodynamics

Introduction

- System or Control Mass (CM):
 - is a collection of matter of fixed identity
 - it may be considered enclosed by an invisible, massless, flexible surface through which no matter can pass
 - the boundary of the system may change position, size, and shape
 - is also called control mass

moving of control
mass

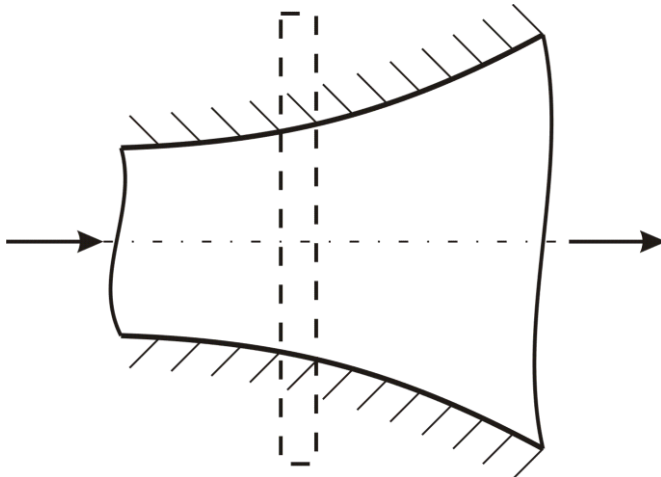


1. Fluid Flow and Thermodynamics

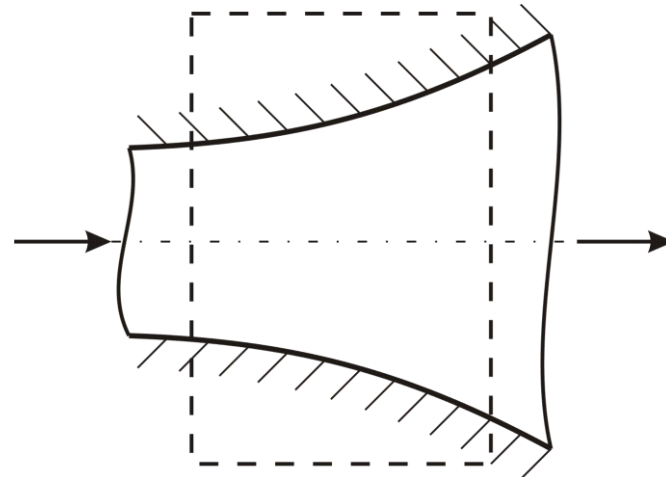
Introduction

- Control Volume (CV):
 - is arbitrary volume fixed to the coordinate system (stationary or moving)
 - bounded by control surface (CS) through which fluid may pass, CV can have differential or finite size

differential CV



finite CV



1. Fluid Flow and Thermodynamics

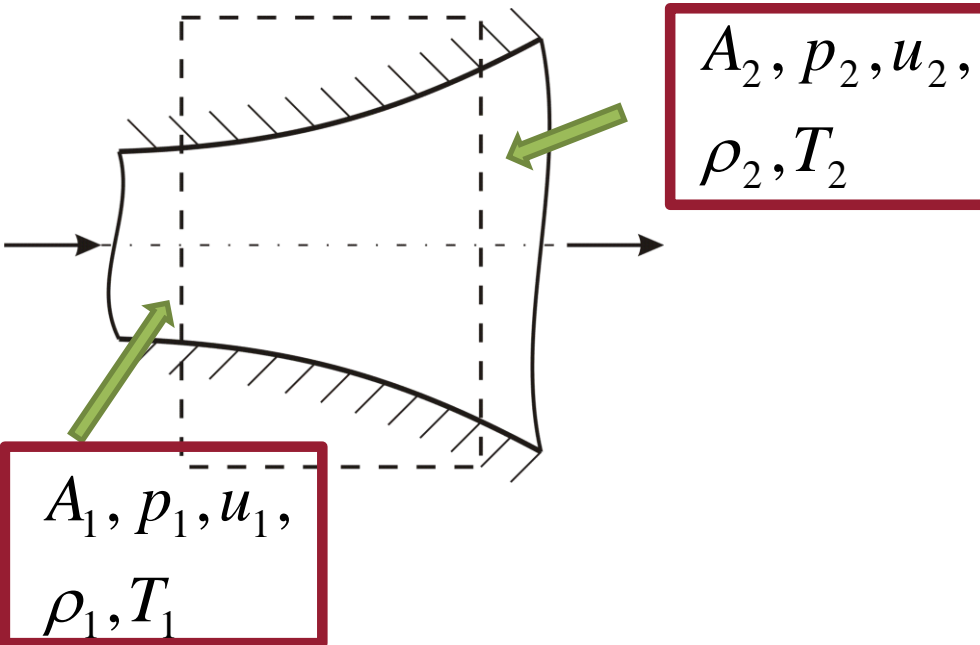
Fundamental equations

- There are 4 fundamental equations, which must be considered:
 - Continuity equation
 - Momentum equation
 - Energy equation
 - Entropy equation

1. Fluid Flow and Thermodynamics

Fundamental equations

- Continuity equation



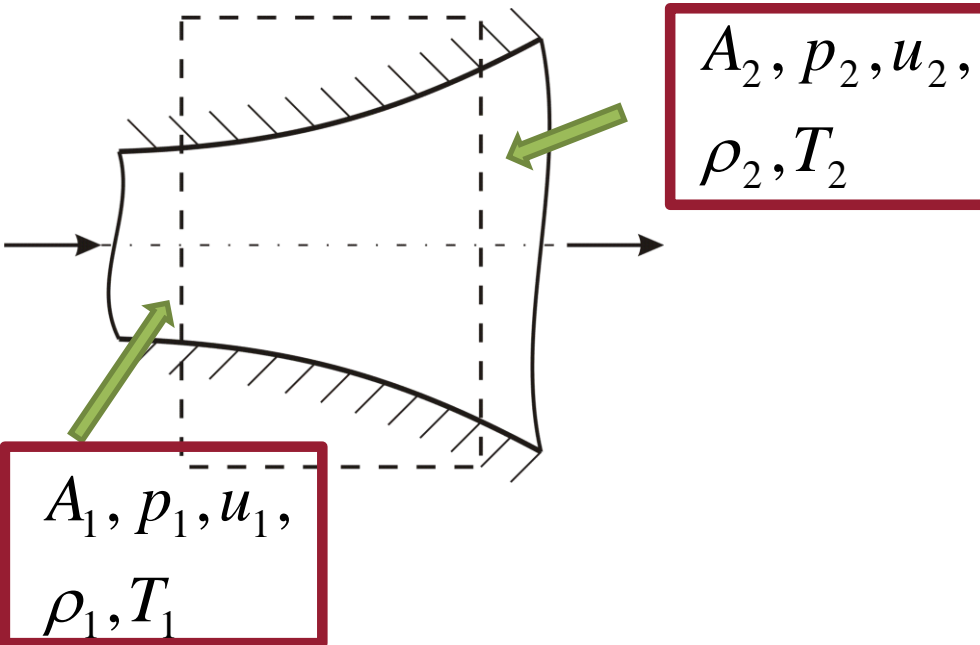
inviscid steady-state flow

$$\left[\begin{array}{c} \text{Rate mass} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate mass} \\ \text{enters CV} \end{array} \right] = 0$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Continuity equation



inviscid steady-state flow

↓

$$\left[\begin{array}{c} \text{Rate mass} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate mass} \\ \text{enters CV} \end{array} \right] = 0$$

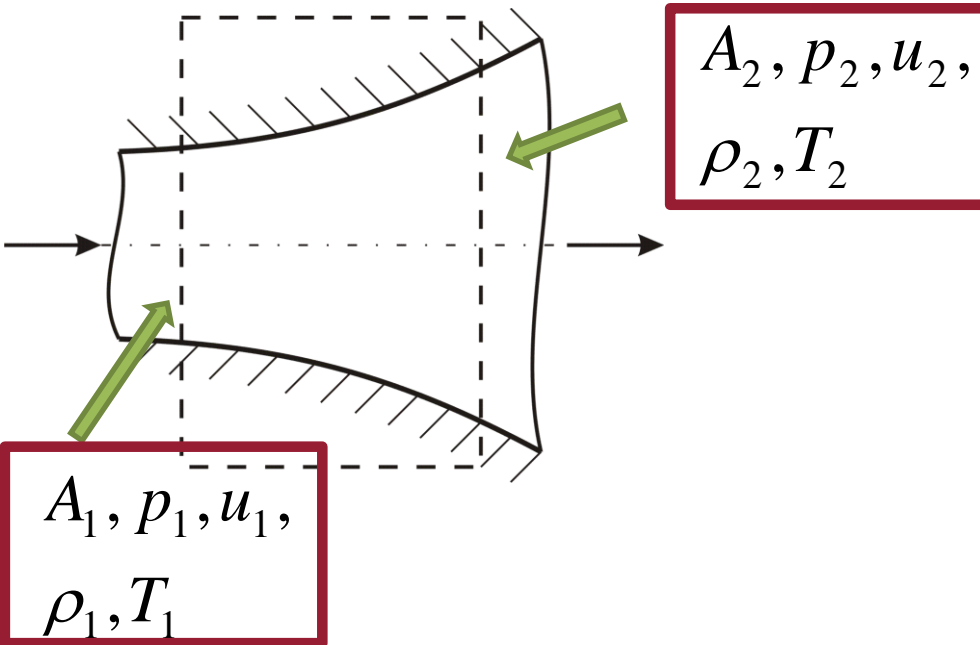
↓ ↓

$$A_2 \rho_2 u_2 \qquad A_1 \rho_1 u_1$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Continuity equation



inviscid steady-state flow

$$\left[\begin{array}{c} \text{Rate mass} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate mass} \\ \text{enters CV} \end{array} \right] = 0$$

$$A_2 \rho_2 u_2$$

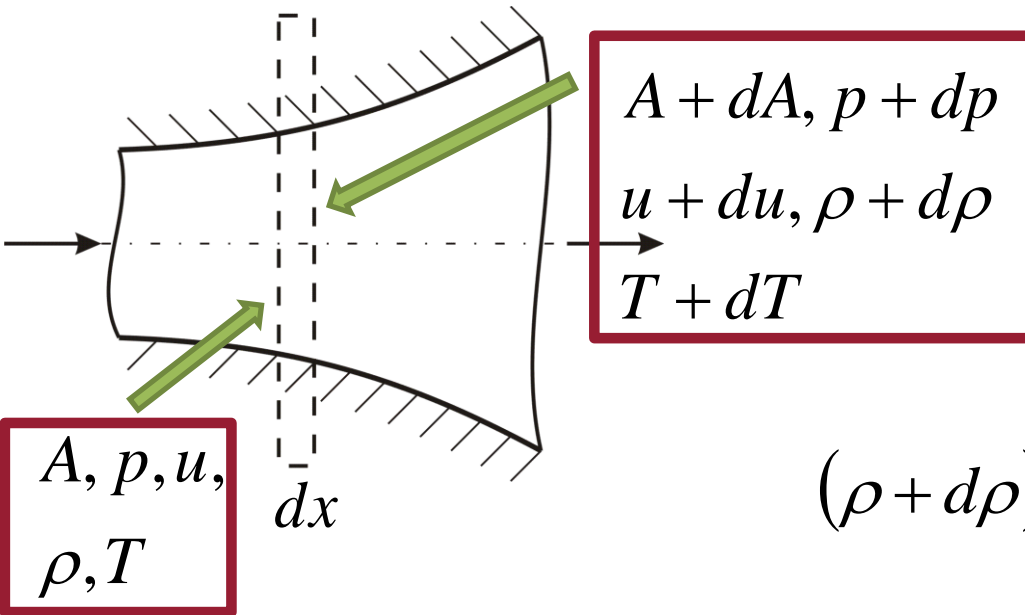
$$A_1 \rho_1 u_1$$

$$A_2 \rho_2 u_2 - A_1 \rho_1 u_1 = 0$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Continuity equation



inviscid steady-state flow

$$\left[\begin{array}{c} \text{Rate mass} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate mass} \\ \text{enters CV} \end{array} \right] = 0$$

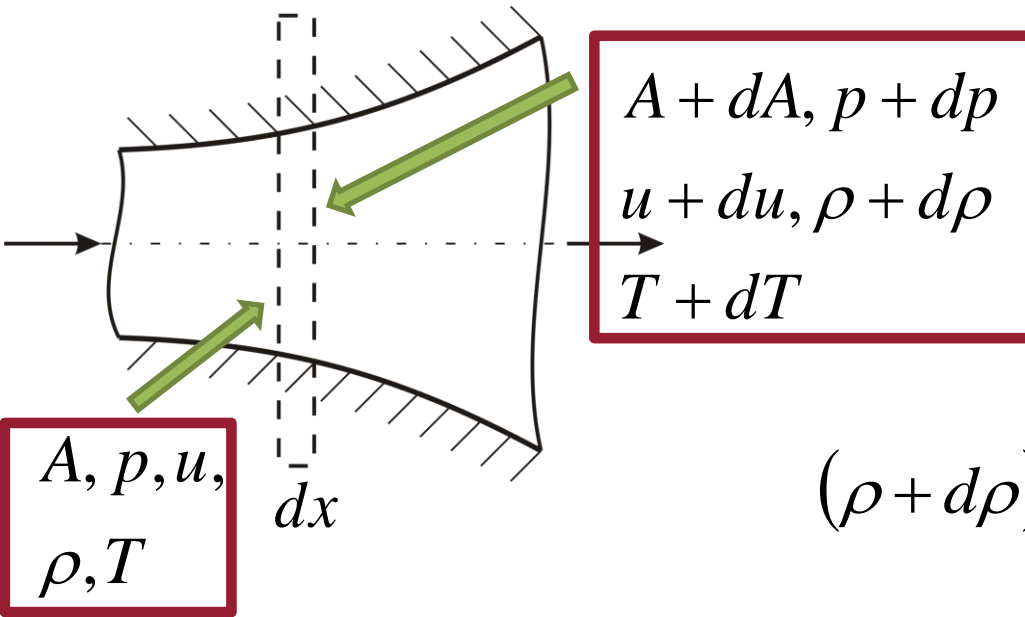
$$(\rho + d\rho)(u + du)(A + dA)$$

$$\rho u A$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Continuity equation



inviscid steady-state flow

$$\left[\begin{array}{c} \text{Rate mass} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate mass} \\ \text{enters CV} \end{array} \right] = 0$$

$$(\rho + d\rho)(u + du)(A + dA)$$

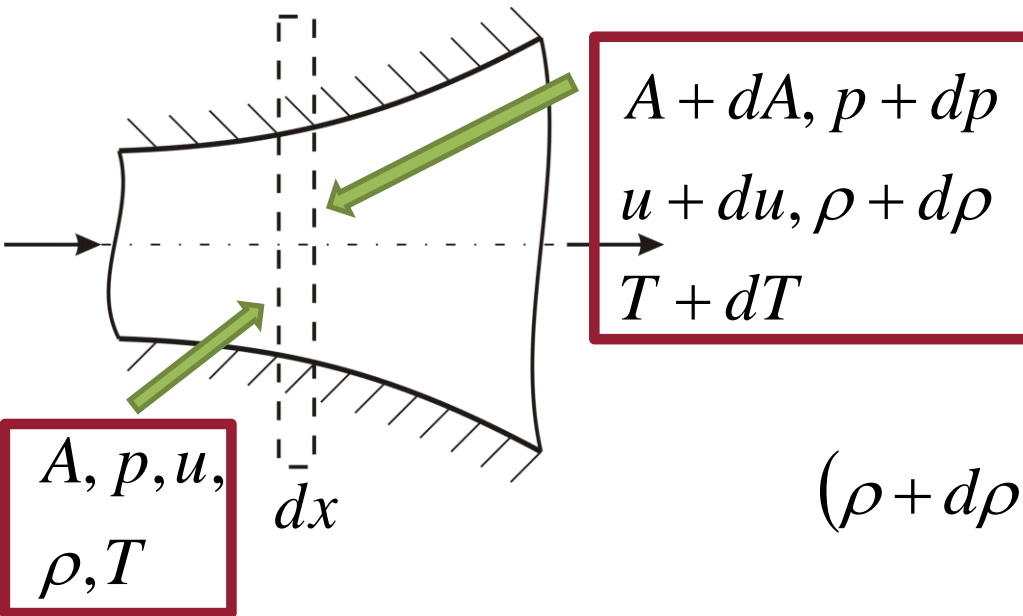
$$\rho u A$$

$$\rho u dA + \rho A du + u A d\rho = 0$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Continuity equation



inviscid steady-state flow

$$\left[\text{Rate mass} \right]_{\text{leaves CV}} - \left[\text{Rate mass} \right]_{\text{enters CV}} = 0$$

$$(\rho + d\rho)(u + du)(A + dA) \quad \rho u A$$

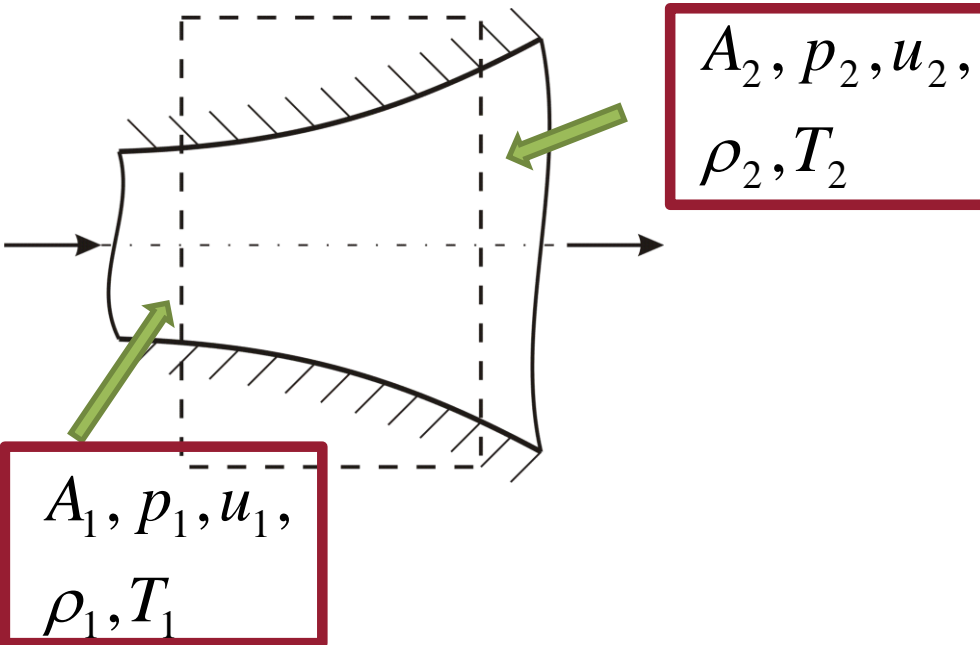
$$\rho u dA + \rho A du + u A d\rho = 0$$

$$\frac{d\rho}{\rho} + \frac{du}{u} + \frac{dA}{A} = 0$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation



inviscid steady-state flow

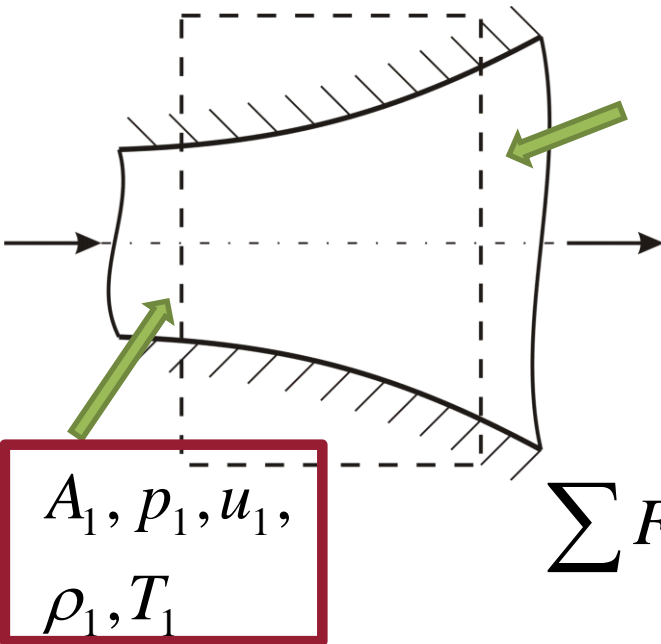
$$\left[\begin{array}{l} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow



$A_2, p_2, u_2,$
 ρ_2, T_2

$A_1, p_1, u_1,$
 ρ_1, T_1

$$\left[\begin{array}{c} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

$$\sum F = p_1 A_1 - p_2 A_2 + \int_{A_1}^{A_2} p dA$$

$$A_2 \rho_2 u_2^2$$

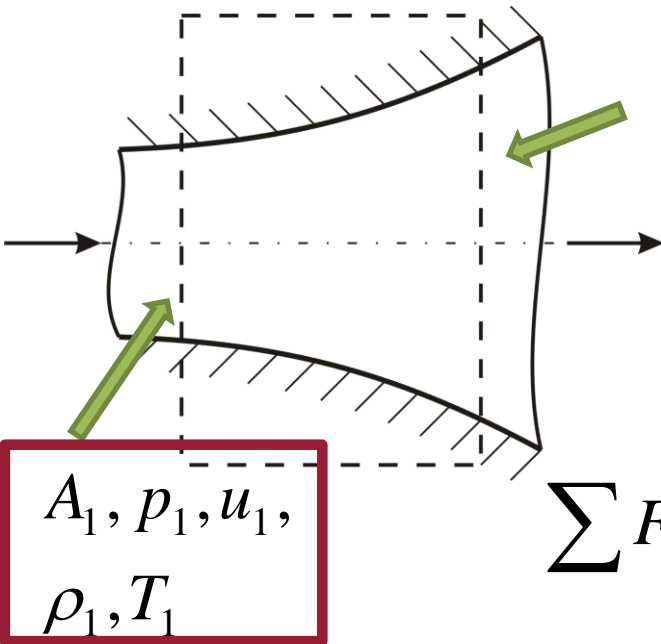
$$A_1 \rho_1 u_1^2$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow



$$A_2, p_2, u_2, \rho_2, T_2$$

$$A_1, p_1, u_1, \rho_1, T_1$$

$$\left[\begin{array}{c} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

$$\sum F = p_1 A_1 - p_2 A_2 + \int_{A_1}^{A_2} p dA$$

$$A_2 \rho_2 u_2^2$$

$$A_1 \rho_1 u_1^2$$

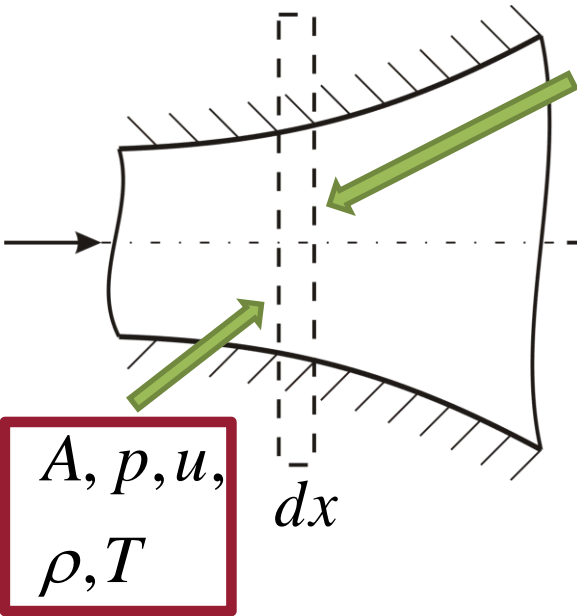
$$p_1 A_1 - p_2 A_2 + \int_{A_1}^{A_2} p dA = A_2 \rho_2 u_2^2 - A_1 \rho_1 u_1^2$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow



$$\begin{aligned} &A + dA, p + dp \\ &u + du, \rho + d\rho \\ &T + dT \end{aligned}$$

$$\left[\begin{array}{c} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

$$(\rho + d\rho)(u + du)^2 (A + dA)$$

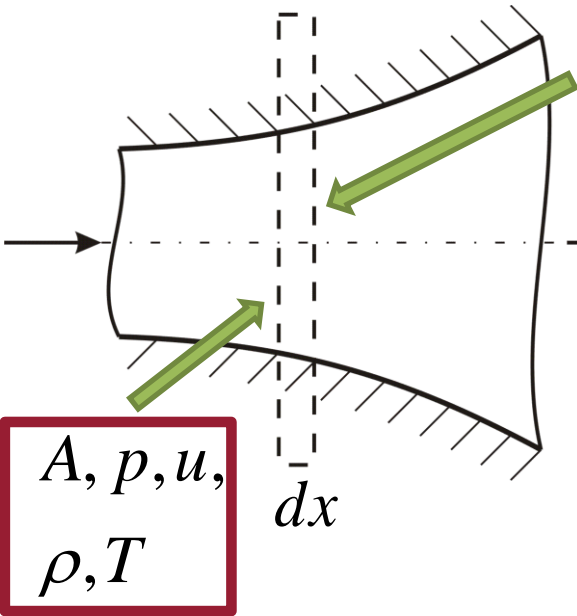
$$\rho u^2 A$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow



$$\begin{aligned} &A + dA, p + dp \\ &u + du, \rho + d\rho \\ &T + dT \end{aligned}$$

$$\left[\begin{array}{c} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

$$(\rho + d\rho)(u + du)^2 (A + dA)$$

$$\rho u^2 A$$

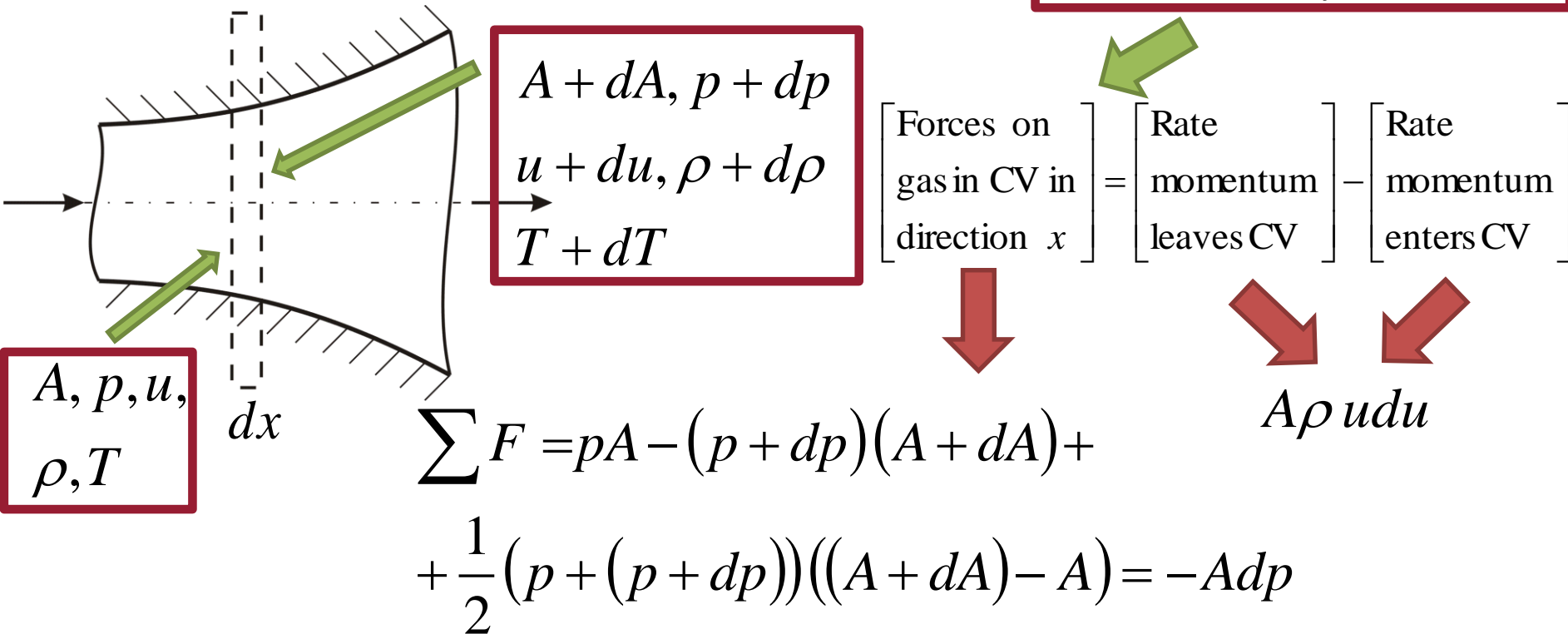
$$A \rho u du$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow

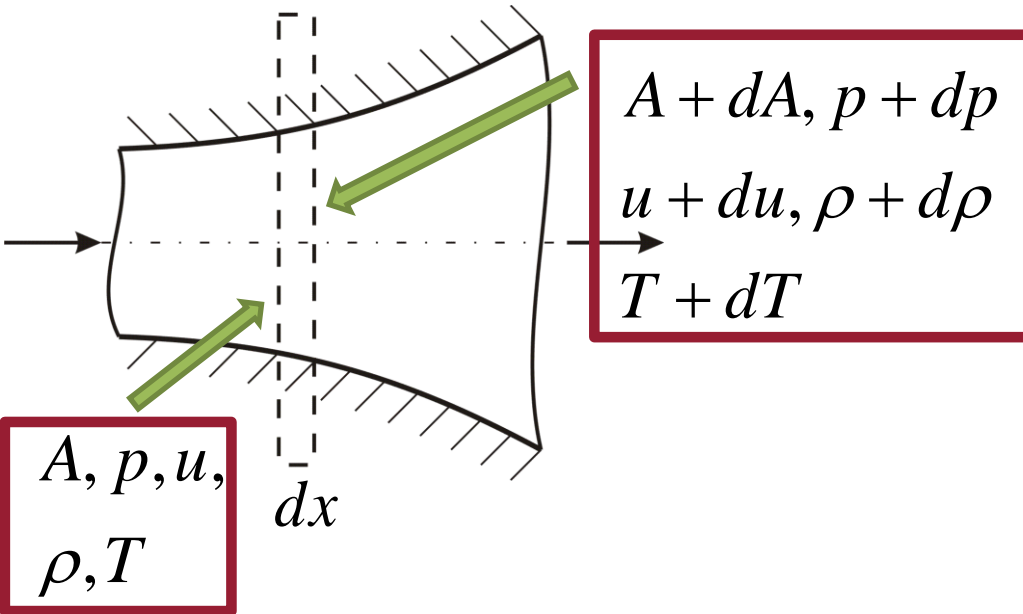


1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow



$$\left[\begin{array}{c} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

$$-Adp$$

$$A\rho u du$$

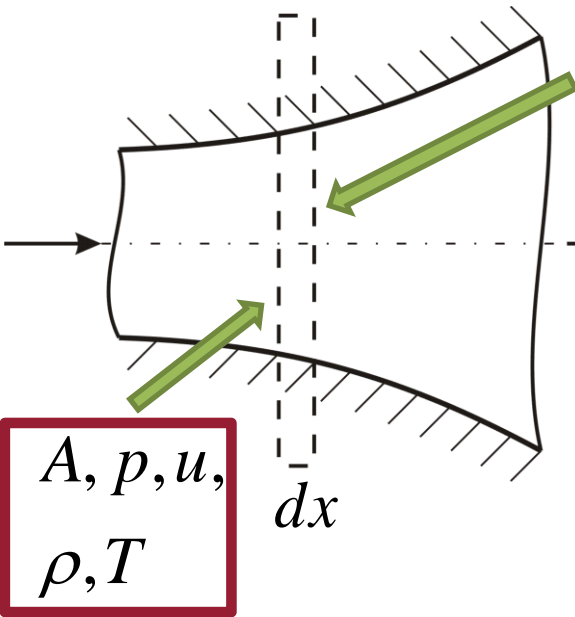
$$-Adp = A\rho u du$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Momentum equation

inviscid steady-state flow



$$\begin{aligned} &A + dA, p + dp \\ &u + du, \rho + d\rho \\ &T + dT \end{aligned}$$

$$\left[\begin{array}{l} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

$$-Adp$$

$$A\rho u du$$

$$-\frac{dp}{\rho} = u du$$

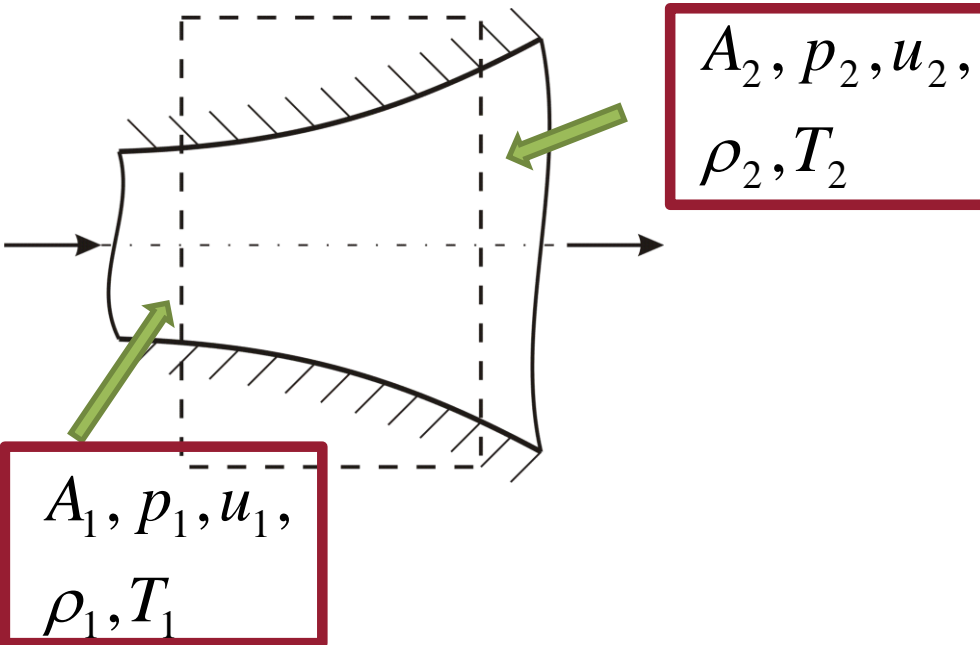
Euler's equation

$$-Adp = A\rho u du$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation



inviscid steady-state flow

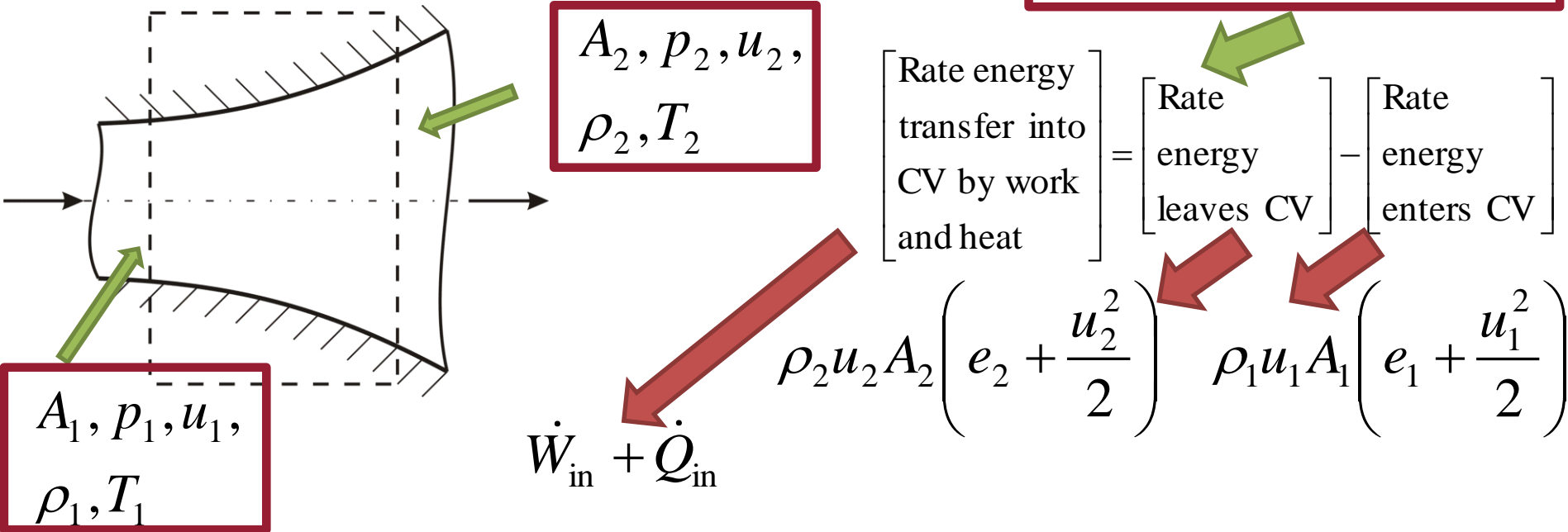
$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow

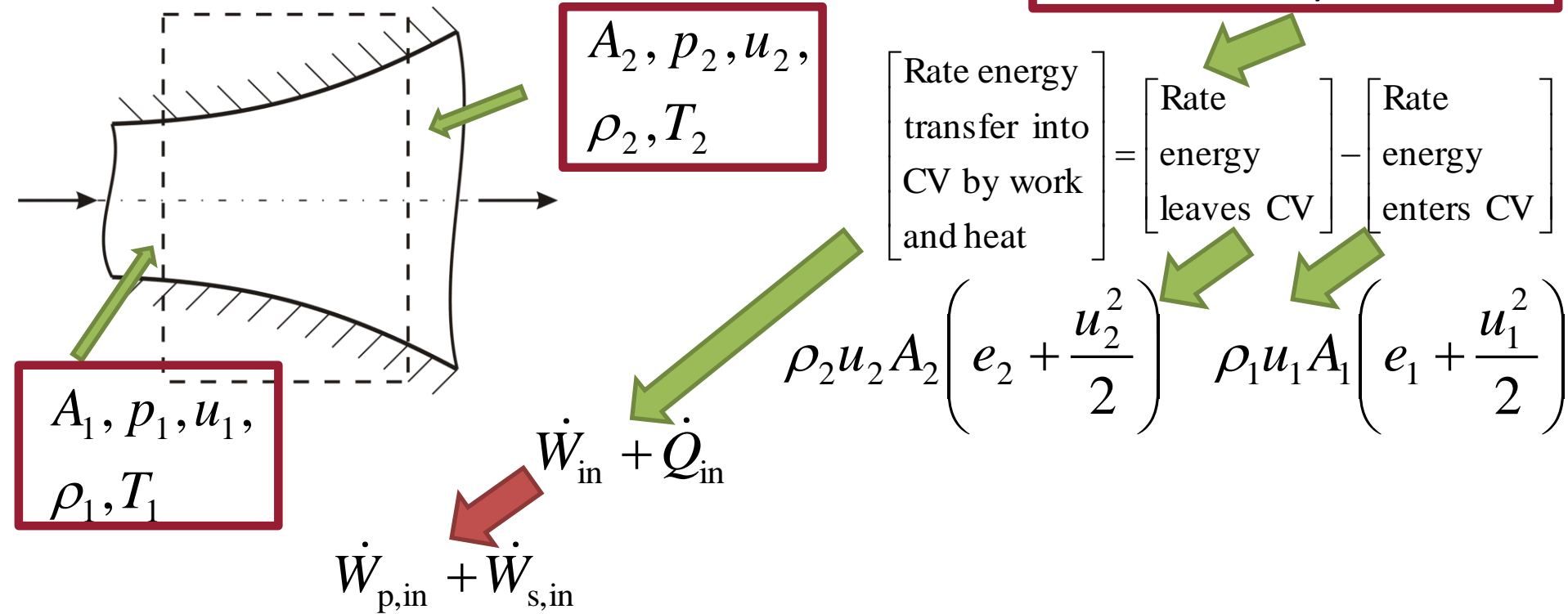


1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow

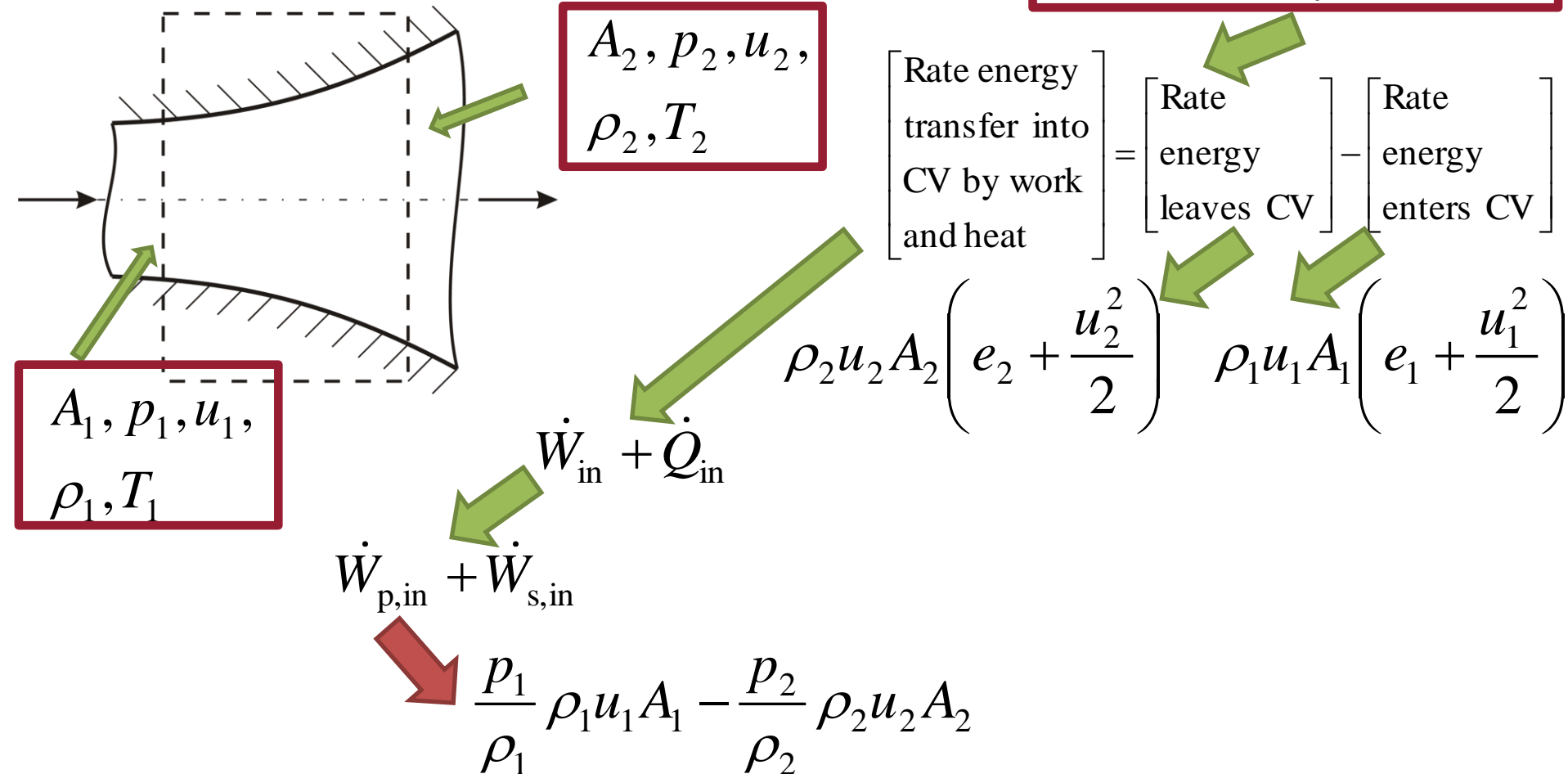


1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow

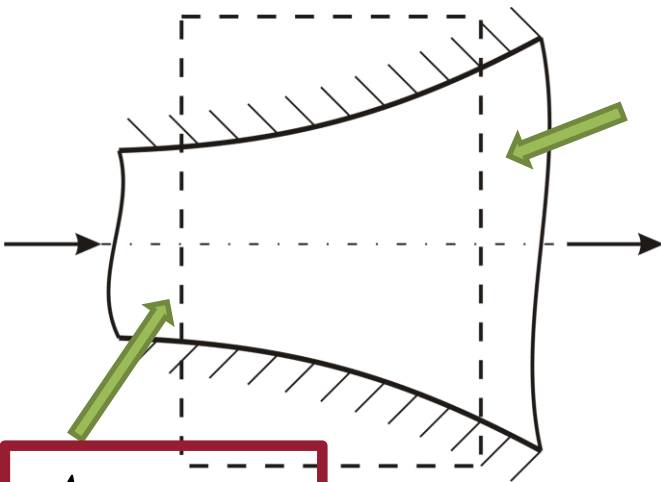


1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow



$$A_1, p_1, u_1, \rho_1, T_1$$

$$A_2, p_2, u_2, \rho_2, T_2$$

$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

$$\rho_2 u_2 A_2 \left(e_2 + \frac{u_2^2}{2} \right)$$

$$\rho_1 u_1 A_1 \left(e_1 + \frac{u_1^2}{2} \right)$$

$$\frac{p_1}{\rho_1} \rho_1 u_1 A_1 - \frac{p_2}{\rho_2} \rho_2 u_2 A_2 + \dot{W}_{s,\text{in}} + \dot{Q}_{\text{in}}$$

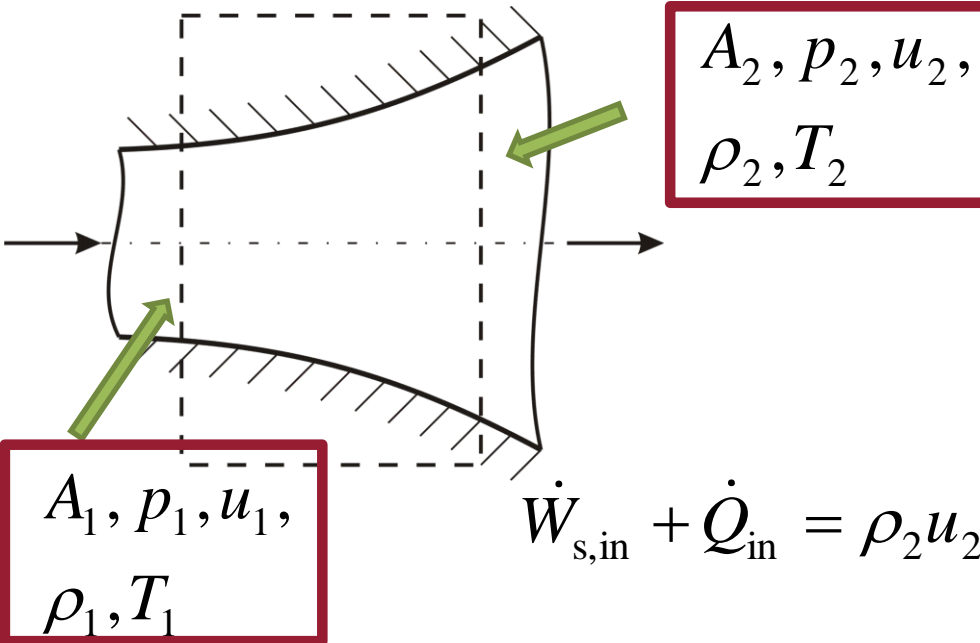
$$\dot{W}_{s,\text{in}} + \dot{Q}_{\text{in}} = \rho_2 u_2 A_2 \left(e_2 + \frac{p_2}{\rho_2} + \frac{u_2^2}{2} \right) - \rho_1 u_1 A_1 \left(e_1 + \frac{p_1}{\rho_1} + \frac{u_1^2}{2} \right)$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow



$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

using enthalpy h

$$\dot{W}_{s,\text{in}} + \dot{Q}_{\text{in}} = \rho_2 u_2 A_2 \left(h_2 + \frac{u_2^2}{2} \right) - \rho_1 u_1 A_1 \left(h_1 + \frac{u_1^2}{2} \right)$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow

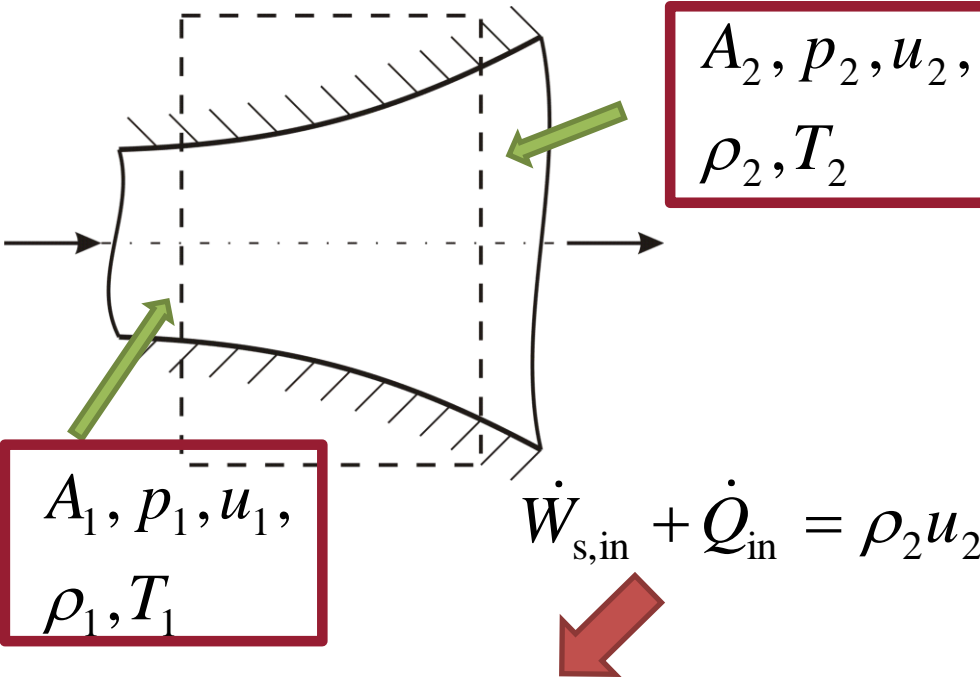
$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

using enthalpy h

$$\dot{W}_{s,\text{in}} + \dot{Q}_{\text{in}} = \rho_2 u_2 A_2 \left(h_2 + \frac{u_2^2}{2} \right) - \rho_1 u_1 A_1 \left(h_1 + \frac{u_1^2}{2} \right)$$

adiabatic process
no shaft work

$$\left(h_1 + \frac{u_1^2}{2} \right) = \left(h_2 + \frac{u_2^2}{2} \right)$$

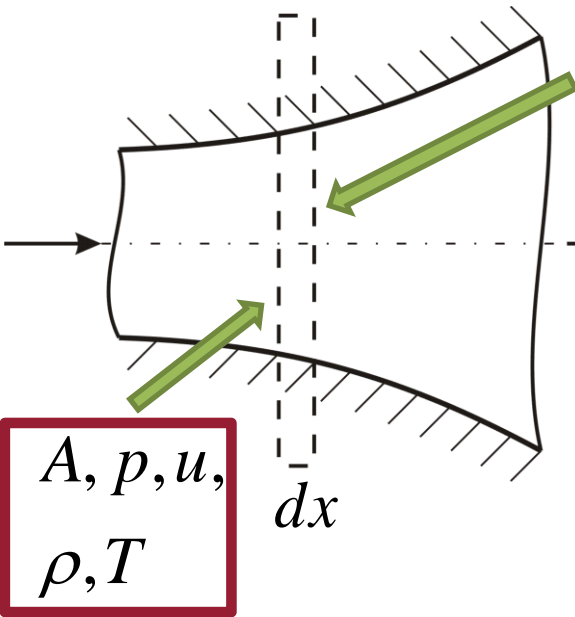


1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow



$$\begin{aligned} &A + dA, p + dp \\ &u + du, \rho + d\rho \\ &T + dT \end{aligned}$$

$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

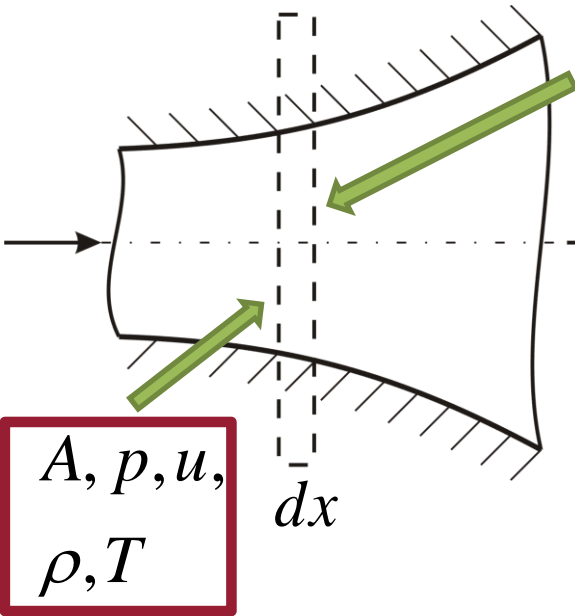
$$\begin{aligned} &+ (\rho + d\rho)(u + du)(A + dA) \cdot \rho u A \left(e + u^2 / 2 \right) \\ &\cdot \left(e + de + (u + du)^2 / 2 \right) \end{aligned}$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow



$$\begin{aligned} &A + dA, p + dp \\ &u + du, \rho + d\rho \\ &T + dT \end{aligned}$$

$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

$$\begin{aligned} &+ (\rho + d\rho)(u + du)(A + dA) \cdot \rho u A \left(e + u^2 / 2 \right) \\ &\cdot \left(e + de + (u + du)^2 / 2 \right) \end{aligned}$$

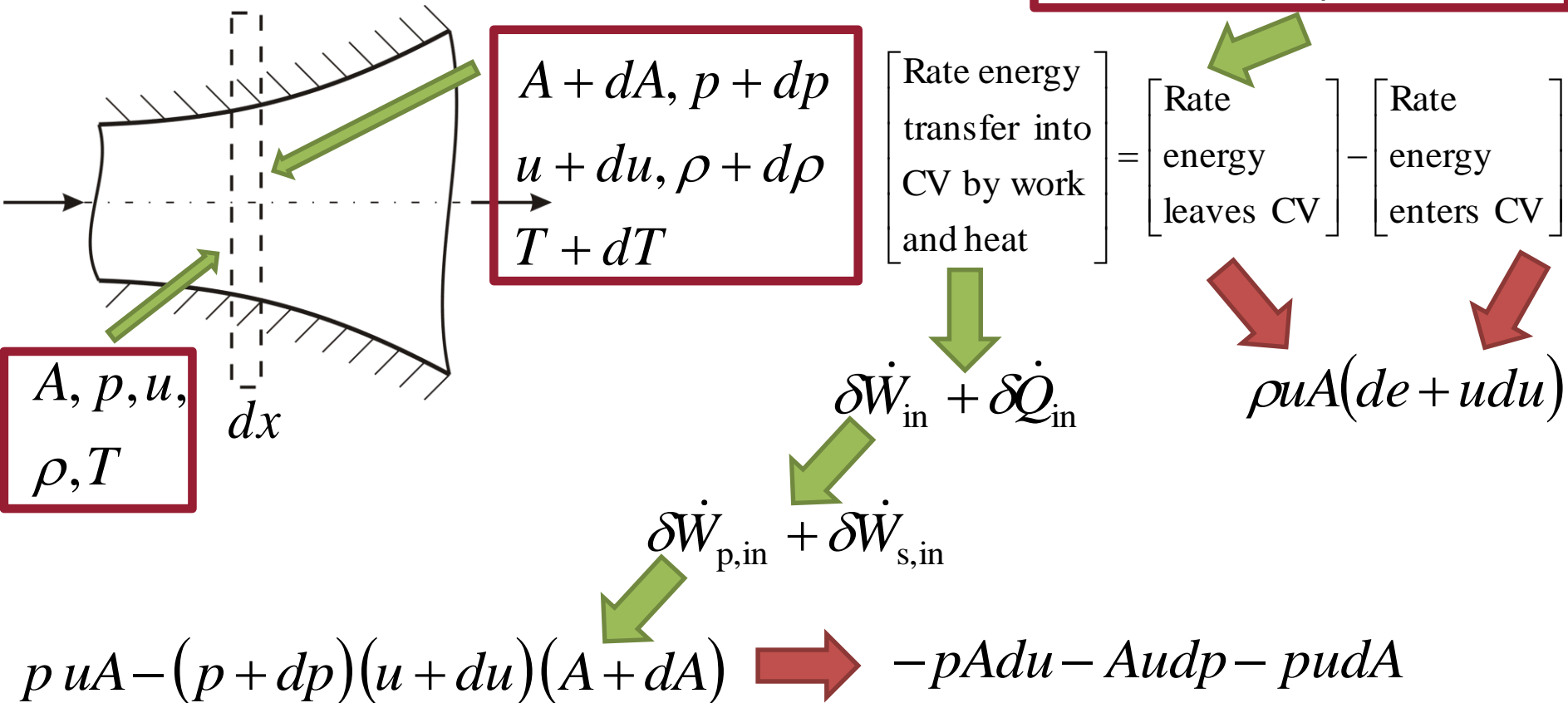
$$\rho u A (de + u du)$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow

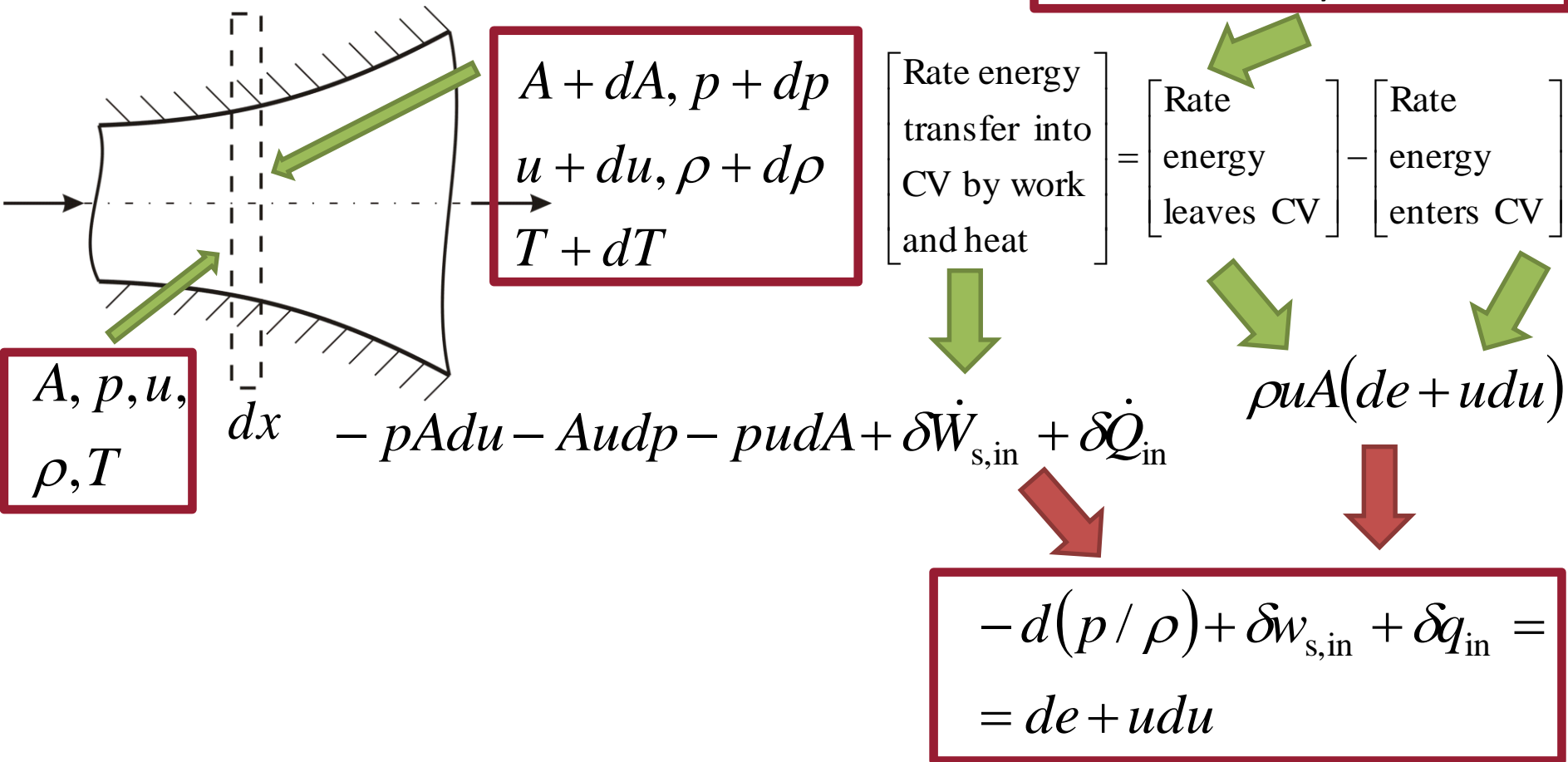


1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

inviscid steady-state flow

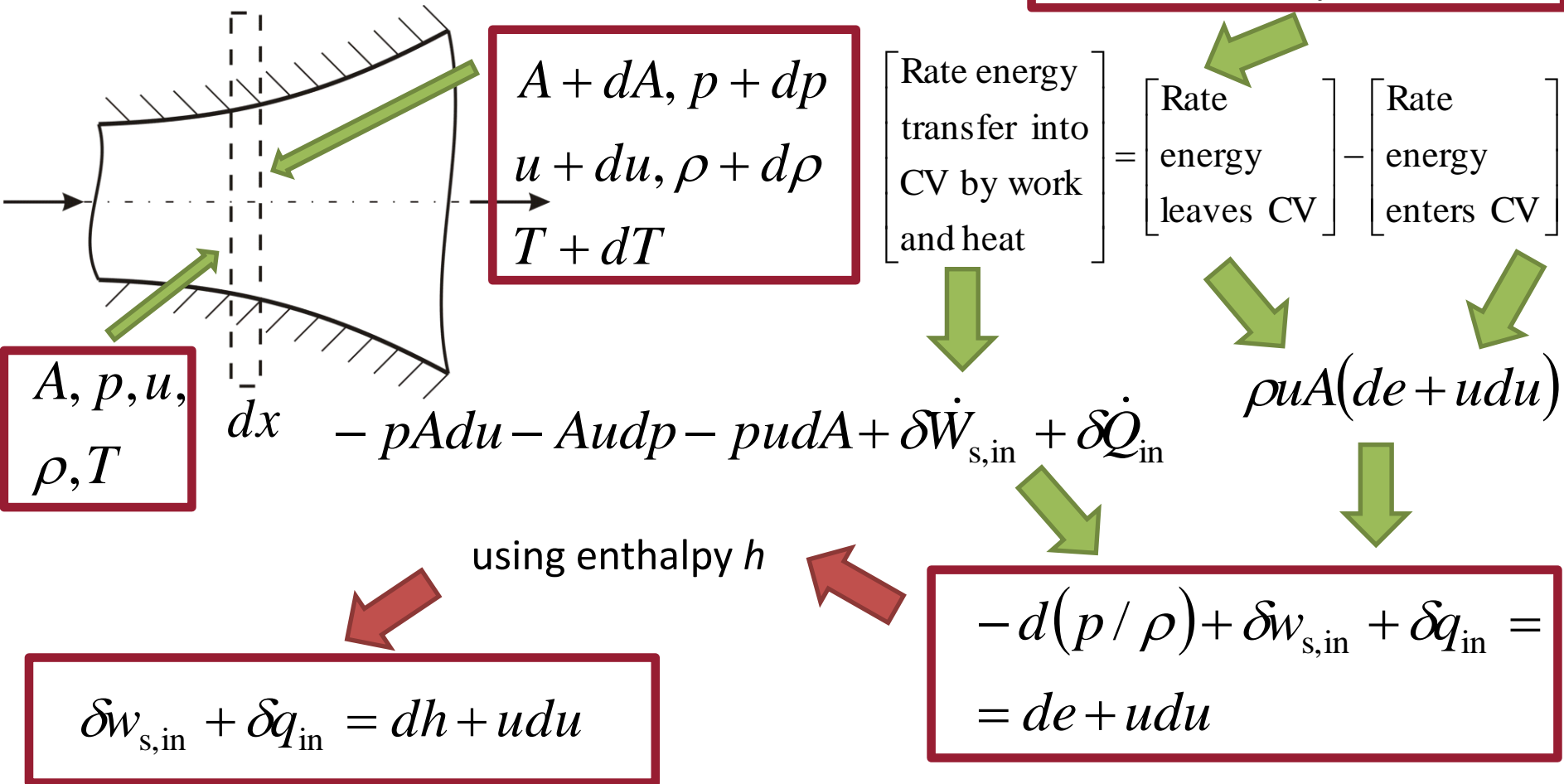


1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation

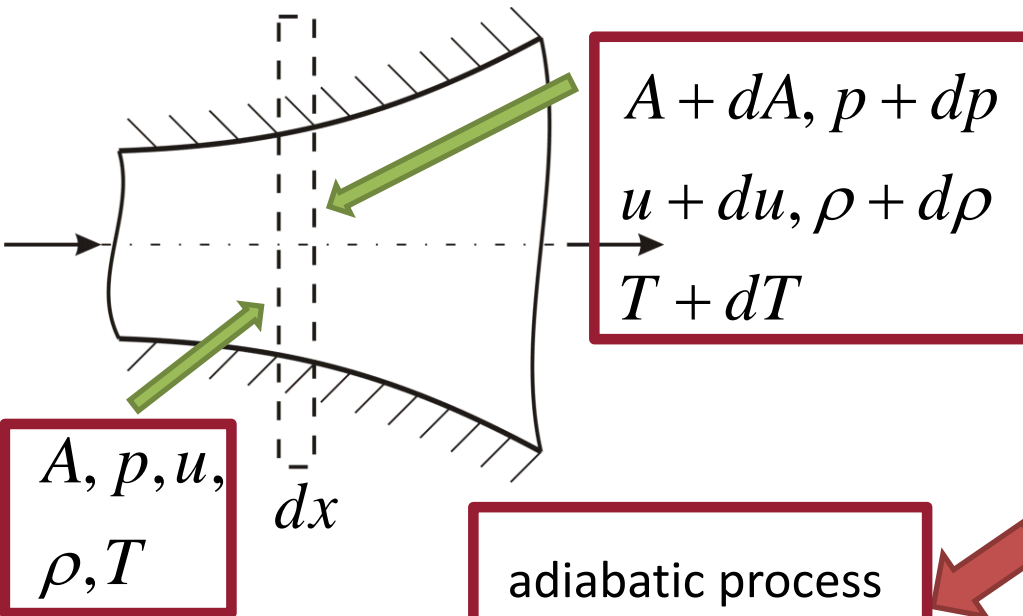
inviscid steady-state flow



1. Fluid Flow and Thermodynamics

Fundamental equations

- Energy equation



inviscid steady-state flow

$$\left[\begin{array}{l} \text{Rate energy} \\ \text{transfer into} \\ \text{CV by work} \\ \text{and heat} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{energy} \\ \text{enters CV} \end{array} \right]$$

$$\delta w_{s,\text{in}} + \delta q_{\text{in}} = dh + u du$$

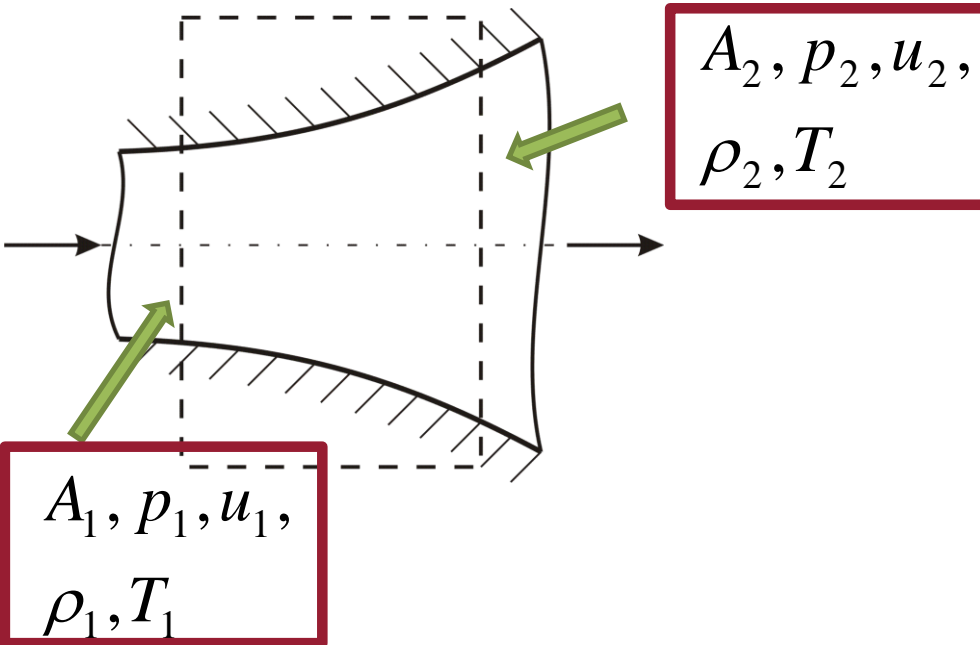
adiabatic process
no shaft work

$$dh + u du = 0$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation



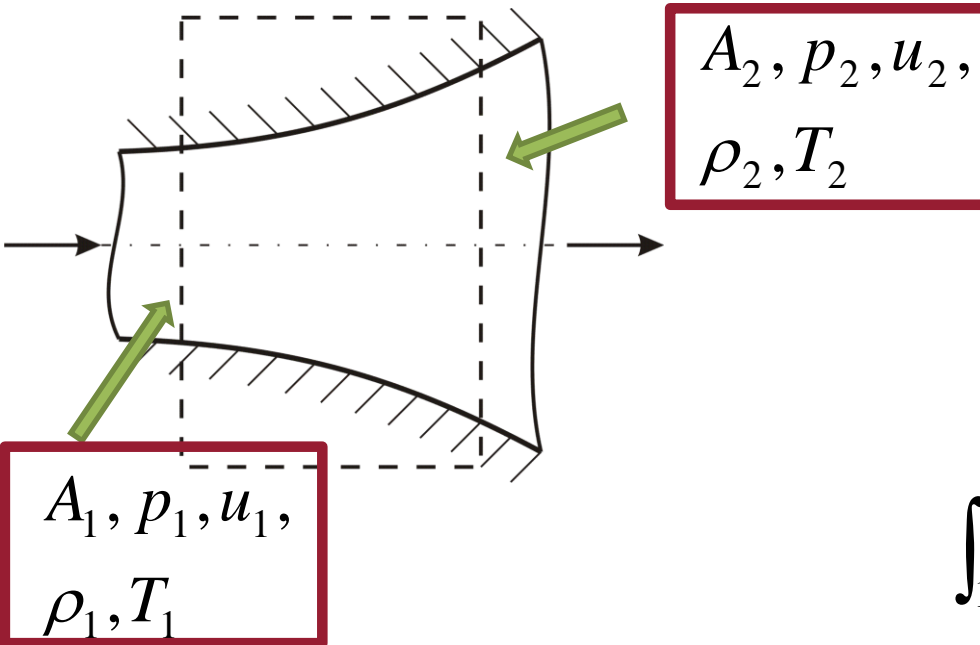
inviscid steady-state flow

$$\left[\begin{array}{l} \text{Rate entropy} \\ \text{transfer into} \\ \text{CV by heat and} \\ \text{irreversibilities} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{leaves} \\ \text{CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{enters} \\ \text{CV} \end{array} \right]$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation



inviscid steady-state flow

$$\left[\begin{array}{l} \text{Rate entropy} \\ \text{transfer into} \\ \text{CV by heat and} \\ \text{irreversibilities} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{leaves} \\ \text{CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{enters} \\ \text{CV} \end{array} \right]$$

$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} + \dot{S}_{\text{gen}}$$

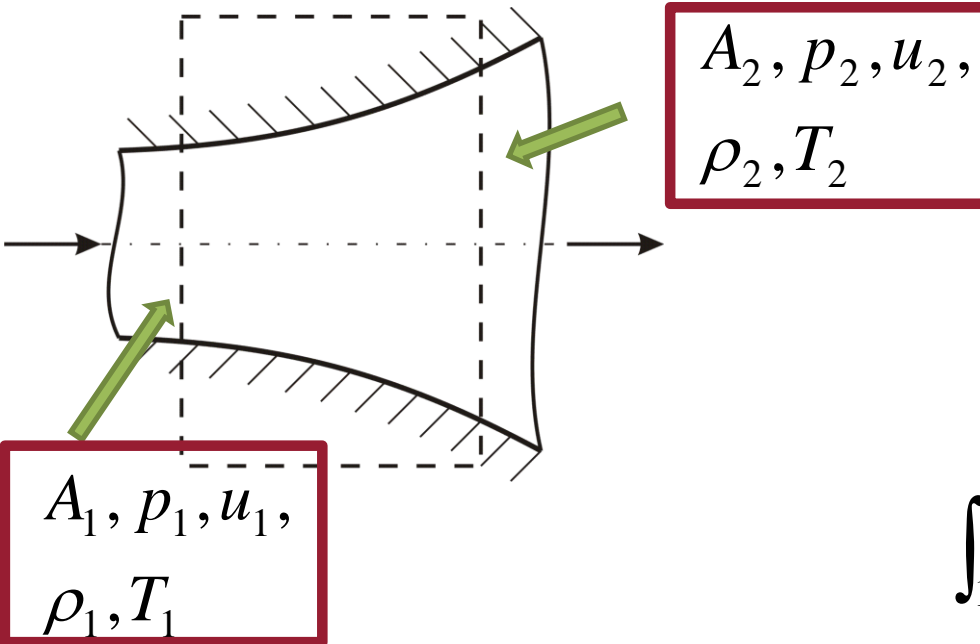
$$\dot{S}_2$$

$$\dot{S}_1$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation



inviscid steady-state flow

$$\left[\begin{array}{l} \text{Rate entropy} \\ \text{transfer into} \\ \text{CV by heat and} \\ \text{irreversibilities} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{leaves} \\ \text{CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{enters} \\ \text{CV} \end{array} \right]$$

$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} + \dot{S}_{\text{gen}}$$

$$\dot{S}_2$$

$$\dot{S}_1$$

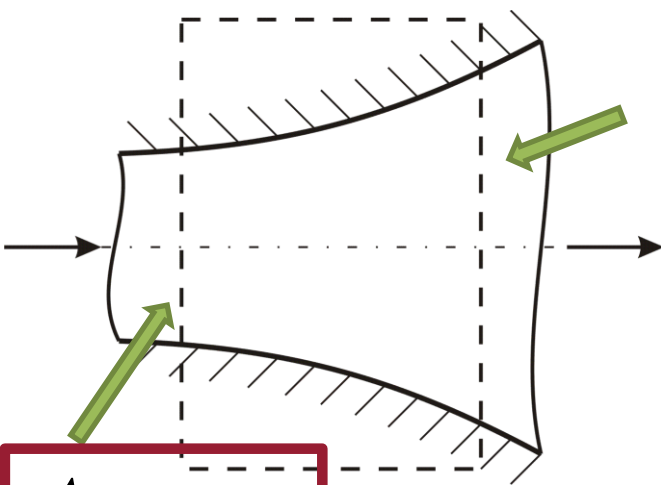
$$\dot{S}_2 - \dot{S}_1$$

$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} + \dot{S}_{\text{gen}} = \dot{S}_2 - \dot{S}_1$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation



$$A_1, p_1, u_1, \rho_1, T_1$$

$$A_2, p_2, u_2, \rho_2, T_2$$

inviscid steady-state flow

$$\left[\begin{array}{l} \text{Rate entropy} \\ \text{transfer into} \\ \text{CV by heat and} \\ \text{irreversibilities} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{leaves} \\ \text{CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{enters} \\ \text{CV} \end{array} \right]$$

$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} + \dot{S}_{\text{gen}}$$

$$\dot{S}_2$$

$$\dot{S}_1$$

reversible
process 1-2

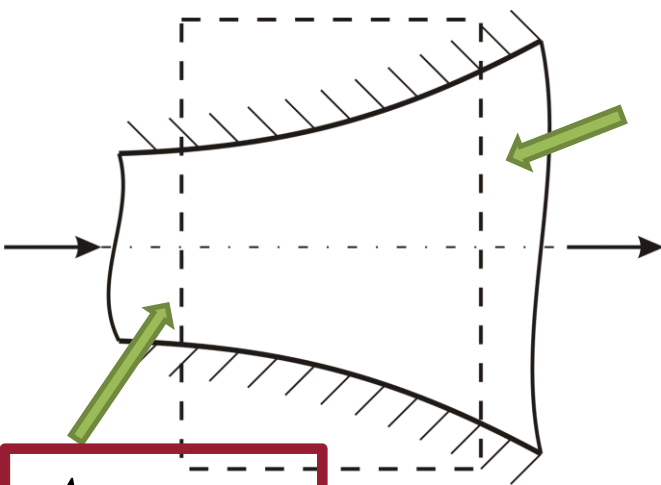
$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} = \dot{S}_2 - \dot{S}_1$$

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1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation



$$A_1, p_1, u_1, \rho_1, T_1$$

$$A_2, p_2, u_2, \rho_2, T_2$$

isentropic
process 1-2

$$0 = \dot{S}_2 - \dot{S}_1$$

adiabatic
reversible
process 1-2

inviscid steady-state flow

$$\left[\begin{array}{l} \text{Rate entropy} \\ \text{transfer into} \\ \text{CV by heat and} \\ \text{irreversibilities} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{leaves} \\ \text{CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{enters} \\ \text{CV} \end{array} \right]$$

$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} + \dot{S}_{\text{gen}}$$

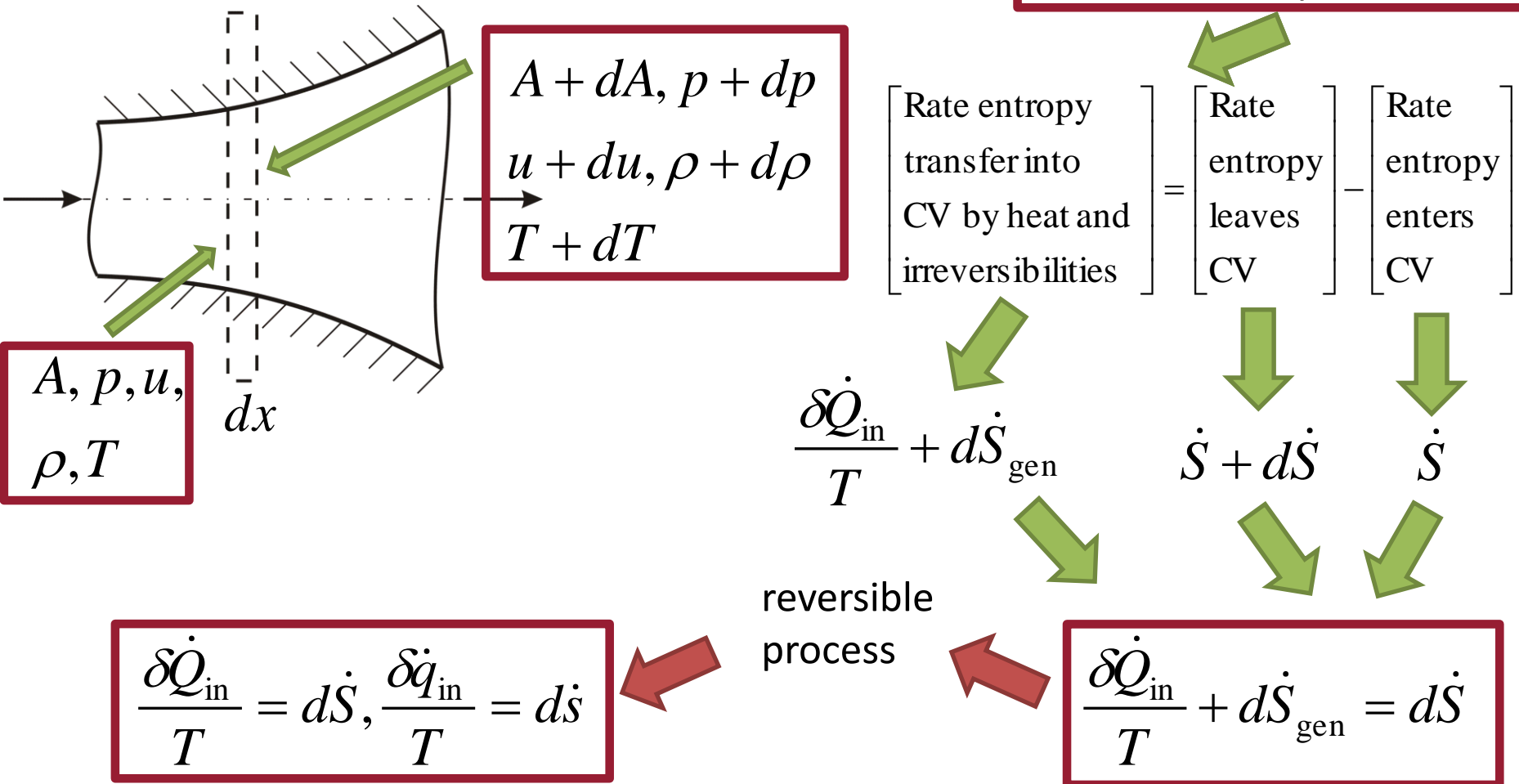
$$\dot{S}_2 \quad \dot{S}_1$$

$$\int_1^2 \frac{\delta \dot{Q}_{\text{in}}}{T} + \dot{S}_{\text{gen}} = \dot{S}_2 - \dot{S}_1$$

1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation

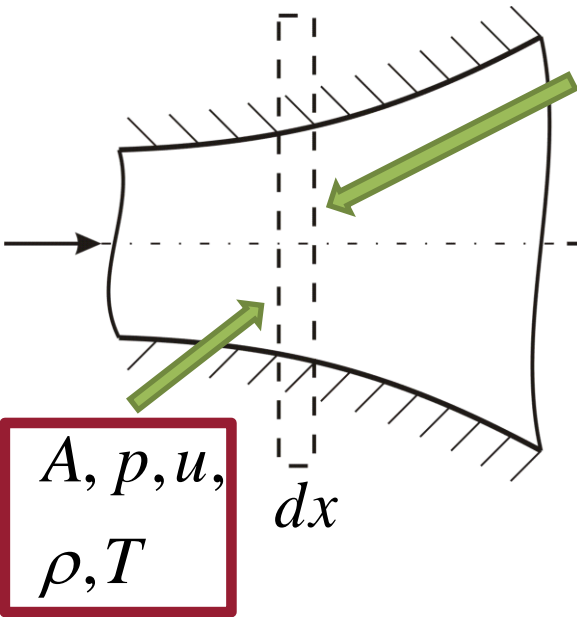


1. Fluid Flow and Thermodynamics

Fundamental equations

- Entropy equation

inviscid steady-state flow



$$\begin{aligned} A + dA, p + dp \\ u + du, \rho + d\rho \\ T + dT \end{aligned}$$

$$\left[\begin{array}{l} \text{Rate entropy} \\ \text{transfer into} \\ \text{CV by heat and} \\ \text{irreversibilities} \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{leaves} \\ \text{CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{entropy} \\ \text{enters} \\ \text{CV} \end{array} \right]$$

$$\frac{\delta \dot{Q}_{\text{in}}}{T} + d\dot{S}_{\text{gen}}$$

$$\dot{S} + d\dot{S}$$

$$\dot{S}$$

isentropic
process

$$0 = d\dot{S}, 0 = d\dot{s}, 0 = ds$$

adiabatic
reversible
process

$$\frac{\delta \dot{Q}_{\text{in}}}{T} + d\dot{S}_{\text{gen}} = d\dot{S}$$

1. Fluid Flow and Thermodynamics

Fundamental equations

Quasi-one dimension fluid flow equations:

- Continuity equation →
- Momentum equation →
- Energy equation →
- Entropy equation →

$$\frac{d\rho}{\rho} + \frac{du}{u} + \frac{dA}{A} = 0$$

$$-\frac{dp}{\rho} = u du$$

$$dh + u du = 0$$

$$0 = ds$$

inviscid steady-state flow

adiabatic flow,
no shaft work

isentropic
flow

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- perfect gas

– equation of state (EOS) $p = \rho RT$ or $p = \rho R_M T / M_m$
 $pv = RT$ or $pv = R_M T / M_m$

1. Fluid Flow and Thermodynamics

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 $pv = RT$ $pv = R_M T / M_m$

- internal energy $e \rightarrow e = e(T) \rightarrow de = c_v dT$

- enthalpy $h = e + pv \rightarrow h = h(T) \rightarrow dh = c_p dT$

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

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 $p v = R T$ or $p v = R_M T / M_m$

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$$e = c_v T$$

$$h = c_p T$$

calorically
perfect gas

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calorically
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- heat capacity ratio $\gamma = c_p / c_v$

- difference between
heat capacities $R = c_p - c_v$

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Thermodynamics of gases

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calorically
perfect gas

- heat capacity ratio $\gamma = c_p / c_v$

- difference between
heat capacities

$$R = c_p - c_v$$

$$c_p = \frac{\gamma R}{\gamma - 1} \quad c_v = \frac{R}{\gamma - 1}$$

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- moving fluid with speed u
 - static parameters $\Rightarrow p, T, \rho \Rightarrow$ “measured” in moving fluid

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- moving fluid with speed u

- static parameters

 p, T, ρ 

“measured” in
moving fluid

- fluid brought to rest
adiabatically

 $u_0 = 0$  p_0, T_0, ρ_0

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- moving fluid with speed u

- static parameters

$\Rightarrow p, T, \rho \Rightarrow$ “measured” in moving fluid

- fluid brought to rest adiabatically

$\downarrow u_0 = 0$
 $\Rightarrow p_0, T_0, \rho_0$

total (stagnation) pressure, temperature, density

- corresponding total enthalpy to total temperature

$\Rightarrow h_0 = c_p T_0$

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- stagnation conditions
 - adiabatic process

total enthalpy is constant through steady, inviscid, adiabatic flow

$$\left(h_1 + \frac{u_1^2}{2} \right) = \text{const.}$$

rest of fluid



$$h_0 = \left(h_1 + \frac{u_1^2}{2} \right)$$



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also **total temperature** is constant

$$h_0 = c_p T_0$$



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also **total temperature** is constant

$$h_0 = c_p T_0$$



- isentropic process



$$p_0, \rho_0$$



total pressure and **total density** are constant

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Thermodynamics of gases

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1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- moving fluid with speed u

- static parameters

 p, T, ρ 

“measured” in
moving fluid

- fluid brought to speed
of sound isentropically

 p^*, T^*, ρ^*  $u = a^*$

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- moving fluid with speed u

- static parameters

→ p, T, ρ → “measured” in moving fluid

- fluid brought to speed of sound isentropically

↓ $u = a^*$
→ p^*, T^*, ρ^*

↓
**critical pressure,
temperature, density**

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

- reversible process
 - added heat – closed system $\Rightarrow \delta w_{\text{in}} + \delta q_{\text{in}} = de$
 - reversible process $\Rightarrow \delta q_{\text{in}} = dsT$
 - internal energy $\Rightarrow de = c_v dT$
 - pressure work $\Rightarrow \delta w_{\text{in}} = -p dv$

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$$-pdv + dsT = c_v dT$$

using

c_v

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$$-pdv + dsT = c_v dT$$

$$-pdv + d(pv) + dsT = de + d(pv)$$

$$dpv + dsT = c_p dT$$

using
 c_v

using
 c_p

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

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using

c_v

using

c_p

$$ds = c_v \frac{dT}{T} + \frac{pdv}{T}$$

$$ds = c_p \frac{dT}{T} - \frac{vdp}{T}$$

$$-pdv + dsT = c_v dT$$

$$-pdv + d(pv) + dsT = de + d(pv)$$

$$dpv + dsT = c_p dT$$

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

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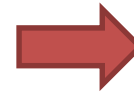
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$$ds = c_v \frac{dT}{T} + \frac{pdv}{T}$$

$$ds = c_p \frac{dT}{T} - \frac{vdp}{T}$$



isentropic
process

$$0 = c_v \frac{dT}{T} + \frac{pdv}{T}$$

$$0 = c_p \frac{dT}{T} - \frac{vdp}{T}$$

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

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$$\frac{v_2}{v_1} = \left(\frac{T_2}{T_1} \right)^{\frac{-1}{\gamma-1}}$$

$$\frac{p_2}{p_1} = \left(\frac{T_2}{T_1} \right)^{\frac{\gamma}{\gamma-1}}$$

$$\gamma c_v = c_p = \frac{\gamma R}{\gamma - 1}$$

$$0 = c_v \frac{dT}{T} + \frac{Rdv}{v}$$

$$0 = c_p \frac{dT}{T} - \frac{Rdp}{p}$$

$$0 = c_v \frac{dT}{T} + \frac{pdv}{T}$$

$$0 = c_p \frac{dT}{T} - \frac{vdp}{T}$$

using EOS

1. Fluid Flow and Thermodynamics

Thermodynamics of gases

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$$\frac{p_2}{p_1} = \left(\frac{T_2}{T_1} \right)^{\frac{\gamma}{\gamma-1}} = \left(\frac{v_1}{v_2} \right)^{\gamma} = \left(\frac{\rho_2}{\rho_1} \right)^{\gamma}$$

isentropic process

1. Fluid Flow and Thermodynamics

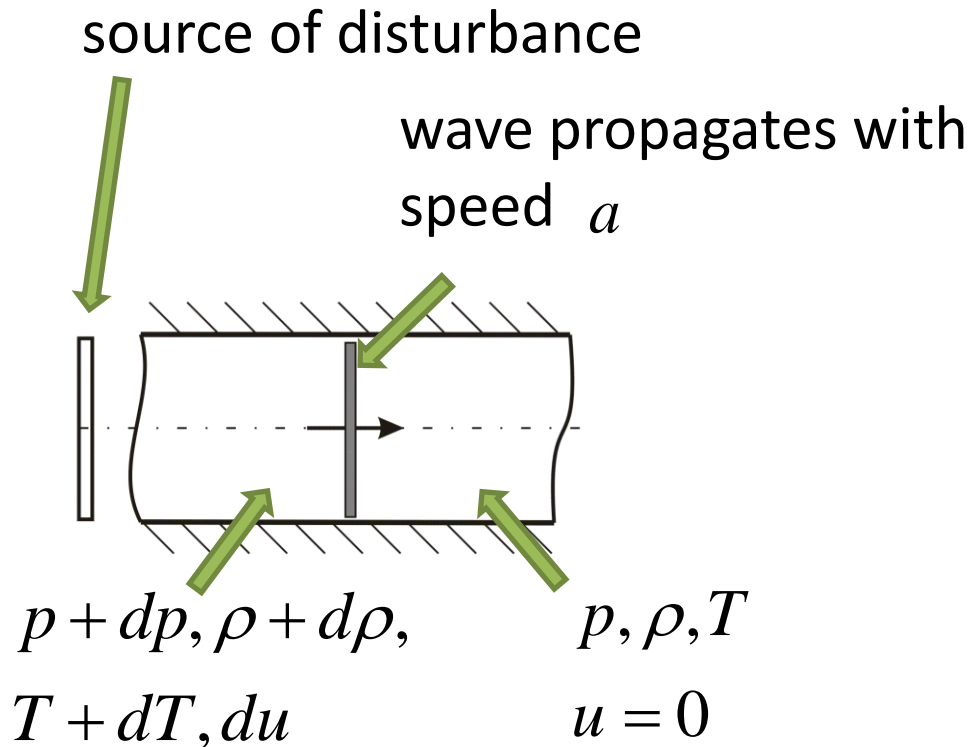
Speed of sound

- physical mechanism:
 - sound propagation in gas is based on molecular motion
 - energy is transfer to gas molecules, they start to move about in random fashion
 - they collide with other molecules and transfer their energy to these molecules
 - the process of collision repeats – energy is propagated
 - macroscopic parameters p, T, ρ are slightly varied by increased microscopic parameter – energy of molecule

1. Fluid Flow and Thermodynamics

Speed of sound

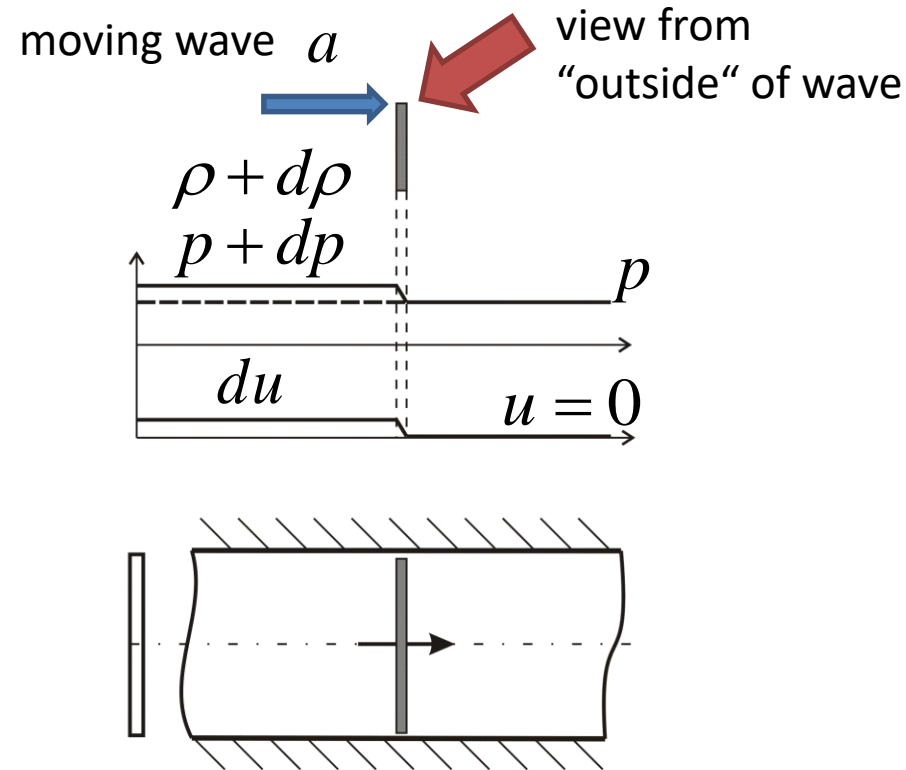
- parameters of fluid



1. Fluid Flow and Thermodynamics

Speed of sound

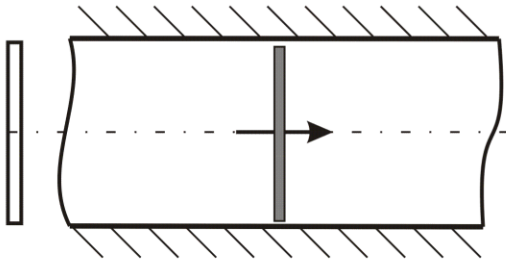
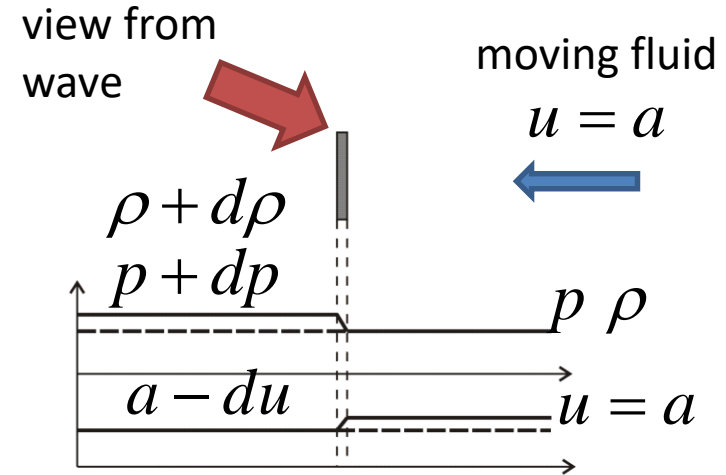
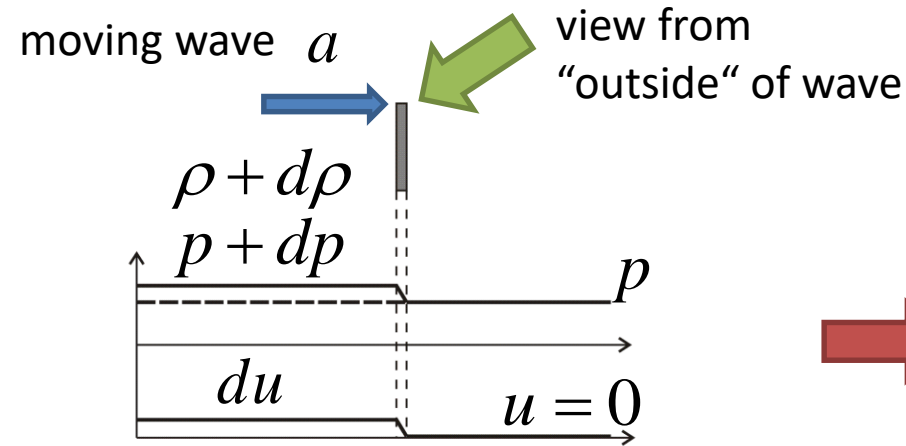
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1. Fluid Flow and Thermodynamics

Speed of sound

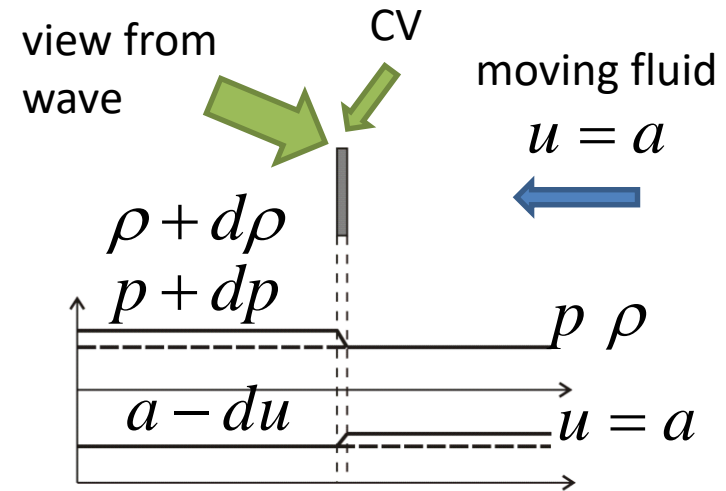
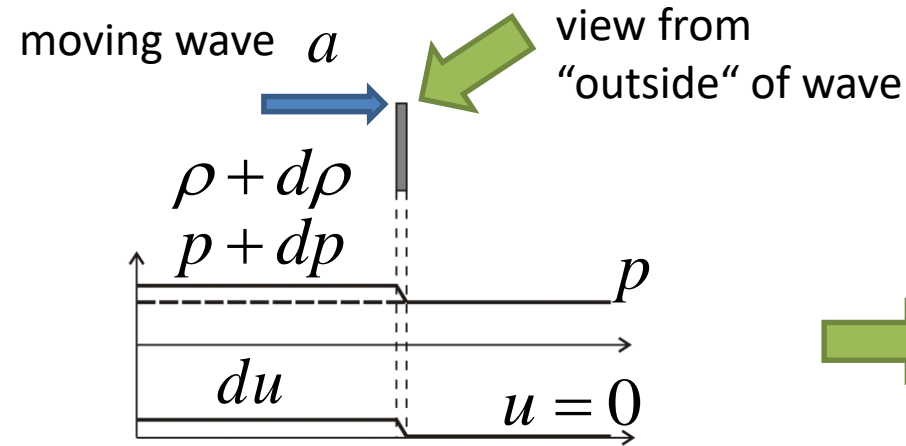
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1. Fluid Flow and Thermodynamics

Speed of sound

- parameters of fluid



momentum equation

$$(\rho + d\rho)(a - du)A - \rho a^2 A = A(p - (p + d p))$$

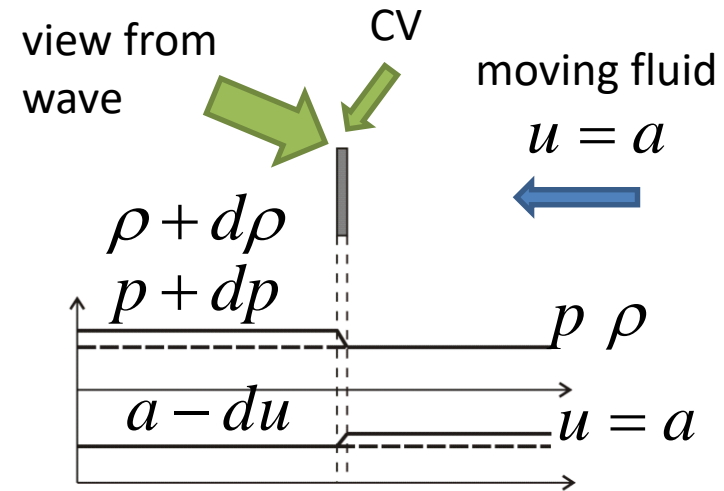
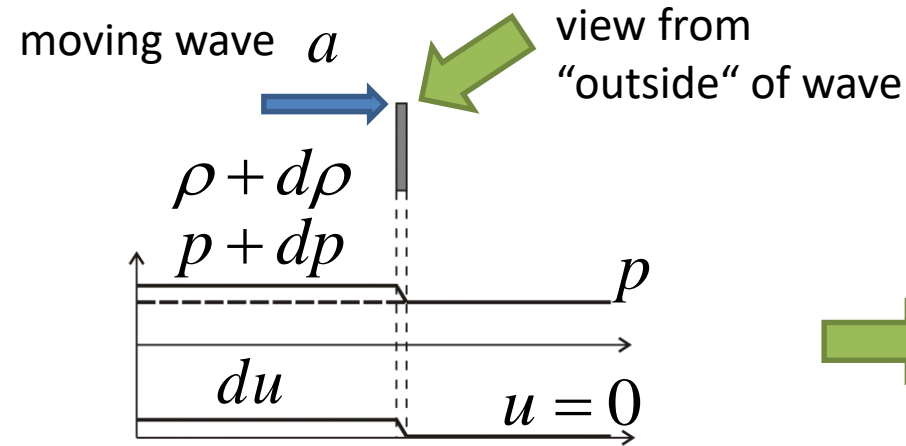
continuity equation

$$(\rho + d\rho)(a - du)A - \rho a A = 0$$

1. Fluid Flow and Thermodynamics

Speed of sound

- parameters of fluid



momentum equation

$$(\rho + d\rho)(a - du)A - \rho a^2 A = A(p - (p + dp))$$

$$a = \frac{dp}{\rho du}$$

continuity equation

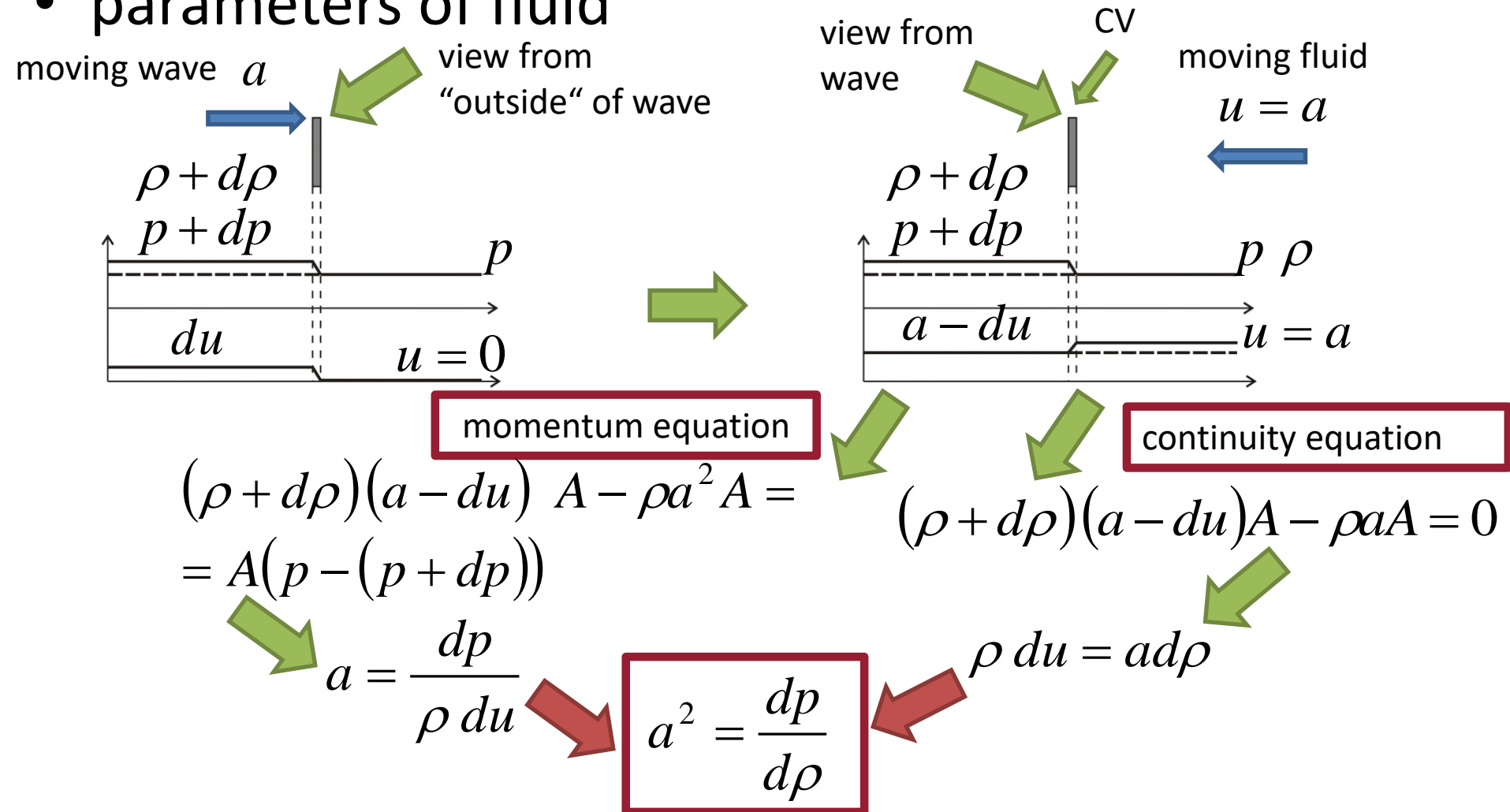
$$(\rho + d\rho)(a - du)A - \rho a A = 0$$

$$\rho du = a d\rho$$

1. Fluid Flow and Thermodynamics

Speed of sound

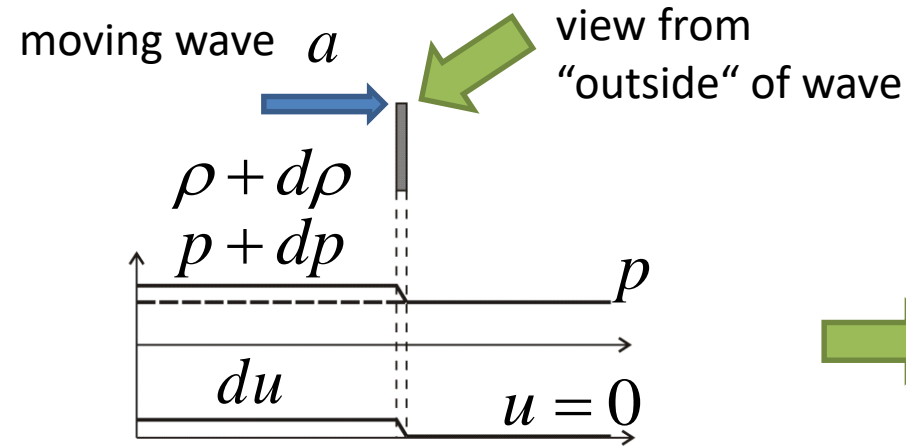
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1. Fluid Flow and Thermodynamics

Speed of sound

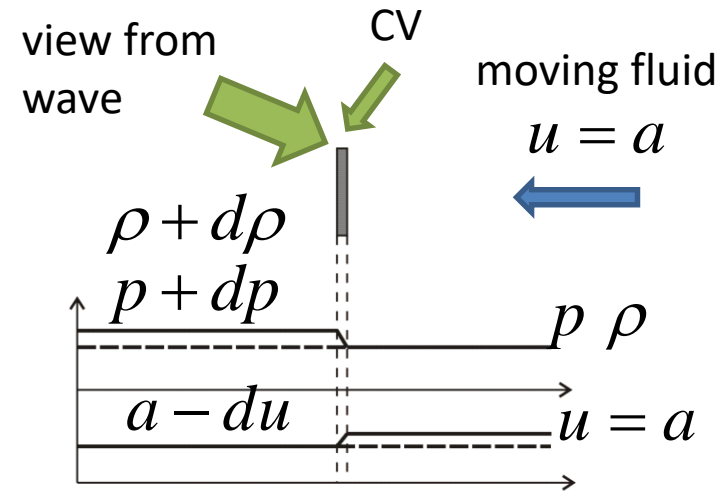
- parameters of fluid



$$\frac{p}{p_1} = \left(\frac{\rho}{\rho_1} \right)^\gamma$$



$$p = \frac{p_1}{\rho_1^\gamma} \rho^\gamma$$

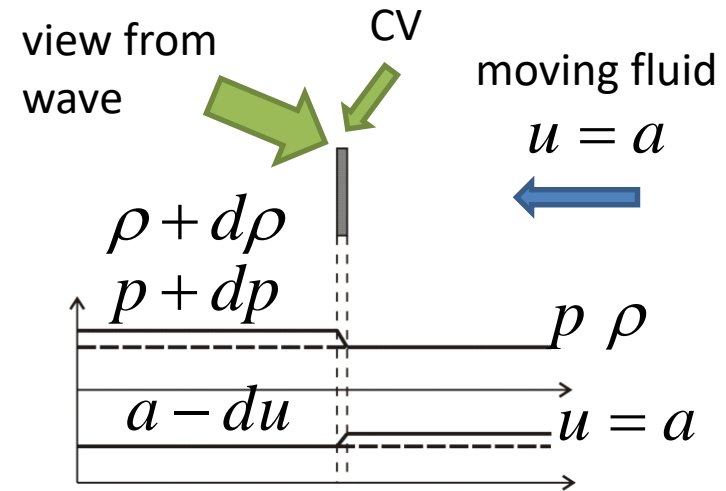
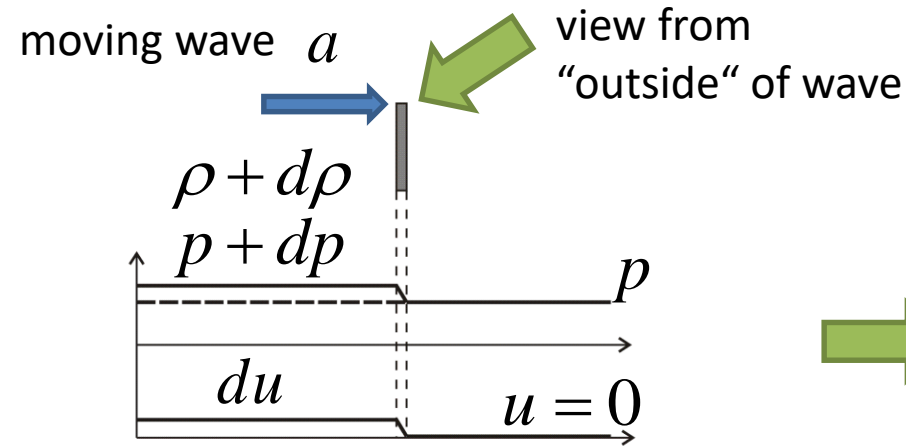


$$a^2 = \frac{dp}{d\rho}$$

1. Fluid Flow and Thermodynamics

Speed of sound

- parameters of fluid



$$\frac{p}{p_1} = \left(\frac{\rho}{\rho_1} \right)^\gamma \quad \frac{dp}{d\rho} = \frac{p_1}{\rho_1^\gamma} \gamma \rho^{\gamma-1} = \frac{p}{\rho} \gamma = \gamma RT$$

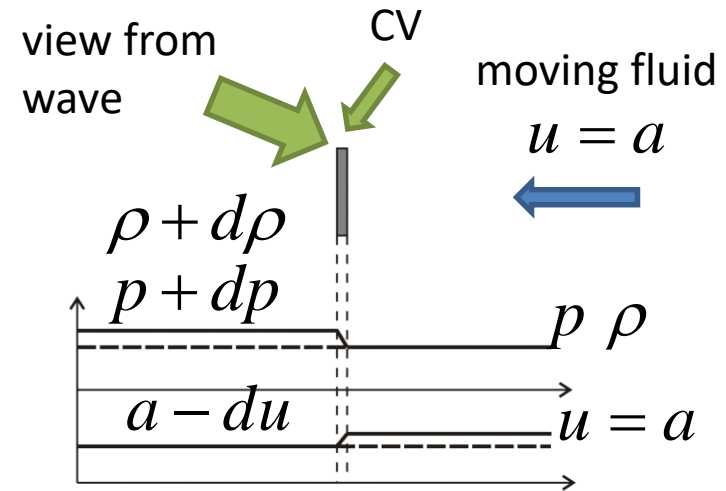
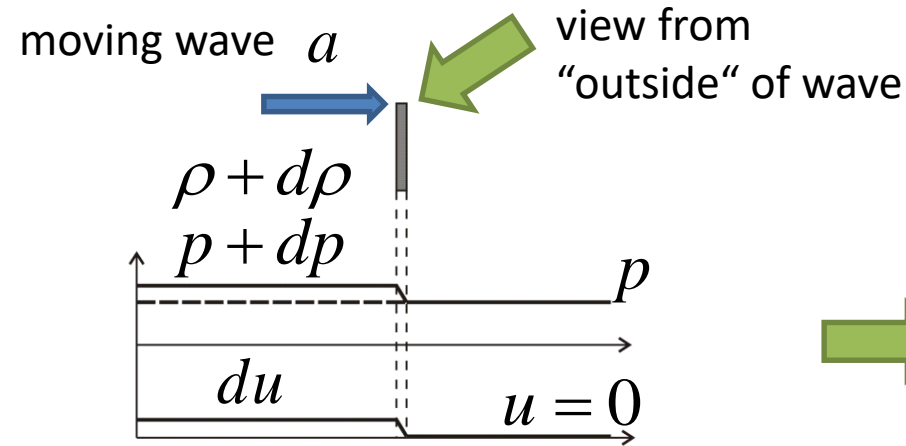
$$a^2 = \frac{dp}{d\rho}$$

$$p = \frac{p_1}{\rho_1^\gamma} \rho^\gamma$$

1. Fluid Flow and Thermodynamics

Speed of sound

- parameters of fluid



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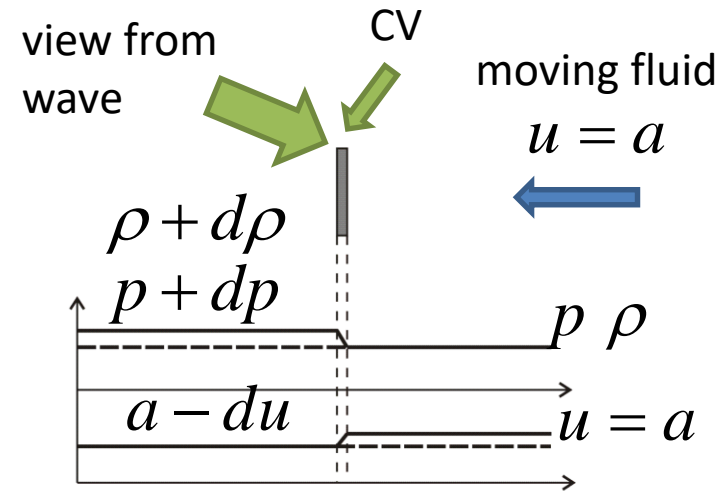
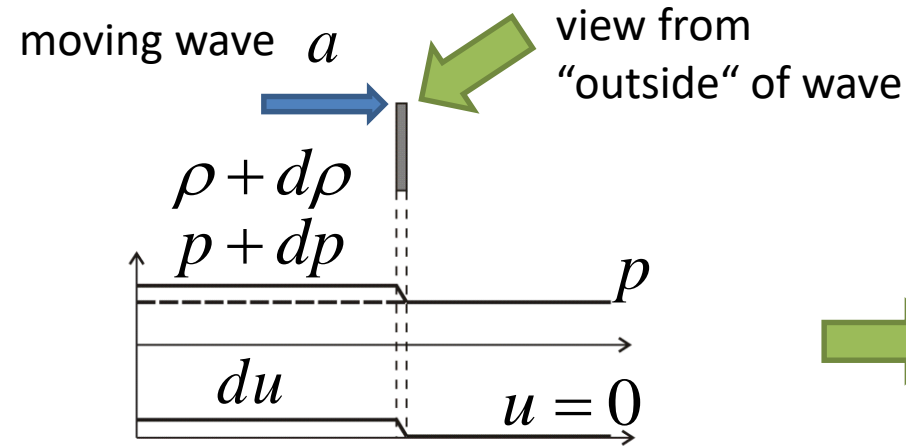
$$a^2 = \frac{dp}{d\rho}$$

$$a = \sqrt{\gamma RT}$$

1. Fluid Flow and Thermodynamics

Speed of sound

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$$\frac{p}{p_1} = \left(\frac{\rho}{\rho_1} \right)^\gamma \quad \frac{dp}{d\rho} = \frac{p_1}{\rho_1^\gamma} \gamma \rho^{\gamma-1} = \frac{p}{\rho} \gamma = \gamma RT$$

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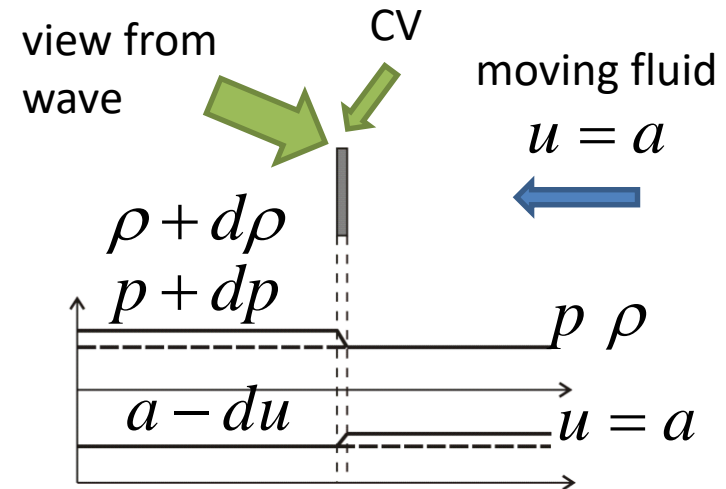
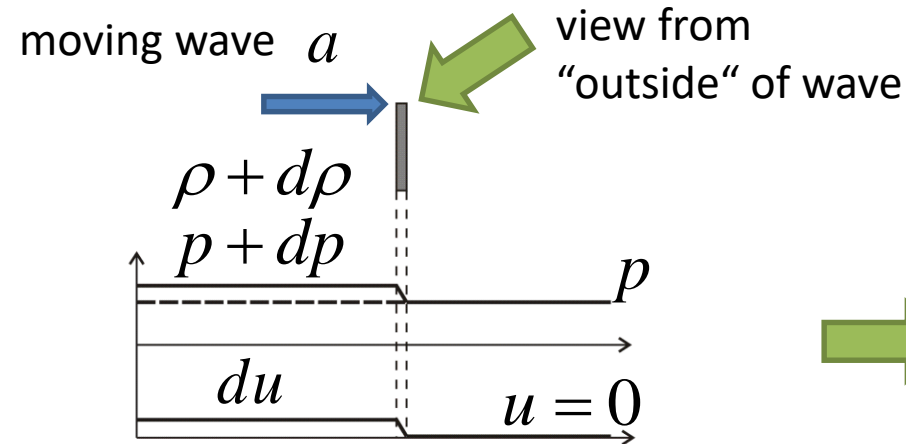
$$a = \sqrt{\gamma RT}$$

$$M = \frac{u}{a} = \frac{u}{\sqrt{\gamma RT}}$$

1. Fluid Flow and Thermodynamics

Speed of sound

- parameters of fluid



Gas	M_m [g/mol]	γ [-]	a at 0°C [m/s]
Air	28.96	1.404	331
Hydrogen	2.016	1.407	1270
Xenon	131.3	1.667	170

$$a = \sqrt{\gamma \frac{R_M}{M_m} T}$$

$$a = \sqrt{\gamma RT}$$


$$a^2 = \frac{dp}{d\rho}$$

$$M = \frac{u}{a} = \frac{u}{\sqrt{\gamma RT}}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- Importance of isentropic flow:
 - isentropic flow is adiabatic in which viscous losses are negligible
 - real flows are not isentropic



the effects of viscosity and heat transfer are restricted to thin layers near the walls

major part of the flow can be assumed to be isentropic

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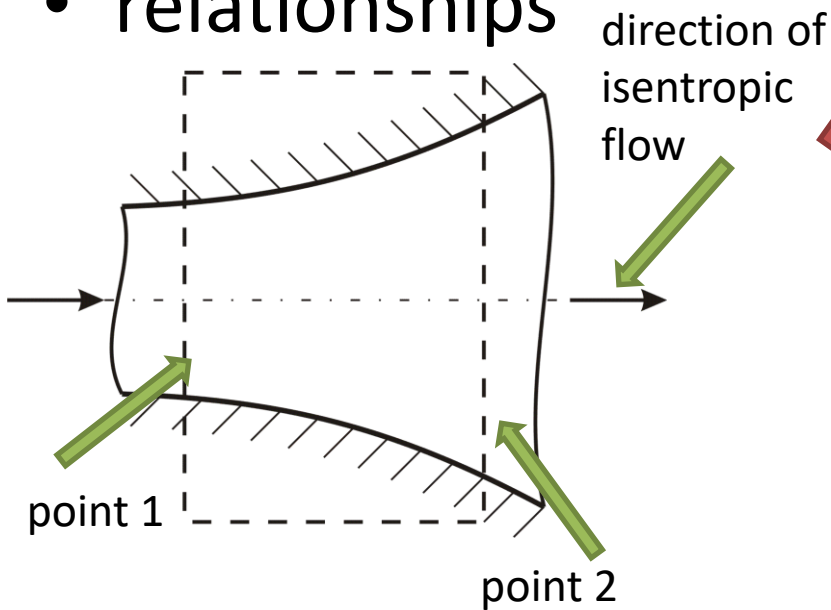
major part of the flow can be assumed to be isentropic

many flows in engineering practice can be adequately modeled by assuming them to be **isentropic** and also **steady-state** and **quasi-one dimensional flow**

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



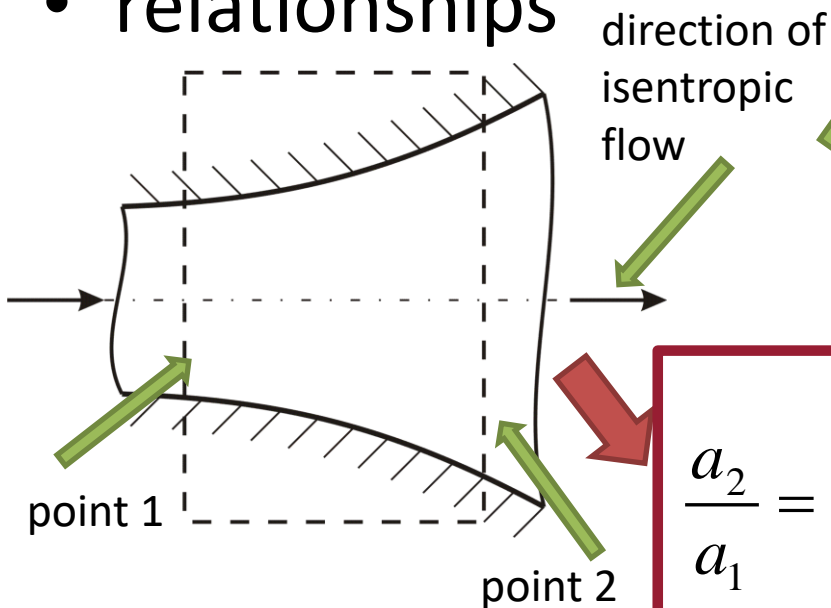
isentropic
process 1-2

$$\frac{p_2}{p_1} = \left(\frac{T_2}{T_1} \right)^{\frac{\gamma}{\gamma-1}} = \left(\frac{\rho_2}{\rho_1} \right)^{\gamma}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



isentropic
process 1-2

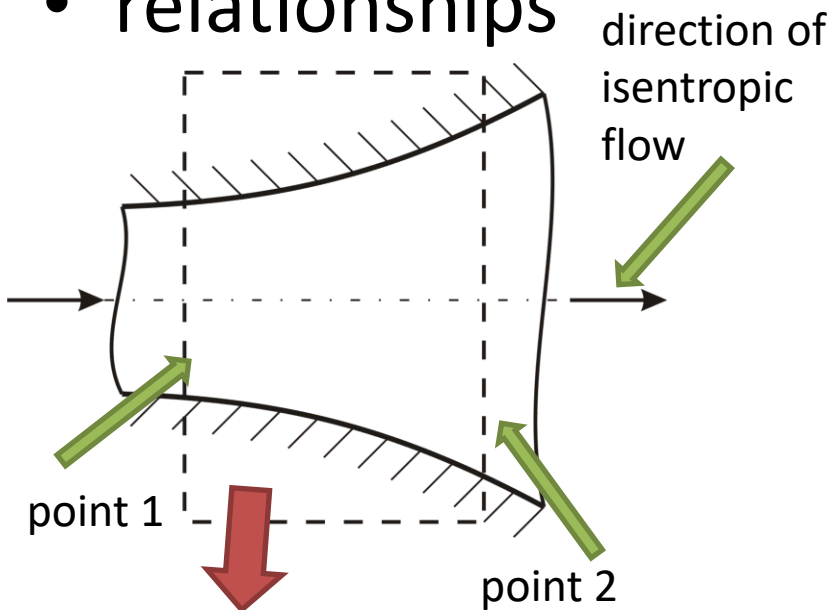
$$\frac{p_2}{p_1} = \left(\frac{T_2}{T_1} \right)^{\frac{\gamma}{\gamma-1}} = \left(\frac{\rho_2}{\rho_1} \right)^{\gamma}$$

$$\frac{a_2}{a_1} = \frac{\sqrt{\gamma RT_2}}{\sqrt{\gamma RT_1}} = \left(\frac{T_2}{T_1} \right)^{\frac{1}{2}} = \left(\frac{p_2}{p_1} \right)^{\frac{\gamma-1}{2\gamma}} = \left(\frac{\rho_2}{\rho_1} \right)^{\frac{\gamma-1}{2}}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



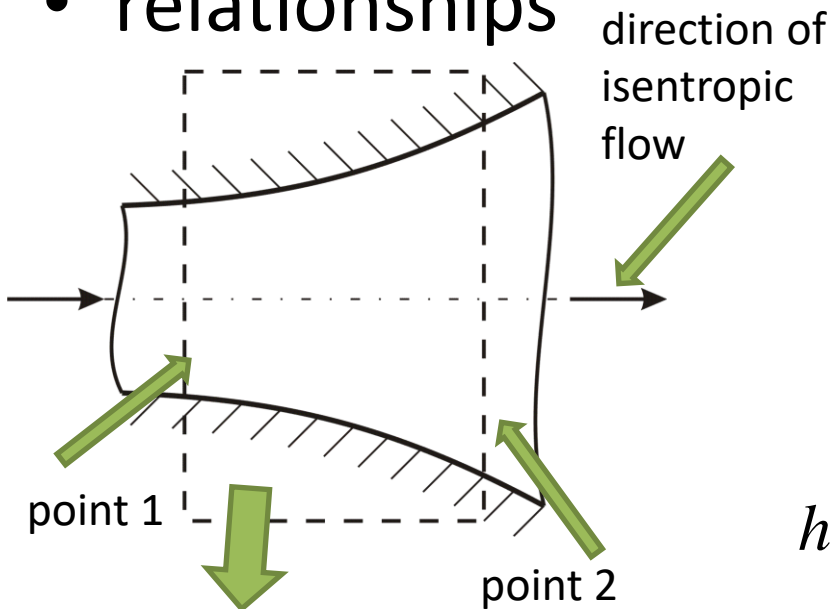
adiabatic energy
equation 1-2

$$\left(h_1 + \frac{u_1^2}{2} \right) = \left(h_2 + \frac{u_2^2}{2} \right)$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



adiabatic energy
equation 1-2

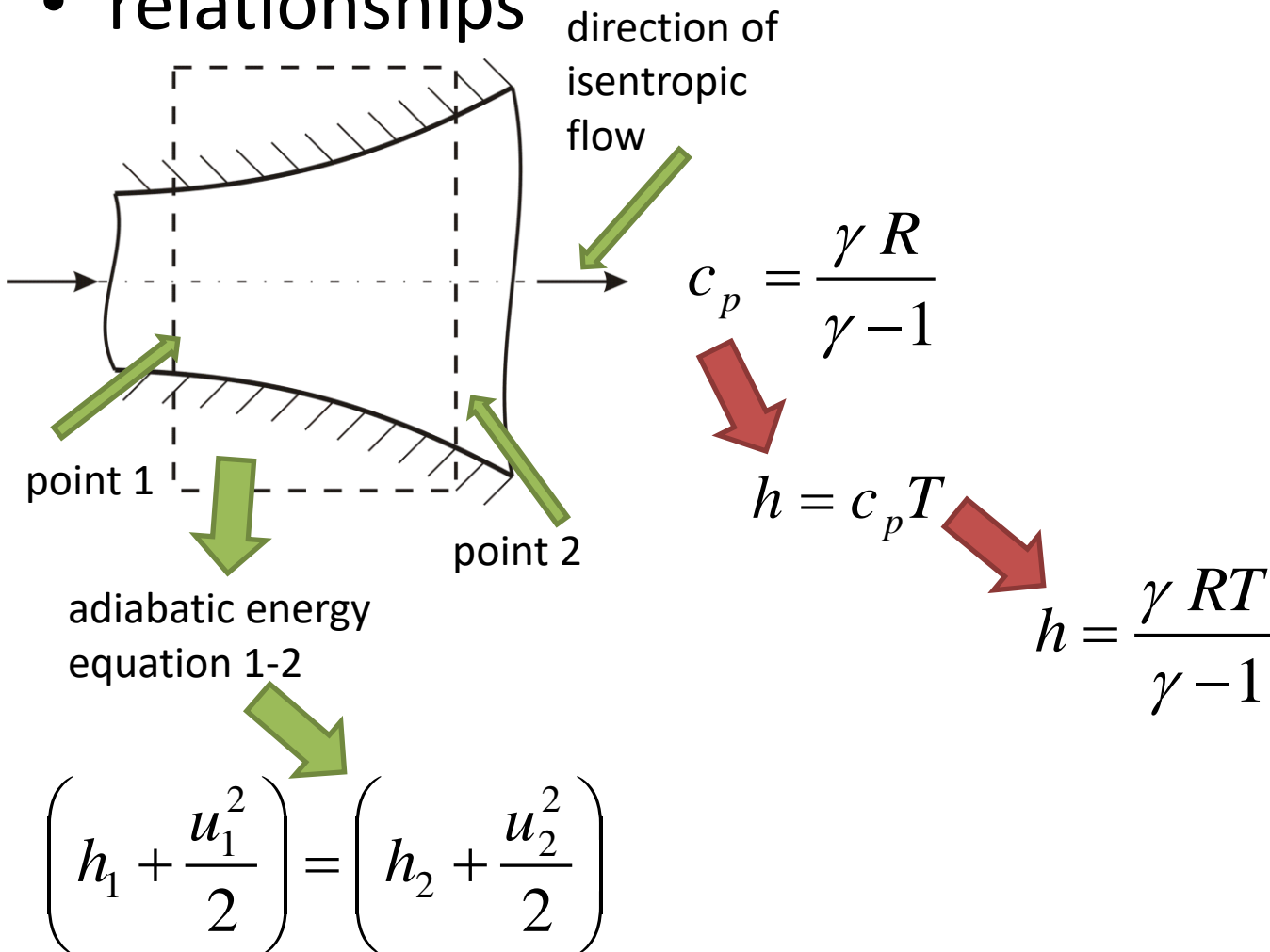
$$\left(h_1 + \frac{u_1^2}{2} \right) = \left(h_2 + \frac{u_2^2}{2} \right)$$

$$h = c_p T$$

1. Fluid Flow and Thermodynamics

Isentropic flow

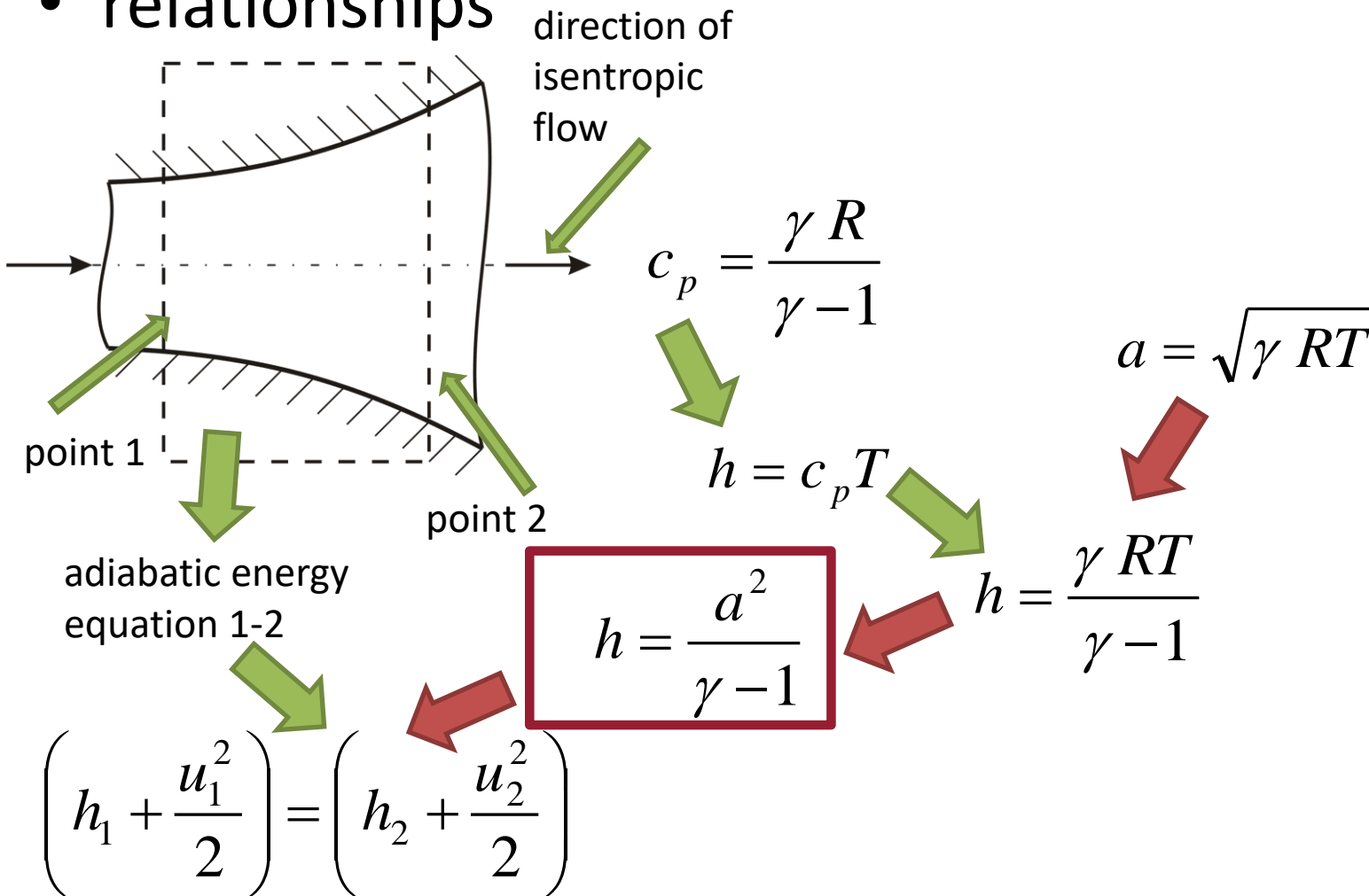
- relationships



1. Fluid Flow and Thermodynamics

Isentropic flow

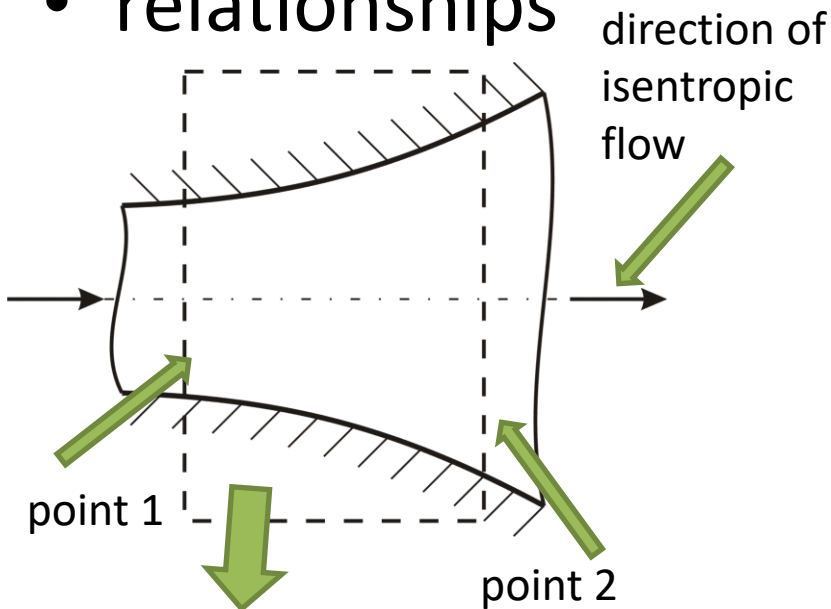
- relationships



1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



adiabatic energy
equation 1-2

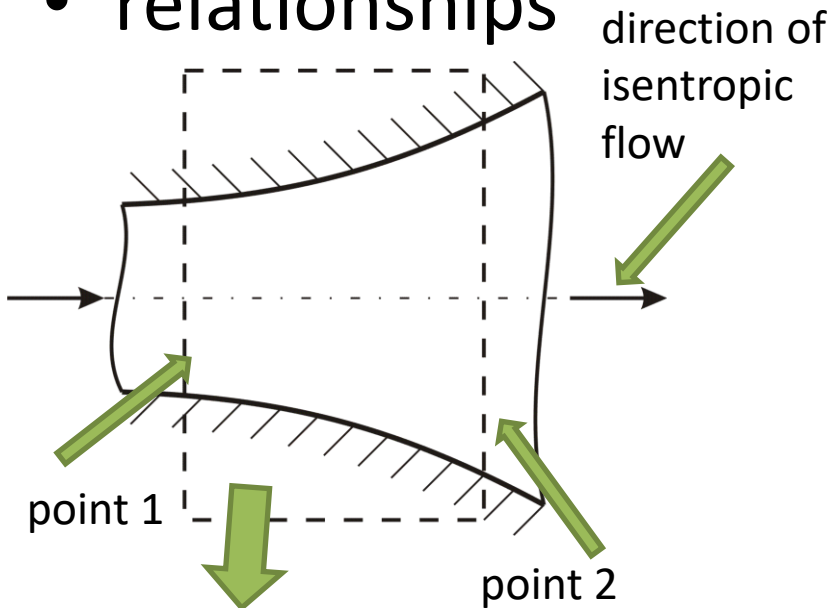
$$h = \frac{a^2}{\gamma - 1}$$

$$\left(h_1 + \frac{u_1^2}{2} \right) = \left(h_2 + \frac{u_2^2}{2} \right) \rightarrow \left(\frac{a_1^2}{\gamma - 1} + \frac{u_1^2}{2} \right) = \left(\frac{a_2^2}{\gamma - 1} + \frac{u_2^2}{2} \right)$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



$$\left(\frac{a^2}{u^2(\gamma-1)} + \frac{1}{2} \right) = \frac{\gamma+1}{2(\gamma-1)} \frac{a^{*2}}{u^2}$$

$$\left(\frac{a^2}{\gamma-1} + \frac{u^2}{2} \right) = \frac{\gamma+1}{2(\gamma-1)} a^{*2}$$

$$h = \frac{a^2}{\gamma-1}$$

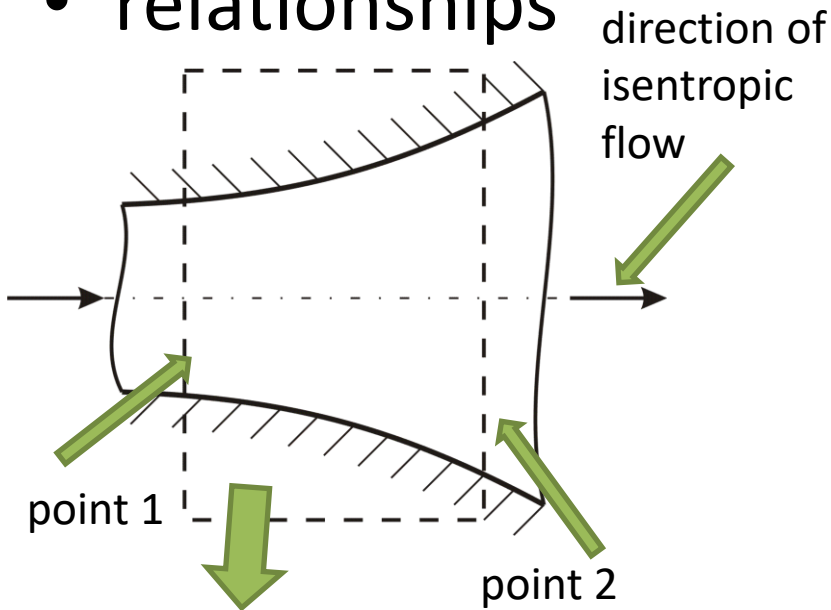
critical point

$$\left(h_1 + \frac{u_1^2}{2} \right) = \left(h_2 + \frac{u_2^2}{2} \right) \rightarrow \left(\frac{a_1^2}{\gamma-1} + \frac{u_1^2}{2} \right) = \left(\frac{a_2^2}{\gamma-1} + \frac{u_2^2}{2} \right)$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



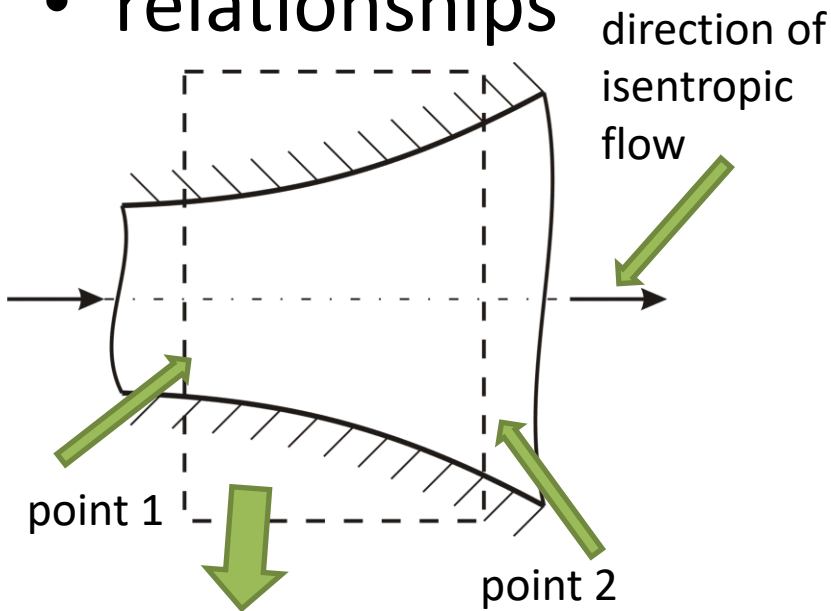
$$\left(\frac{a^2}{u^2(\gamma - 1)} + \frac{1}{2} \right) = \frac{\gamma + 1}{2(\gamma - 1)} \frac{a^{*2}}{u^2}$$

$$\left(\frac{1}{M^2(\gamma - 1)} + \frac{1}{2} \right) = \frac{\gamma + 1}{2(\gamma - 1)} \frac{1}{(M^*)^2}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



$$\left(\frac{a^2}{u^2(\gamma - 1)} + \frac{1}{2} \right) = \frac{\gamma + 1}{2(\gamma - 1)} \frac{a^{*2}}{u^2}$$

$$\left(\frac{1}{M^2(\gamma - 1)} + \frac{1}{2} \right) = \frac{\gamma + 1}{2(\gamma - 1)} \frac{1}{(M^*)^2}$$

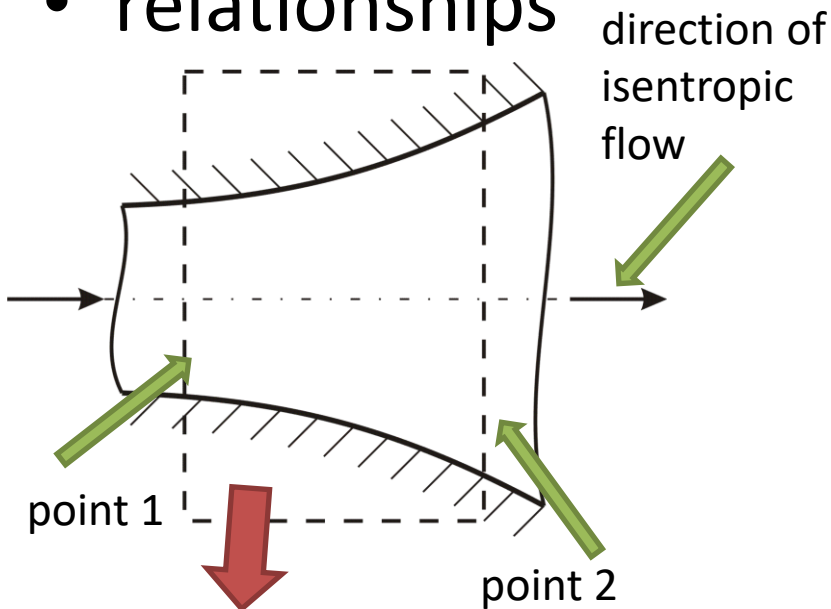
$$(M^*)^2 = \frac{(\gamma + 1)M^2}{2 + (\gamma - 1)M^2}$$

$$M^2 = \frac{2}{(\gamma + 1)/(M^*)^2 - (\gamma - 1)}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



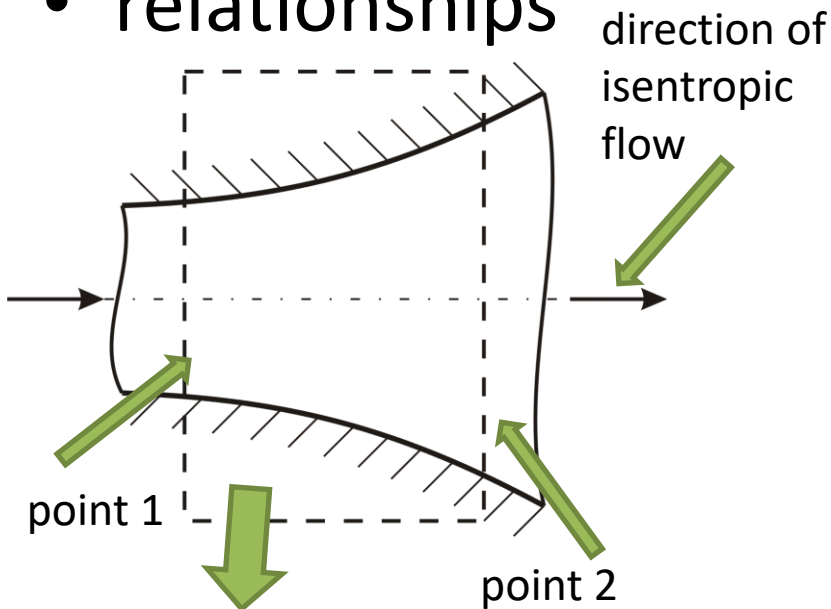
adiabatic energy equation

$$\left(h + \frac{u^2}{2} \right) = h_0 \rightarrow \left(c_p T + \frac{u^2}{2} \right) = c_p T_0$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



adiabatic energy equation

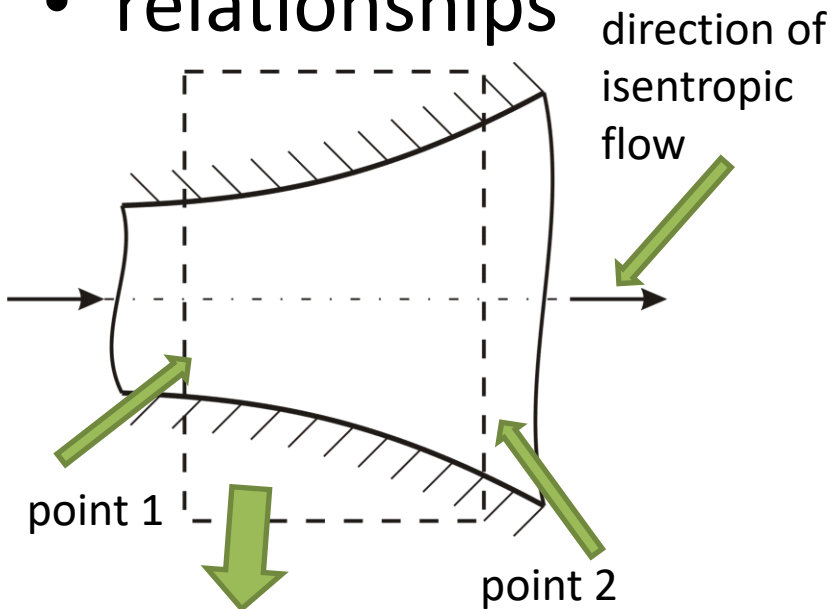
$$\left(h + \frac{u^2}{2} \right) = h_0 \longrightarrow \left(c_p T + \frac{u^2}{2} \right) = c_p T_0$$

$$1 + \frac{u^2}{2c_p T} = \frac{T_0}{T}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



$$1 + \frac{(\gamma - 1)u^2}{2\gamma RT} = \frac{T_0}{T}$$

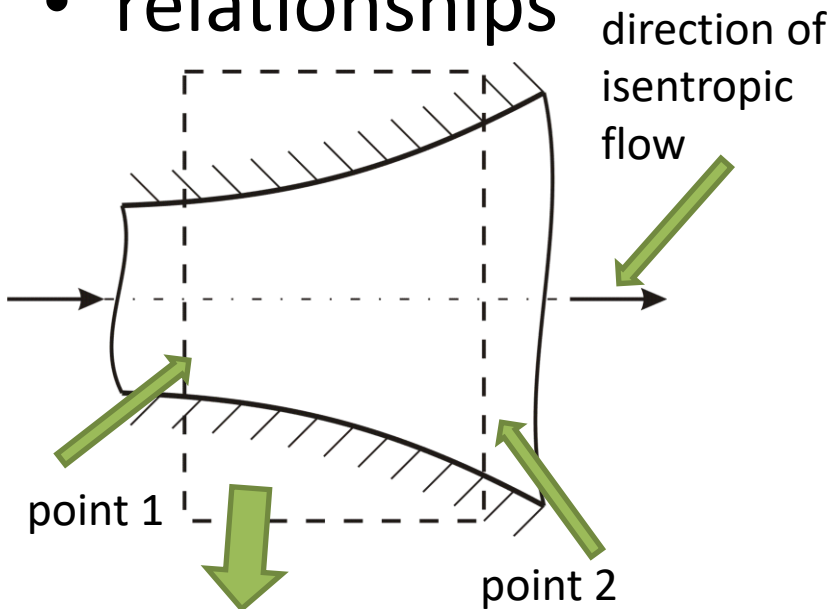
$$1 + \frac{u^2}{2c_p T} = \frac{T_0}{T}$$

$$\left(h + \frac{u^2}{2} \right) = h_0 \rightarrow \left(c_p T + \frac{u^2}{2} \right) = c_p T_0$$
$$c_p = \frac{\gamma R}{\gamma - 1}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



adiabatic energy equation

$$\left(h + \frac{u^2}{2} \right) = h_0 \longrightarrow \left(c_p T + \frac{u^2}{2} \right) = c_p T_0$$

$$1 + \frac{(\gamma - 1)u^2}{2\gamma RT} = \frac{T_0}{T}$$

$$1 + \frac{(\gamma - 1)}{2} M^2 = \frac{T_0}{T}$$

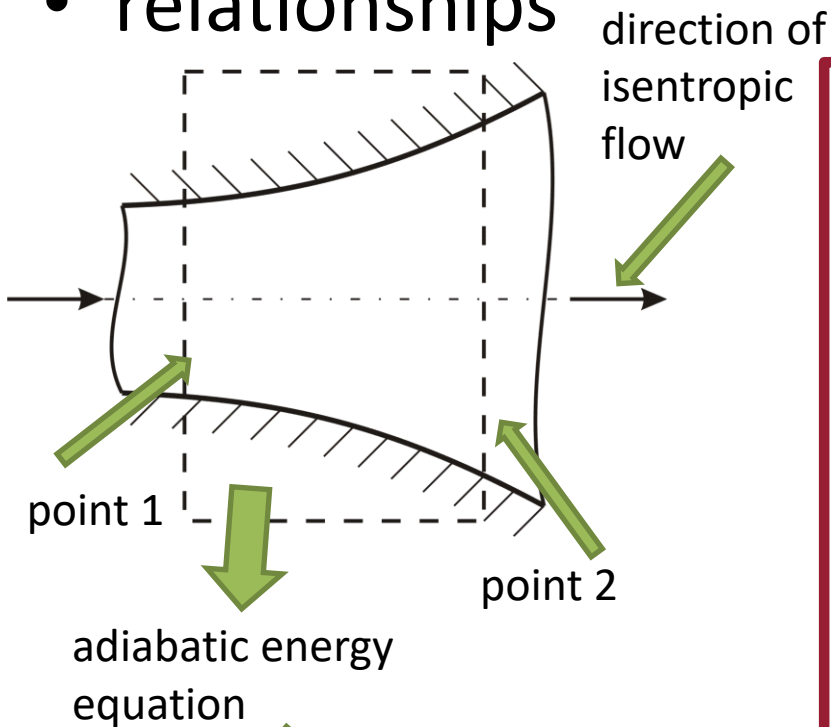
$$1 + \frac{u^2}{2c_p T} = \frac{T_0}{T}$$

$$c_p = \frac{\gamma R}{\gamma - 1}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



$$\left(1 + \frac{(\gamma - 1)}{2} M^2\right)^{\frac{1}{\gamma - 1}} = \frac{\rho_0}{\rho}$$

$$\left(1 + \frac{(\gamma - 1)}{2} M^2\right)^{\frac{\gamma}{\gamma - 1}} = \frac{p_0}{p}$$

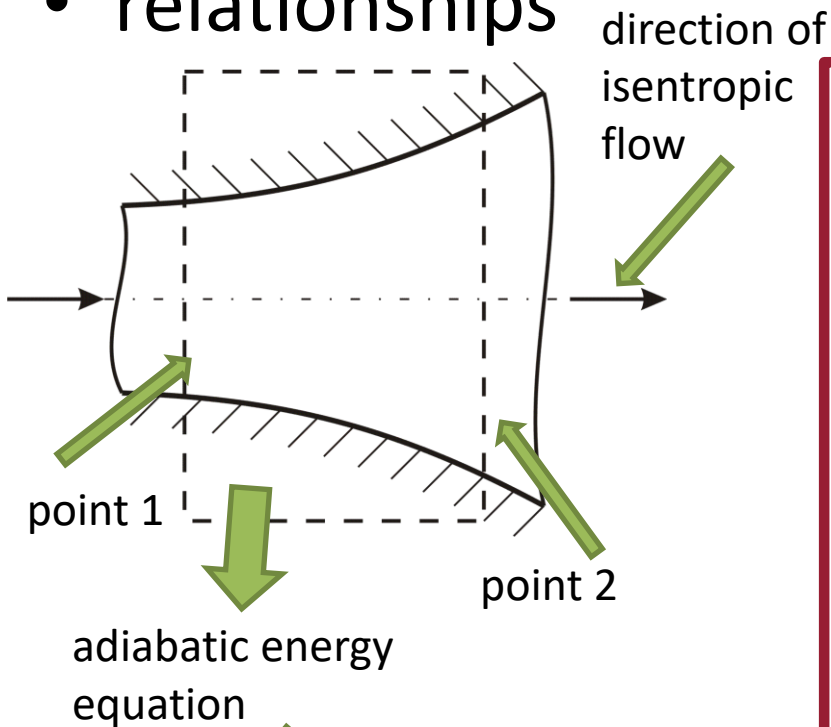
$$1 + \frac{(\gamma - 1)}{2} M^2 = \frac{T_0}{T}$$

$$\left(h + \frac{u^2}{2}\right) = h_0 \rightarrow \left(c_p T + \frac{u^2}{2}\right) = c_p T_0 \quad \frac{p_0}{p} = \left(\frac{T_0}{T}\right)^{\frac{\gamma}{\gamma - 1}} = \left(\frac{\rho_0}{\rho}\right)^\gamma$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



$$\left(1 + \frac{(\gamma - 1)}{2} M^2\right)^{\frac{1}{\gamma - 1}} = \frac{\rho_0}{\rho}$$

$$\left(1 + \frac{(\gamma - 1)}{2} M^2\right)^{\frac{\gamma}{\gamma - 1}} = \frac{p_0}{p}$$

$$1 + \frac{(\gamma - 1)}{2} M^2 = \frac{T_0}{T}$$

if $M = 1$

$$\rho = \rho^*$$

$$p = p^*$$

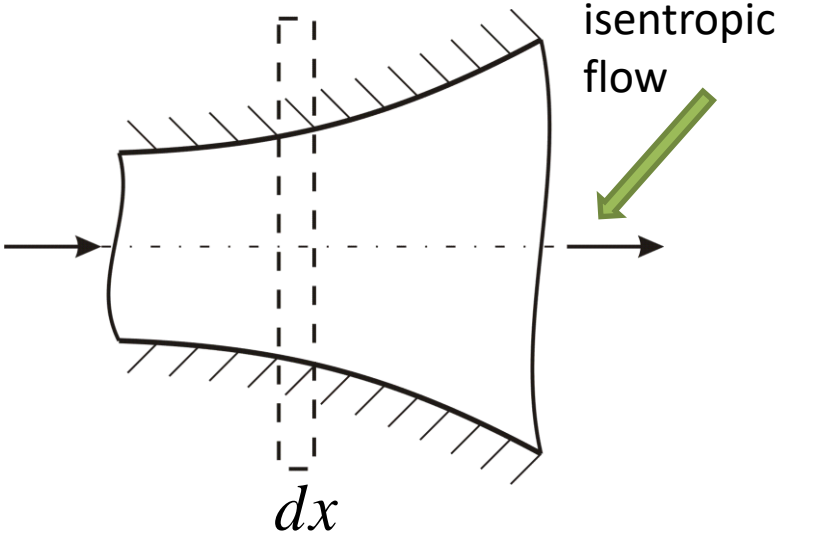
$$T = T^*$$

$$\left(h + \frac{u^2}{2}\right) = h_0 \rightarrow \left(c_p T + \frac{u^2}{2}\right) = c_p T_0 \rightarrow \frac{p_0}{p} = \left(\frac{T_0}{T}\right)^{\frac{\gamma}{\gamma - 1}} = \left(\frac{\rho_0}{\rho}\right)^\gamma$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



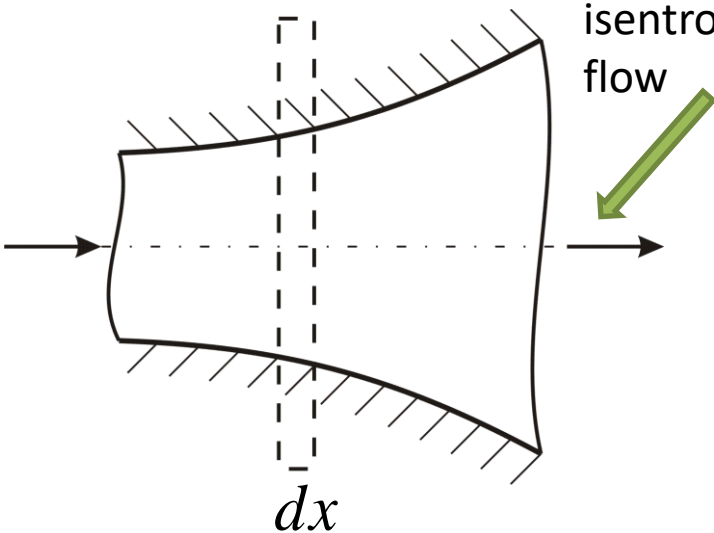
momentum equation

$$-\frac{dp}{\rho} = u du$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



direction of
isentropic
flow

momentum equation

$$-\frac{dp}{\rho} = u du$$

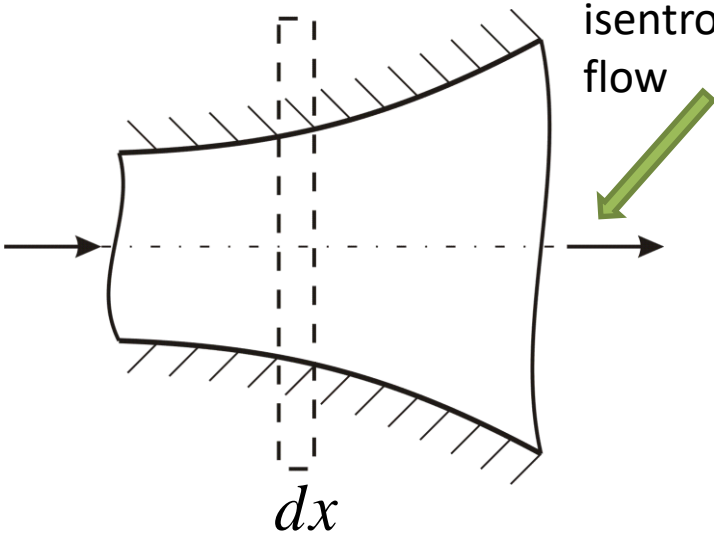


$$-\frac{dp}{p} = \frac{\rho u}{p u} u du$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



direction of
isentropic
flow

momentum equation

speed of sound

$$-\frac{dp}{\rho} = u du$$

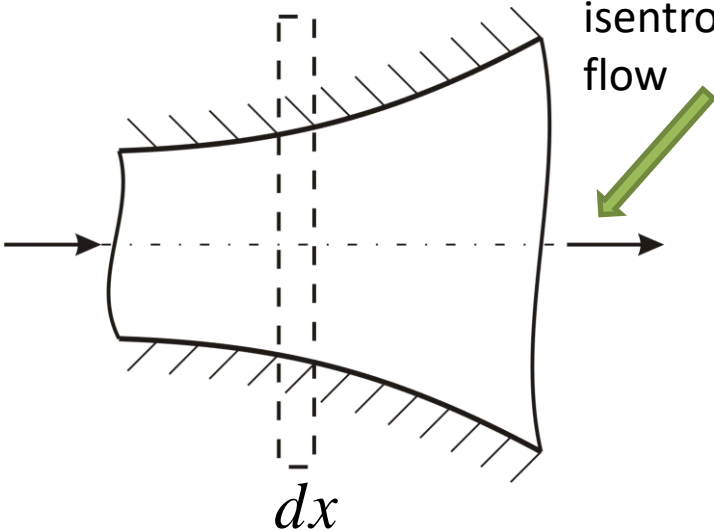
$$a^2 = \gamma RT = \gamma \frac{p}{\rho}$$

$$-\frac{dp}{p} = \frac{\rho u}{p u} u du$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



momentum equation

$$-\frac{dp}{\rho} = u du$$

speed of sound

$$a^2 = \gamma RT = \gamma \frac{p}{\rho}$$

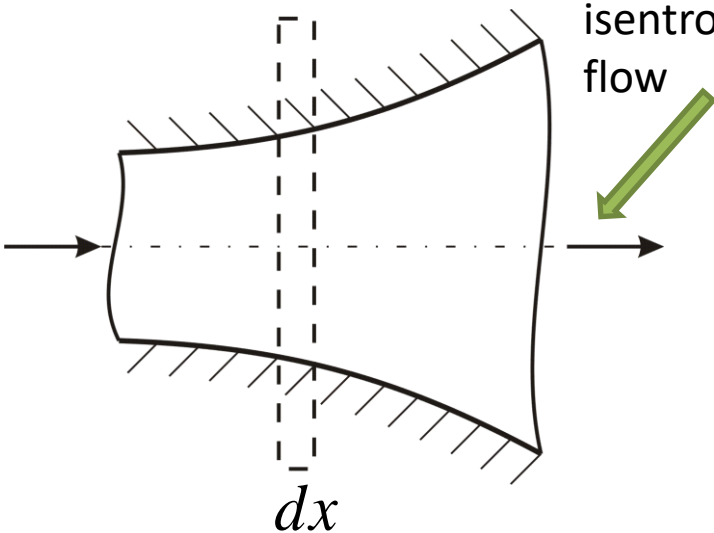
$$-\frac{dp}{p} = \frac{\rho u}{p u} u du$$

$$\frac{dp}{p} = -\gamma \frac{u^2}{a^2} \frac{du}{u}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



momentum equation

$$-\frac{dp}{\rho} = u du$$

speed of sound

$$a^2 = \gamma RT = \gamma \frac{p}{\rho}$$

$$-\frac{dp}{p} = \frac{\rho u}{\rho u} u du$$

$$\frac{dp}{p} = -\gamma \frac{u^2}{a^2} \frac{du}{u}$$

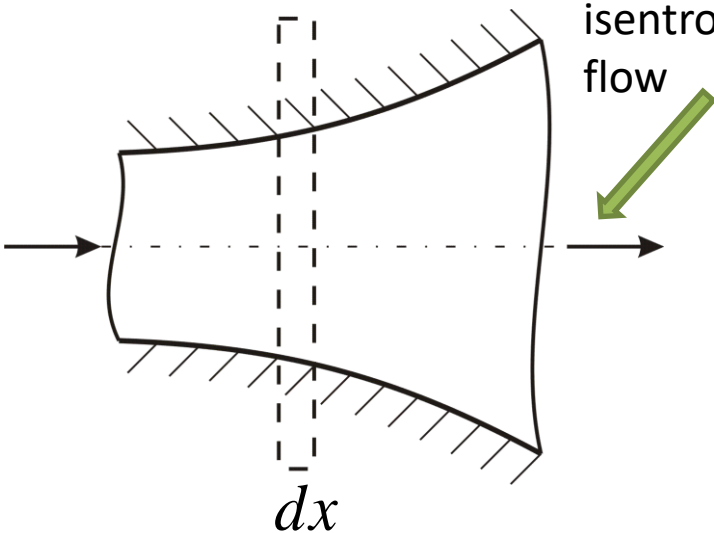
$$\frac{dp}{p} = -\gamma M^2 \frac{du}{u}$$

• magnitude of fractional pressure change induced by a given fractional velocity change depends on square of Mach number

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



direction of
isentropic
flow

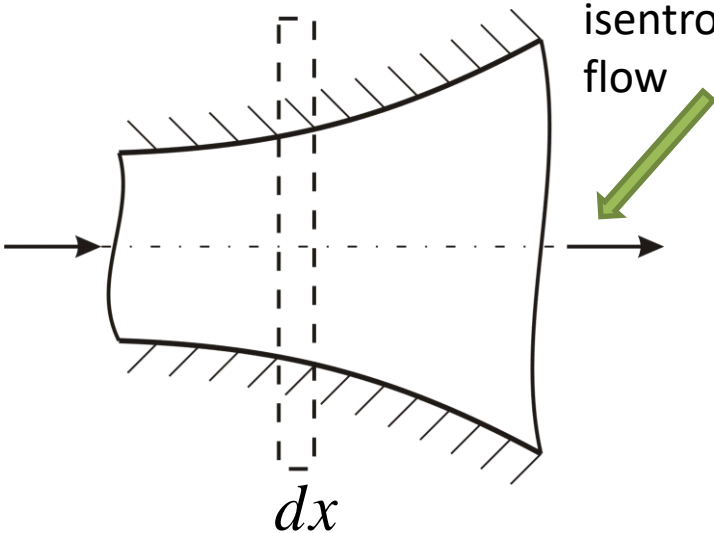
energy equation

$$dh + udu = 0$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



energy equation

$$dh + udu = 0$$

calorically perfect gas

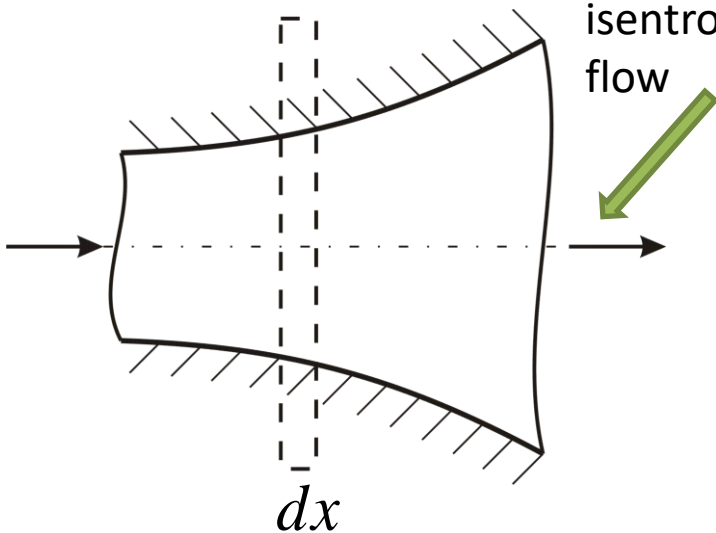
$$dh = c_p dT$$



1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



energy equation

$$dh + udu = 0$$

calorically perfect gas

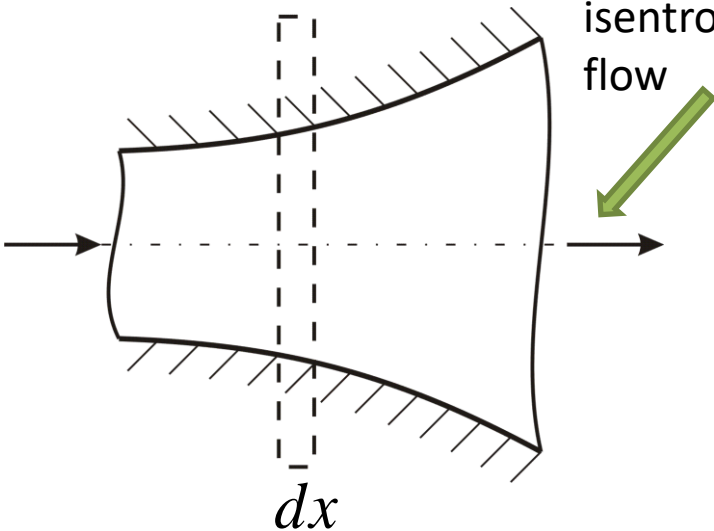
$$dh = c_p dT$$

$$c_p dT = -udu$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



energy equation

$$dh + udu = 0$$

calorically perfect gas

$$dh = c_p dT$$

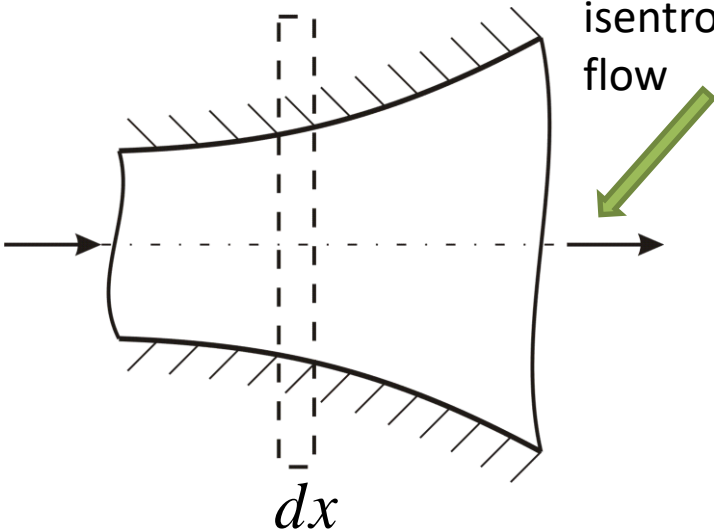
$$c_p dT = -u du$$

$$\frac{dT}{T} = -\frac{u}{c_p T} \frac{u}{u} du$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



energy equation

$$dh + udu = 0$$

calorically perfect gas

$$dh = c_p dT$$

$$c_p dT = -u du$$

$$\frac{dT}{T} = -\frac{u}{c_p T} \frac{u}{u} du$$

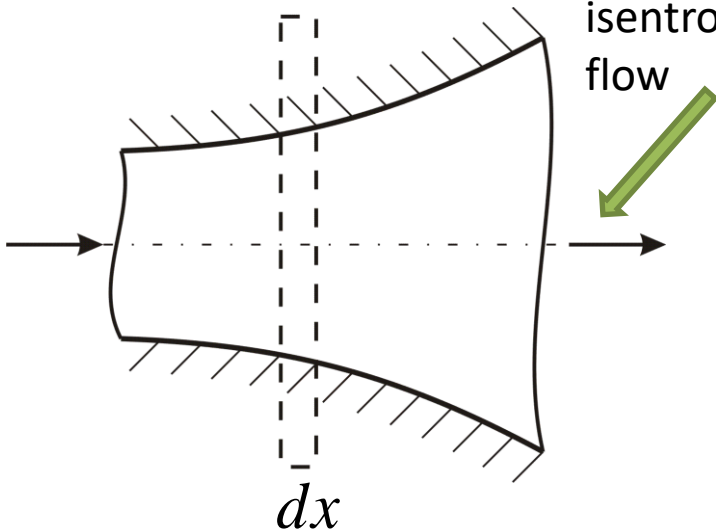
speed of sound

$$a^2 = \gamma RT = \gamma \frac{p}{\rho}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



energy equation

$$dh + u du = 0$$

calorically perfect gas

$$dh = c_p dT$$

$$c_p dT = -u du$$

speed of sound

$$a^2 = \gamma R T = \gamma \frac{p}{\rho}$$

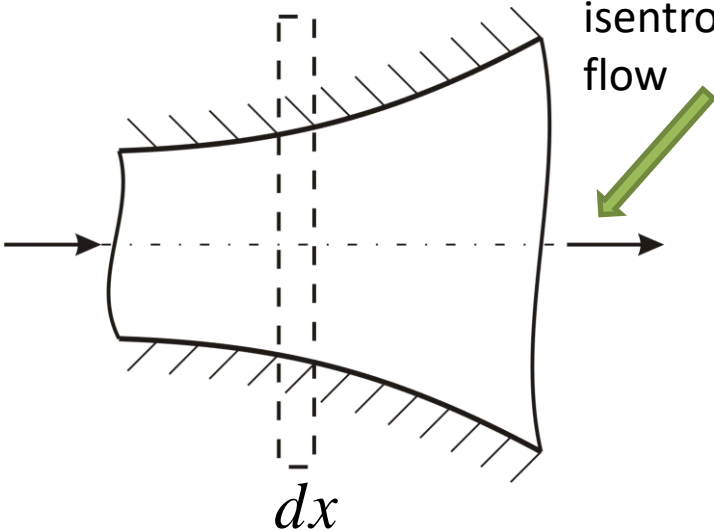
$$\frac{dT}{T} = -\frac{u}{c_p T} \frac{u}{u} du$$

$$\frac{dT}{T} = -\frac{\gamma R u^2}{c_p a^2} \frac{du}{u}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



energy equation

$$dh + u du = 0$$

calorically perfect gas

$$dh = c_p dT$$

$$c_p dT = -u du$$

speed of sound
 $a^2 = \gamma R T = \gamma \frac{p}{\rho}$

$$\frac{dT}{T} = -\frac{u}{c_p T} \frac{u}{u} du$$

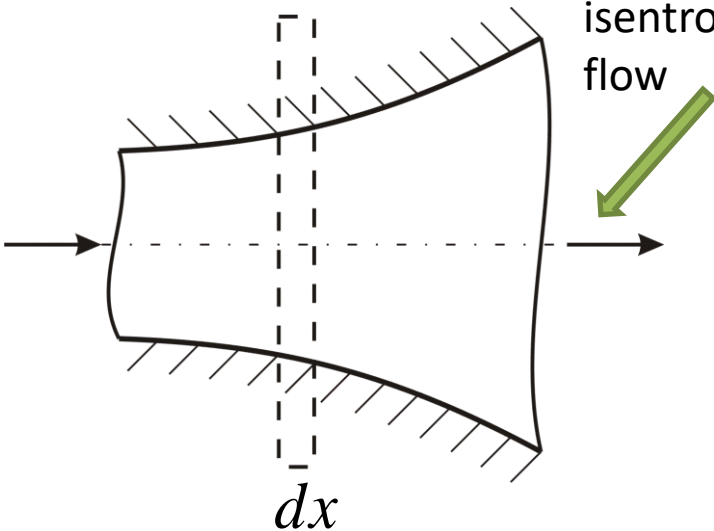
$$\frac{dT}{T} = -\frac{\gamma R u^2}{c_p a^2} \frac{du}{u}$$

$$\frac{dT}{T} = -\frac{\gamma R}{c_p} M^2 \frac{du}{u}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

relationships



energy equation

$$dh + udu = 0$$

calorically perfect gas

$$dh = c_p dT$$

$$c_p dT = -u du$$

speed of sound

$$a^2 = \gamma R T = \gamma \frac{p}{\rho}$$

$$\frac{dT}{T} = -\frac{u}{c_p T} \frac{u}{u} du$$

$$\frac{dT}{T} = -(\gamma - 1) M^2 \frac{du}{u}$$

$$\frac{dT}{T} = -\frac{\gamma R u^2}{c_p a^2} \frac{du}{u}$$

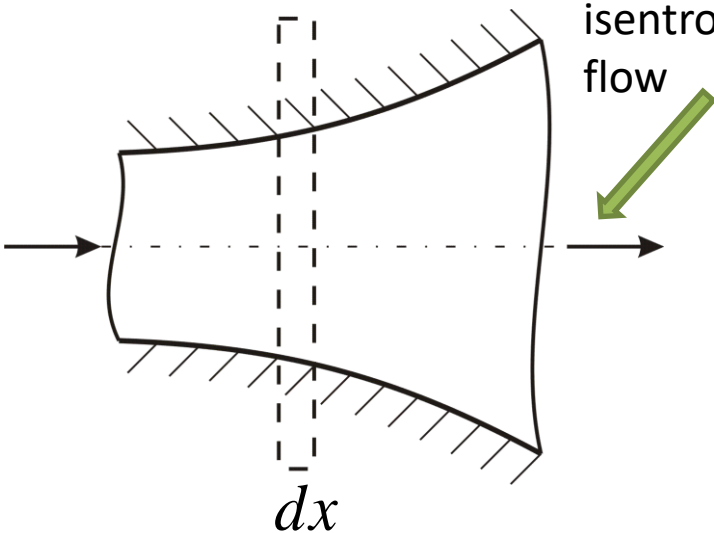
$$\frac{dT}{T} = -\frac{\gamma R}{c_p} M^2 \frac{du}{u}$$

• magnitude of fractional temperature change induced by a given fractional velocity change depends on square of Mach number

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



direction of
isentropic
flow

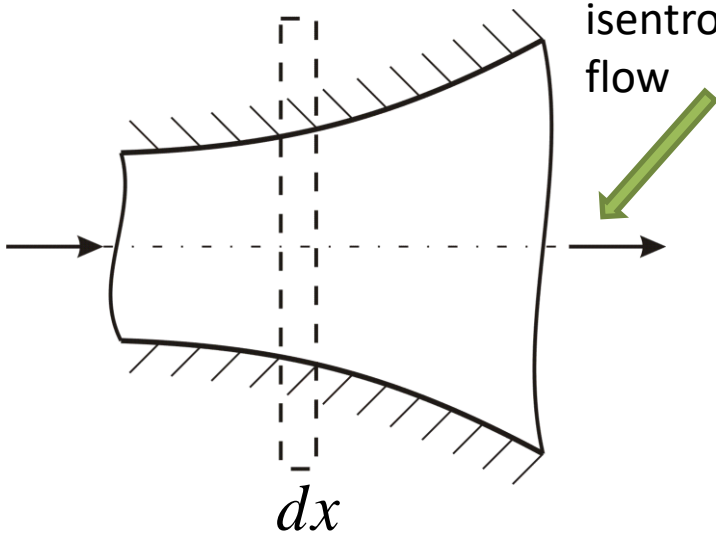
equation of state

$$\frac{p}{\rho T} = R$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



equation of state

$$\frac{p}{\rho T} = R$$

$$\frac{dp}{p} - \frac{d\rho}{\rho} - \frac{dT}{T} = 0$$

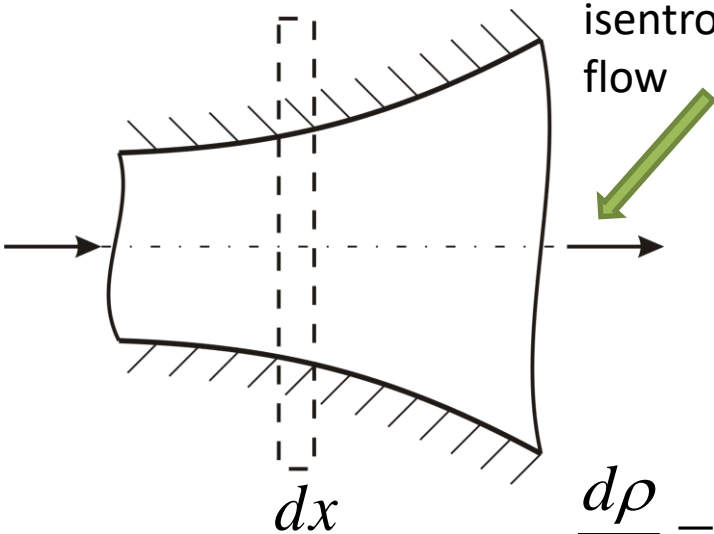
$$\frac{dT}{T} = -(\gamma - 1)M^2 \frac{du}{u}$$

$$\frac{dp}{p} = -\gamma M^2 \frac{du}{u}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



equation of state

$$\frac{p}{\rho T} = R$$

$$\frac{dp}{p} - \frac{d\rho}{\rho} - \frac{dT}{T} = 0$$

$$\frac{dT}{T} = -(\gamma - 1)M^2 \frac{du}{u}$$

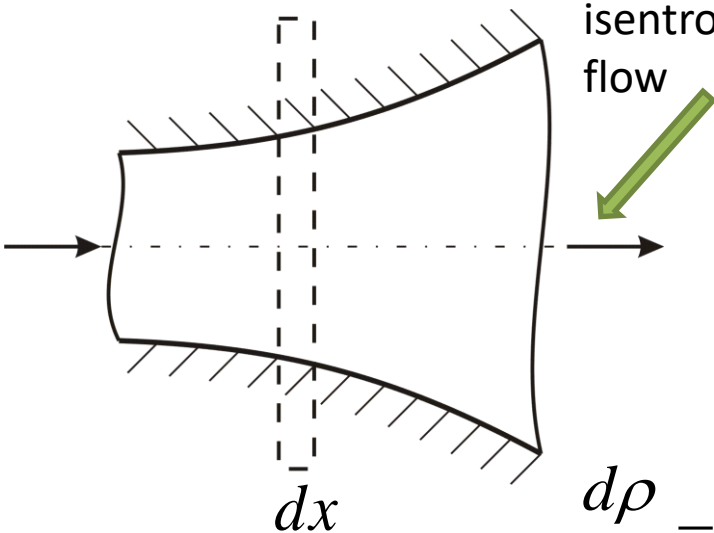
$$\frac{dp}{p} = -\gamma M^2 \frac{du}{u}$$

$$\frac{d\rho}{\rho} = -\gamma M^2 \frac{du}{u} + (\gamma - 1)M^2 \frac{du}{u}$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



equation of state

$$\frac{p}{\rho T} = R$$

$$\frac{dT}{T} = -(\gamma - 1) M^2 \frac{du}{u}$$

$$\frac{dp}{p} - \frac{d\rho}{\rho} - \frac{dT}{T} = 0$$

$$\frac{dp}{p} = -\gamma M^2 \frac{du}{u}$$

$$\frac{d\rho}{\rho} = -\gamma M^2 \frac{du}{u} + (\gamma - 1) M^2 \frac{du}{u}$$

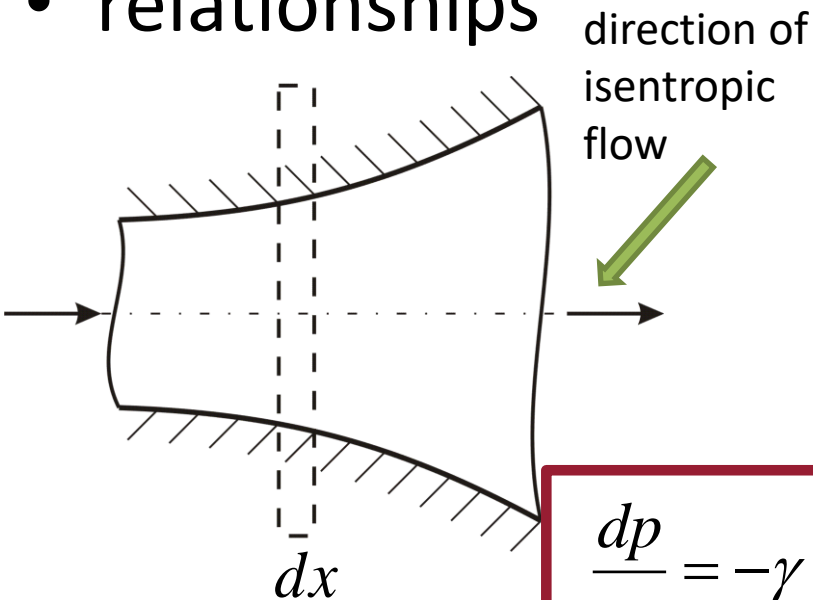
$$\frac{d\rho}{\rho} = -M^2 \frac{du}{u}$$

- magnitude of fractional temperature change induced by a given fractional velocity change depends on square of Mach number

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



- magnitude of fractional properties change induced by a given fractional velocity change depends on square of Mach number

$$\frac{dp}{p} = -\gamma M^2 \frac{du}{u}$$

$$\frac{dT}{T} = -(\gamma - 1) M^2 \frac{du}{u}$$

$$\frac{d\rho}{\rho} = -M^2 \frac{du}{u}$$



$$\frac{dp/p}{du/u} = -\gamma M^2$$

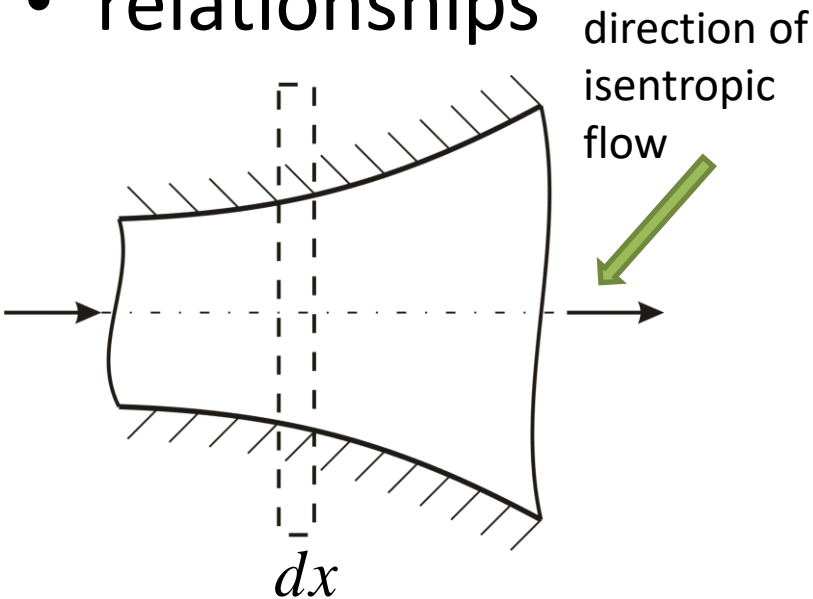
$$\frac{dT/T}{du/u} = -(\gamma - 1) M^2$$

$$\frac{d\rho/\rho}{du/u} = -M^2$$

1. Fluid Flow and Thermodynamics

Isentropic flow

- relationships



- magnitude of fractional properties change induced by a given fractional velocity change depends on square of Mach number

	fractional propert. change induced by fractional velocity change of air [%]		
Mach num.	density	temp.	pressure
0.1	1	1.4	0.4
0.33	10.9	15.2	4.4
0.4	16	22.4	6.4

$$\frac{dp/p}{du/u} = -\gamma M^2$$

$$\frac{dT/T}{du/u} = -(\gamma - 1)M^2$$

$$\frac{d\rho/\rho}{du/u} = -M^2$$



1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

continuity

$$\frac{d\rho}{\rho} + \frac{du}{u} + \frac{dA}{A} = 0$$

momentum

$$-\frac{dp}{\rho} = u du$$

energy

$$dh + u du = 0$$

state

$$\frac{dp}{p} - \frac{d\rho}{\rho} - \frac{dT}{T} = 0$$

1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$\frac{d\rho}{\rho} + \frac{du}{u} + \frac{dA}{A} = 0$$
$$\frac{d\rho}{\rho} = -M^2 \frac{du}{u}$$
$$-\frac{dp}{\rho} = u du$$
$$dh + u du = 0$$
$$\frac{dp}{p} - \frac{d\rho}{\rho} - \frac{dT}{T} = 0$$

1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$\frac{d\rho}{\rho} + \frac{du}{u} + \frac{dA}{A} = 0 \quad \Rightarrow \quad -M^2 \frac{du}{u} + \frac{du}{u} + \frac{dA}{A} = 0$$
$$\frac{d\rho}{\rho} = -M^2 \frac{du}{u} \quad \leftarrow \quad -\frac{dp}{\rho} = u du$$
$$dh + u du = 0$$
$$\frac{dp}{p} - \frac{d\rho}{\rho} - \frac{dT}{T} = 0$$

Nozzle fluid flow

Exit of nozzle

$$\frac{dA}{A} = (M^2 - 1) \frac{du}{u}$$

1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$\frac{dA}{A} = (M^2 - 1) \frac{du}{u}$$



1. subsonic flow: $0 \leq M < 1$



increase in velocity $+du$ is associated with decrease in area $-dA$

2. supersonic flow: $M > 1$



increase in velocity $+du$ is associated with increase in area $+dA$

3. sonic flow: $M = 1$



$dA = 0$ area reaches an extremum – minimum

1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$\frac{dA}{A} = (M^2 - 1) \frac{du}{u}$$



increase in velocity $+du$ is associated with decrease in area $-dA$

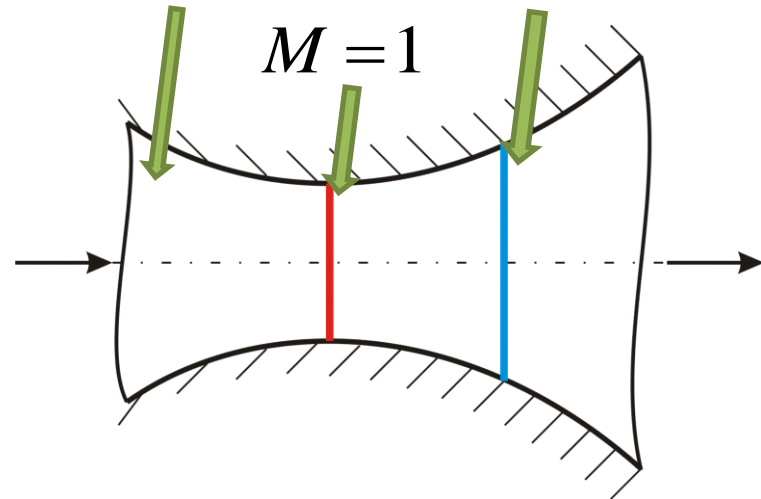
$0 \leq M < 1$ $M > 1$

$M = 1$



increase in velocity $+du$ is associated with increase in area $+dA$

$dA = 0$ area reaches an extremum – minimum

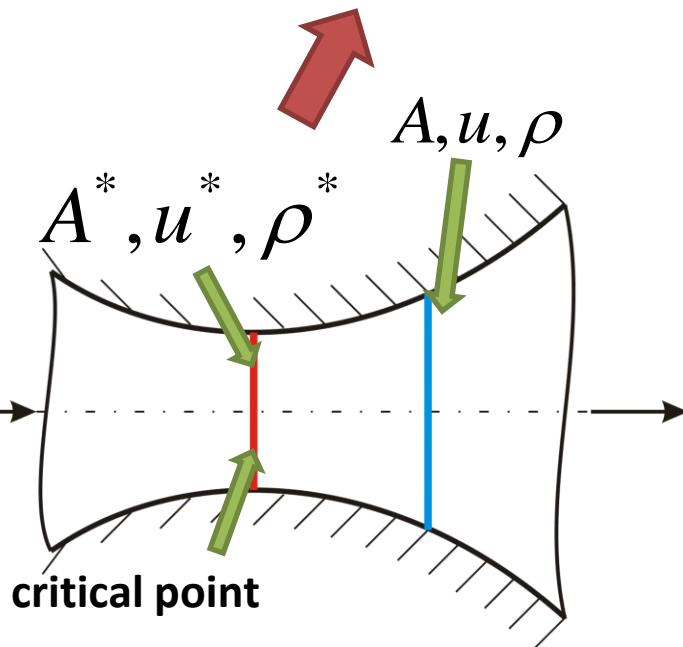


1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$A^* u^* \rho^* = A u \rho$$

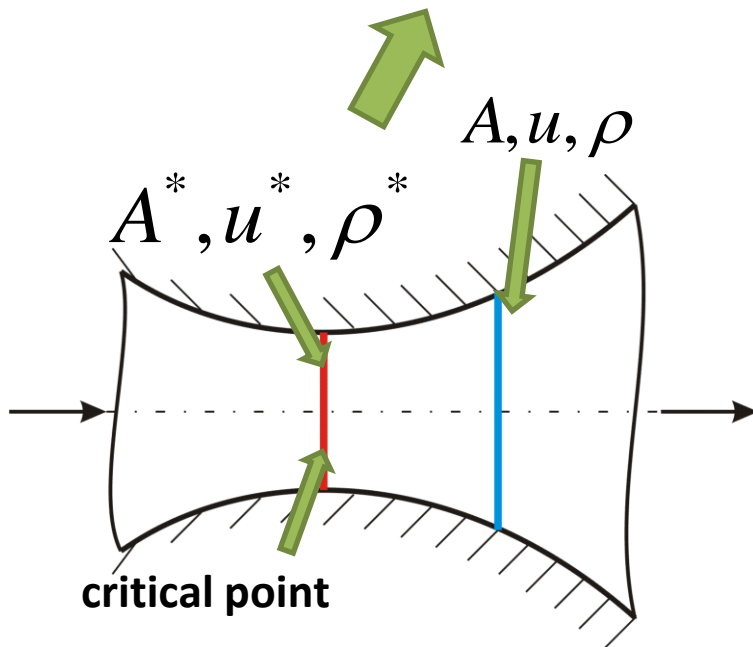


1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$A^* u^* \rho^* = A u \rho \quad \Rightarrow \quad \frac{A}{A^*} = \frac{u^*}{u} \frac{\rho^*}{\rho} = \frac{a^*}{u} \frac{\rho^*}{\rho_0} \frac{\rho_0}{\rho}$$

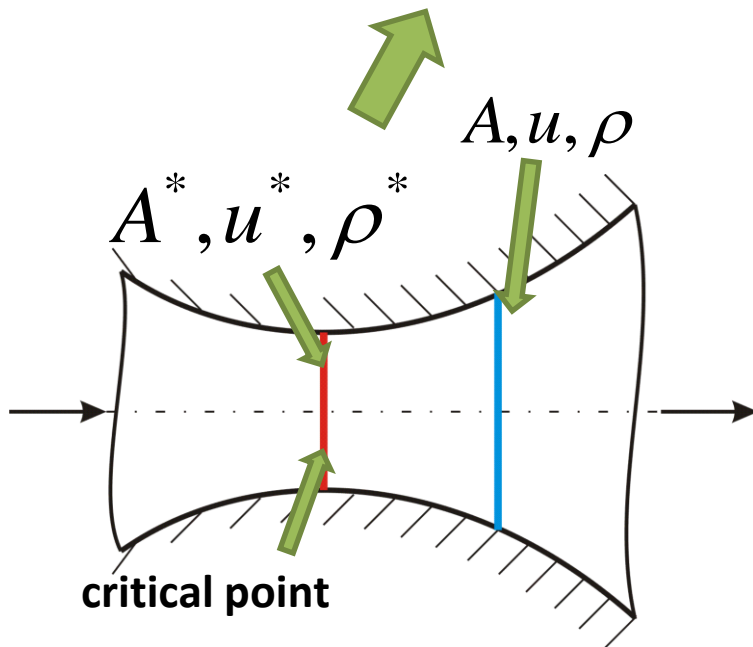


1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$A^* u^* \rho^* = A u \rho \quad \Rightarrow \quad \frac{A}{A^*} = \frac{u^*}{u} \frac{\rho^*}{\rho} = \frac{a^*}{u} \frac{\rho^*}{\rho_0} \frac{\rho_0}{\rho}$$



$$\frac{\rho_0}{\rho} = \left(1 + \frac{(\gamma - 1)}{2} M^2 \right)^{\frac{1}{\gamma - 1}}$$

$$\frac{\rho_0}{\rho^*} = \left(1 + \frac{(\gamma - 1)}{2} \right)^{\frac{1}{\gamma - 1}}$$

$$(M^*)^2 = \left(\frac{u}{a^*} \right)^2 = \frac{(\gamma + 1) M^2}{2 + (\gamma - 1) M^2}$$

1. Fluid Flow and Thermodynamics

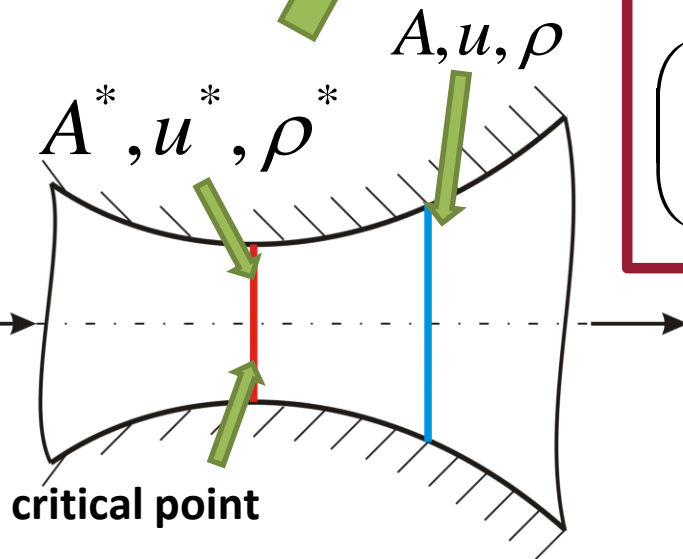
Nozzle fluid flow

- governing equations for analysis of nozzle

$$A^* u^* \rho^* = A u \rho \quad \Rightarrow \quad \frac{A}{A^*} = \frac{u^*}{u} \frac{\rho^*}{\rho} = \frac{a^*}{u} \frac{\rho^*}{\rho_0} \frac{\rho_0}{\rho}$$

$$\left(\frac{A}{A^*} \right)^2 = \frac{1}{M^2} \left[\frac{2}{\gamma + 1} \left(1 + \frac{\gamma - 1}{2} M^2 \right) \right]^{\frac{\gamma + 1}{\gamma - 1}}$$

area – Mach number relation

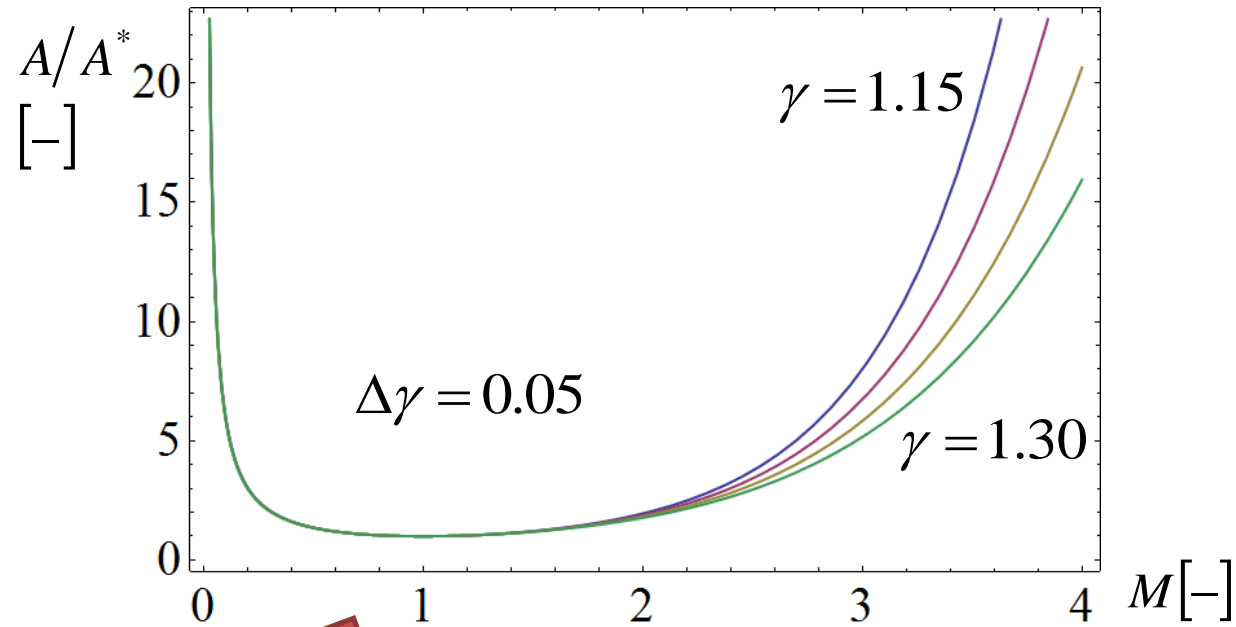
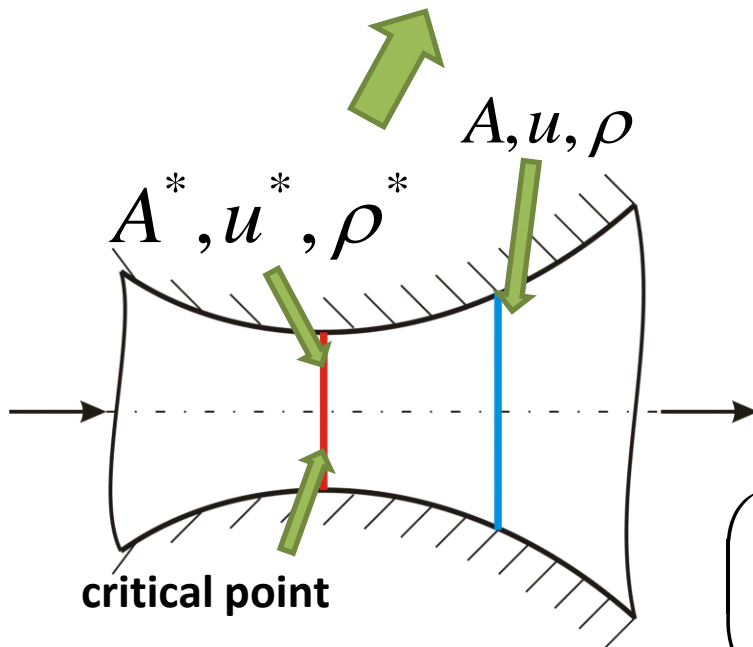


1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- governing equations for analysis of nozzle

$$A^* u^* \rho^* = A u \rho$$



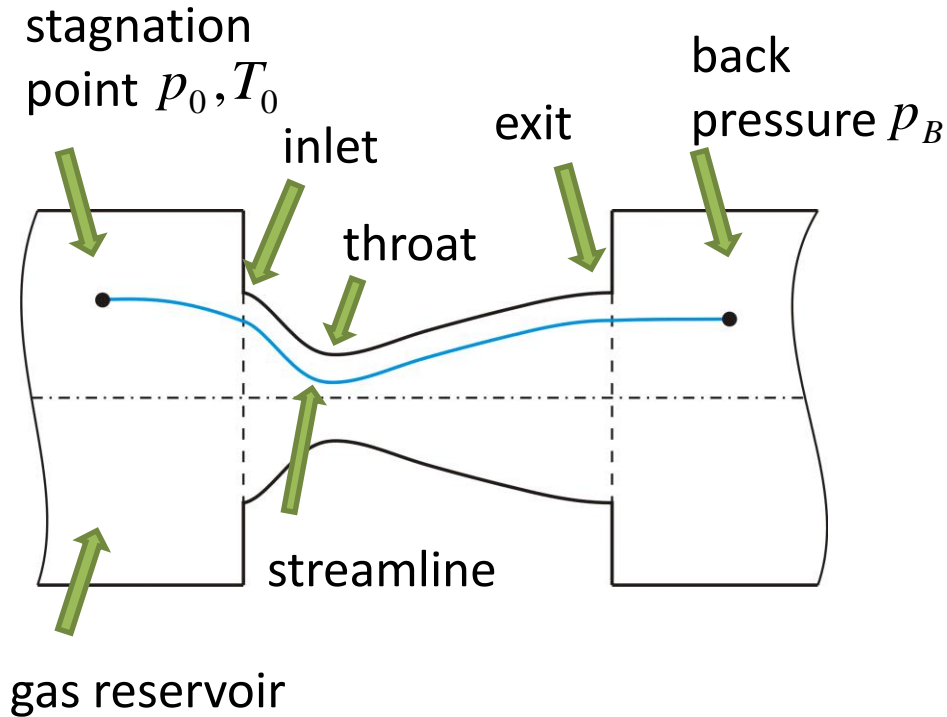
area – Mach number relation

$$\left(\frac{A}{A^*}\right)^2 = \frac{1}{M^2} \left[\frac{2}{\gamma+1} \left(1 + \frac{\gamma-1}{2} M^2 \right) \right]^{\frac{(\gamma+1)}{(\gamma-1)}}$$

1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- variation of parameters in nozzle



- stagnation parameters:

$$T_0 = 500^\circ\text{K}$$

$$p_0 = 1\text{MPa}$$

- nozzle area:

$$\text{inlet area (location 0 m): } 0.004 \text{ m}^2$$

$$\text{throat area (loc. 0.05 m): } 0.002 \text{ m}^2$$

$$\text{exit area (loc. 0.2 m): } 0.004 \text{ m}^2$$

- air:

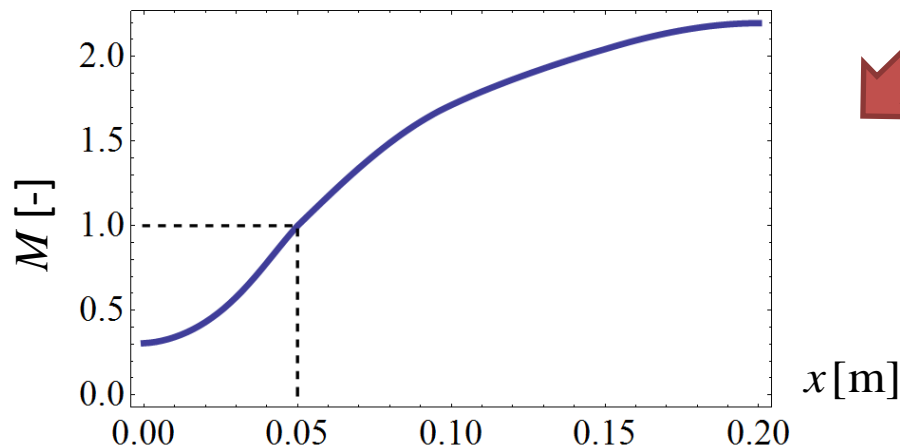
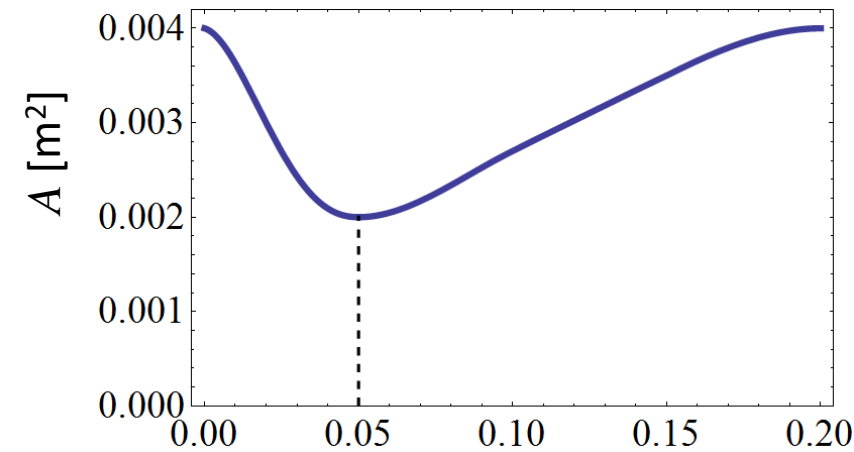
$$R = 288 \text{ J/kgK}$$

$$\gamma = 1.4$$

1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- variation of parameters in nozzle



x [m]

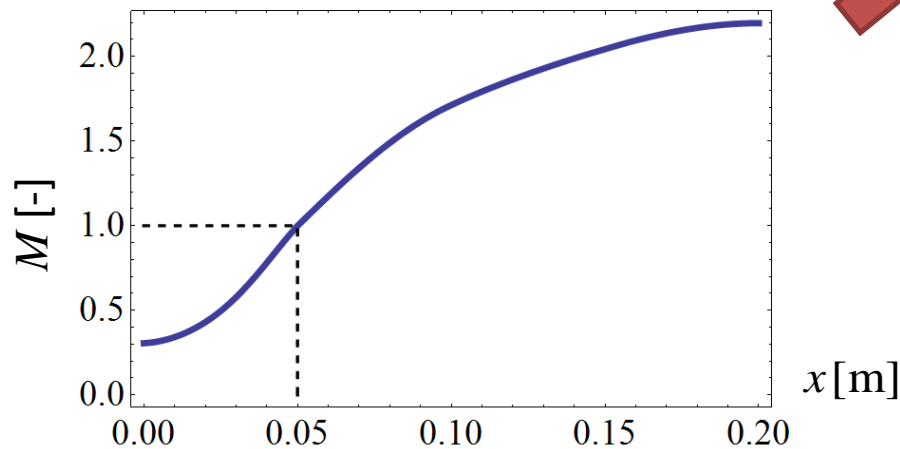
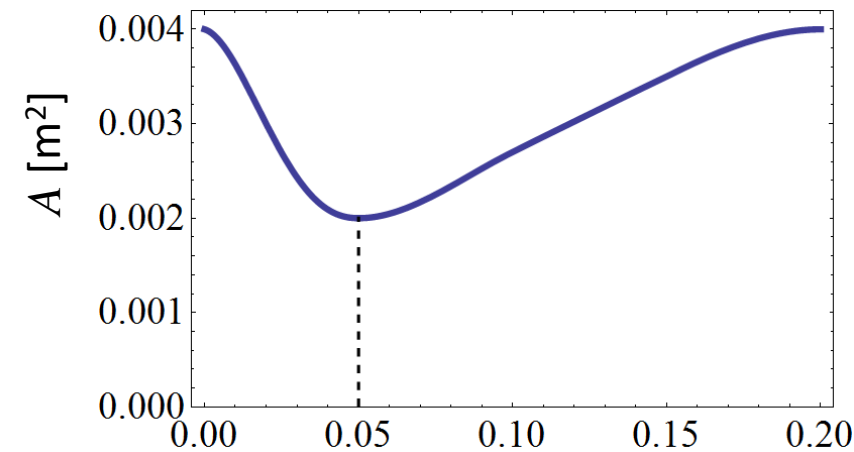
x [m]

$$\left(\frac{A}{A^*}\right)^2 = \frac{1}{M^2} \left[\frac{2}{\gamma+1} \left(1 + \frac{\gamma-1}{2} M^2 \right) \right]^{\frac{\gamma+1}{\gamma-1}}$$

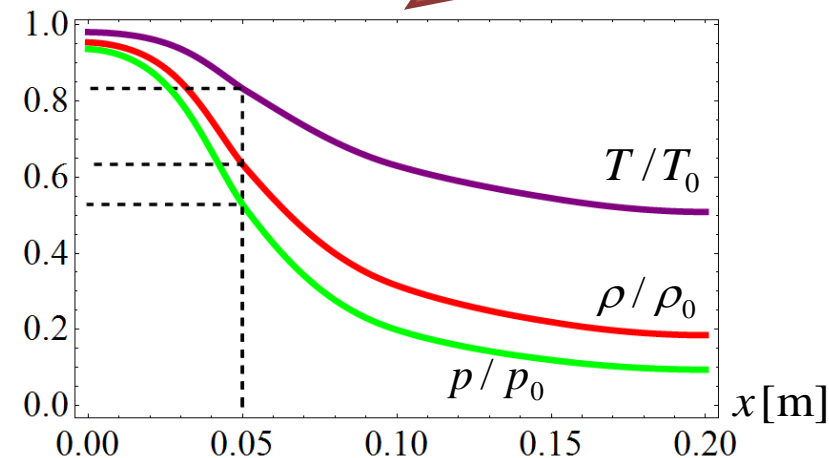
1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- variation of parameters in nozzle



$$1 + \frac{(\gamma - 1)}{2} M^2 = \frac{T_0}{T}$$
$$\left(1 + \frac{(\gamma - 1)}{2} M^2 \right)^{\frac{1}{\gamma - 1}} = \frac{\rho_0}{\rho}$$
$$\left(1 + \frac{(\gamma - 1)}{2} M^2 \right)^{\frac{\gamma}{\gamma - 1}} = \frac{p_0}{p}$$

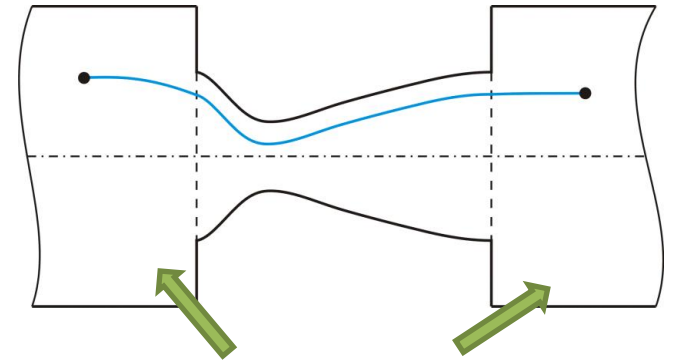


1. Fluid Flow and Thermodynamics

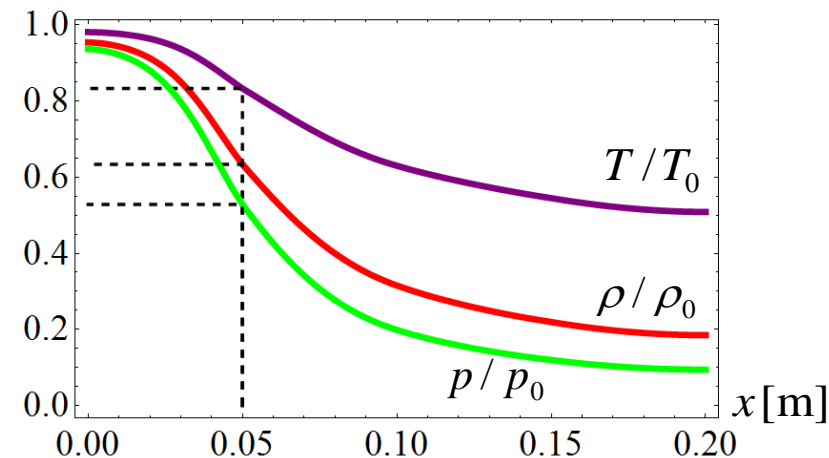
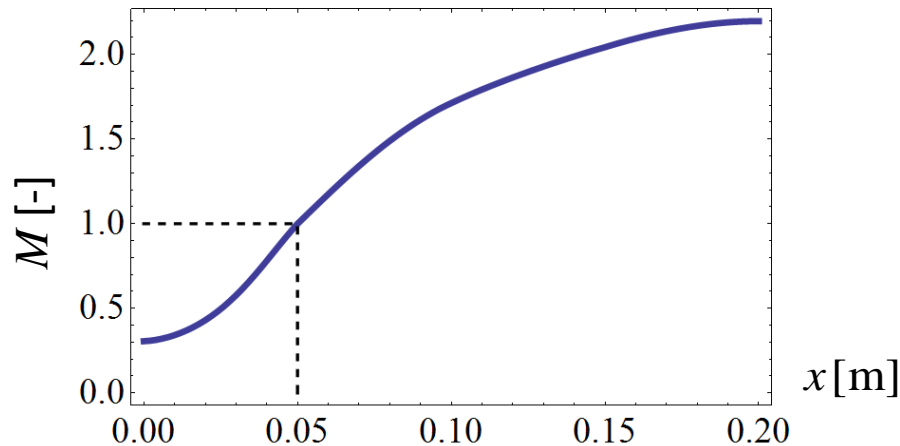
Nozzle fluid flow

- variation of parameters in nozzle

- no pressure difference – no fluid flow
- if pressure ratio p_e/p_0 is different from isentropic value, the flow will be different (inside or outside the nozzle)
- exit pressure for isentropic flow with supersonic speed is p_e



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

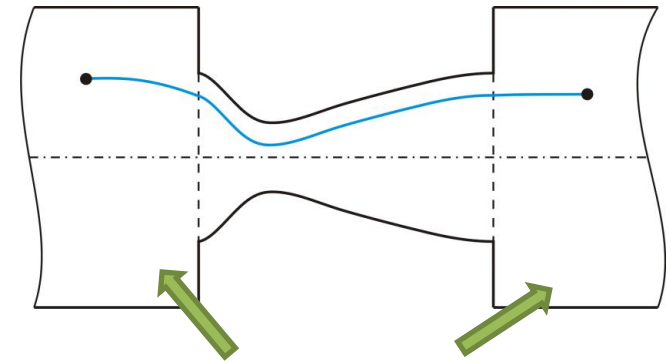
Nozzle fluid flow

- variation of parameters in nozzle

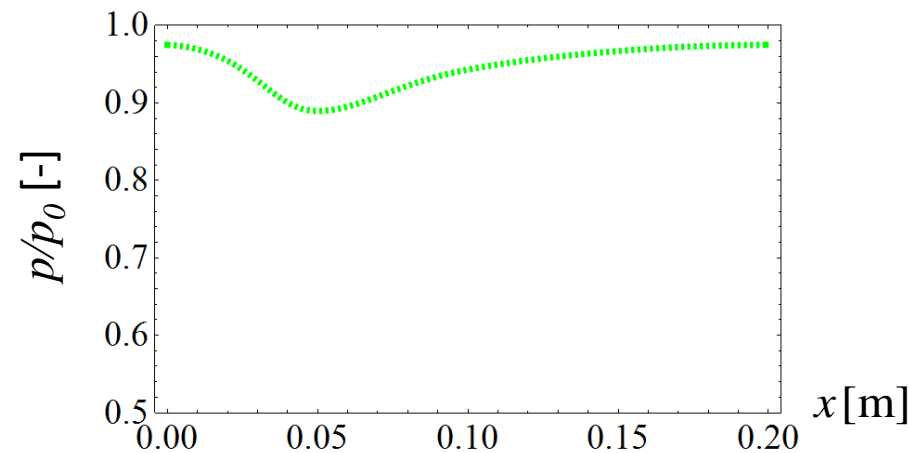
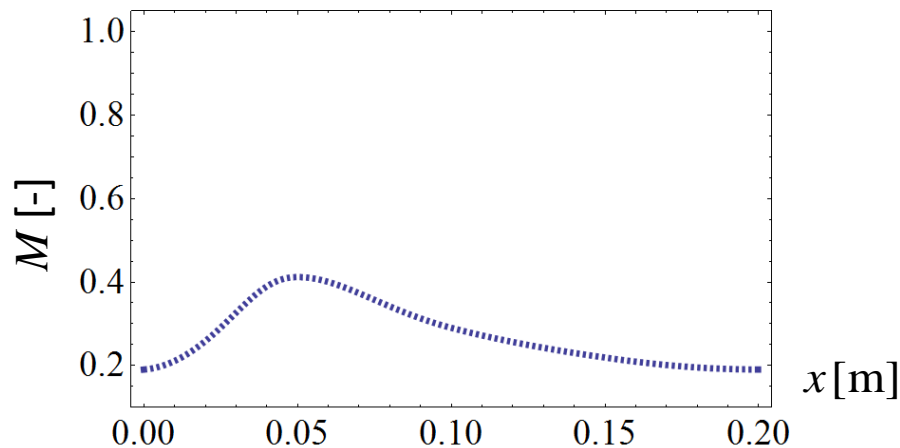
- $p_{B,1}$ is reduced below p_0



very low-speed subsonic flow, $p_{B,1} = p_{e,1}$



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

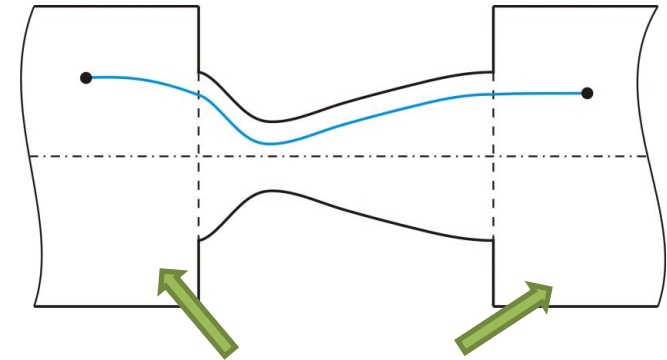
Nozzle fluid flow

- variation of parameters in nozzle

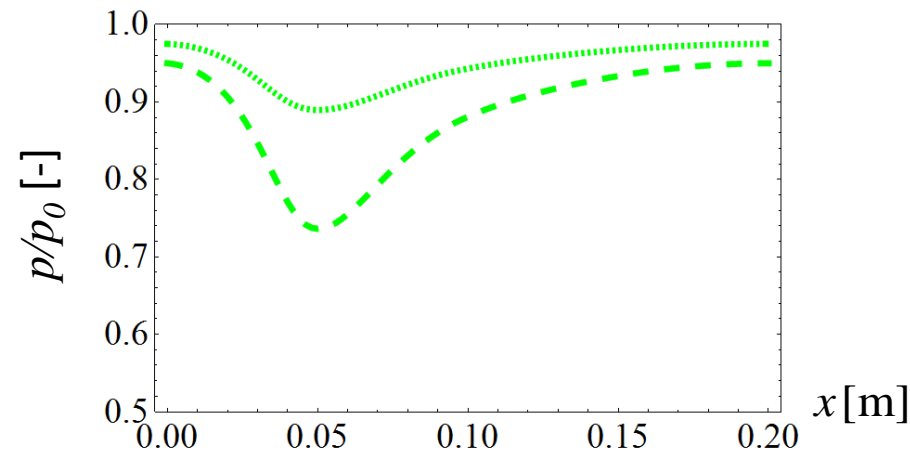
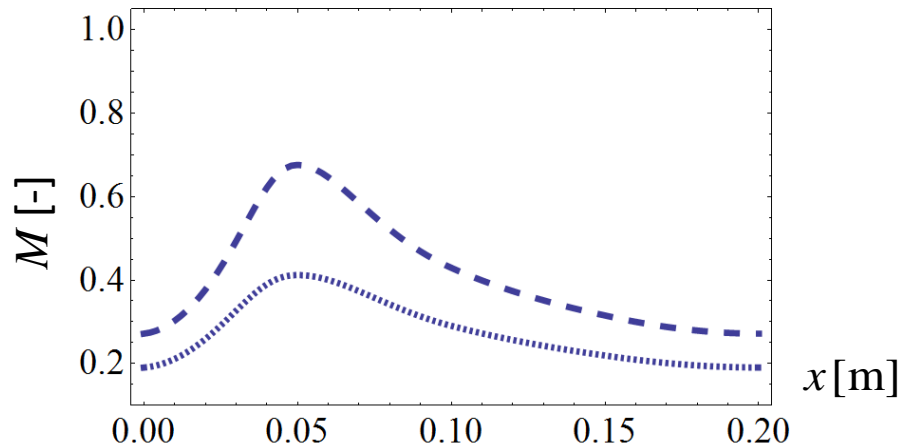
- $p_{B,2}$ is reduced below $p_{B,1}$



flow moves faster through nozzle, still subsonic flow, mass flow increases, $p_{B,2} = p_{e,2}$



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

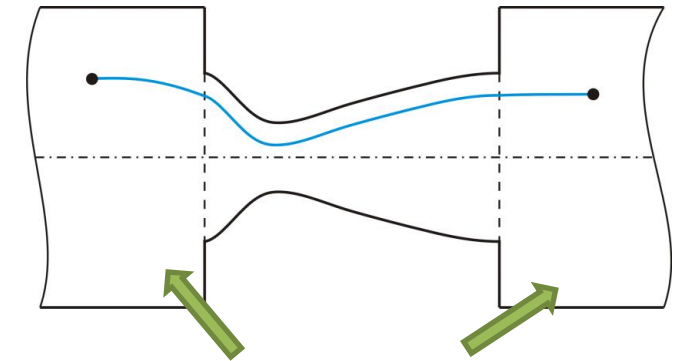
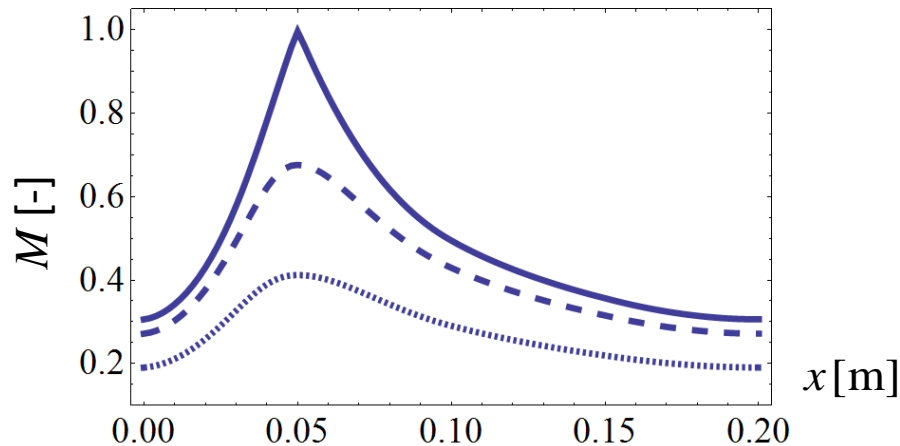
Nozzle fluid flow

- variation of parameters in nozzle

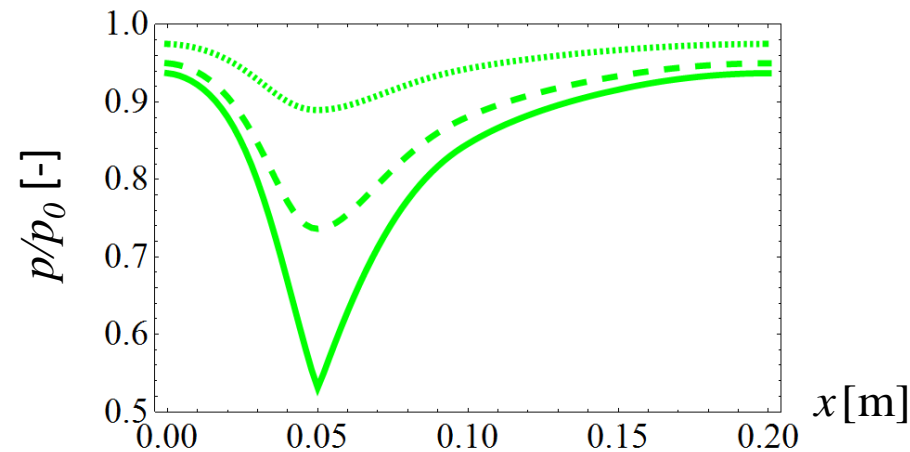
- $p_{B,3}$ is such, that it produces sonic flow in throat



only in throat is flow sonic, in other parts of nozzle is flow subsonic, mass flow increases and reaches max. value, $p_{B,2} = p_{e,2}$



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

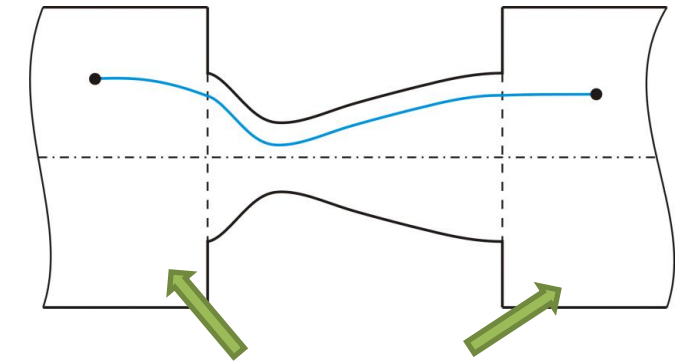
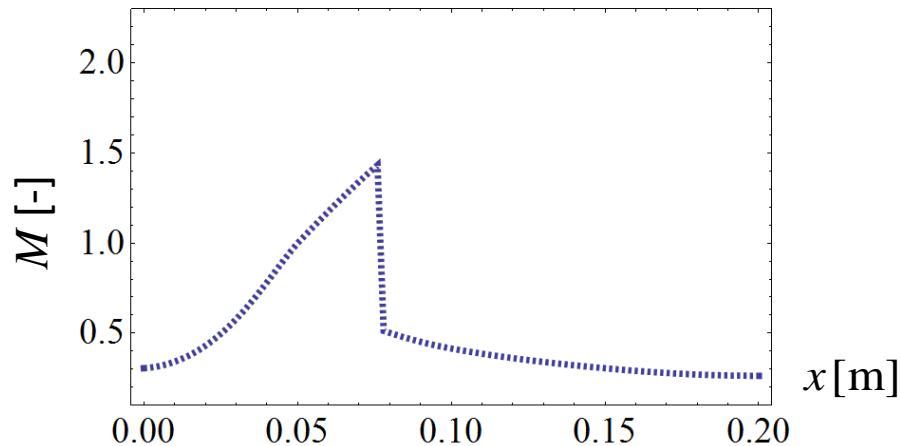
Nozzle fluid flow

- variation of parameters in nozzle

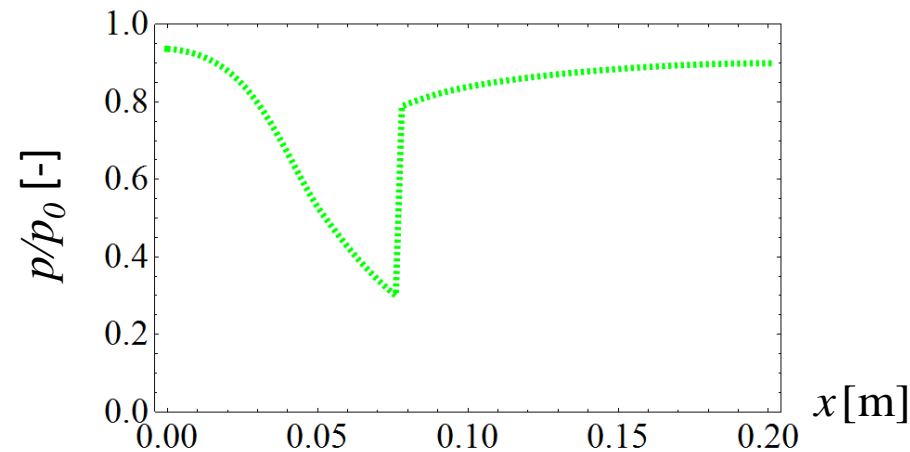
- $p_{B,4}$ is reduced below $p_{B,3}$



in divergent nozzle flow is at first supersonic, then shock wave is formed and flow is subsonic, mass flow is constant – choked flow, $p_{B,4} = p_{e,4}$



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

Nozzle fluid flow

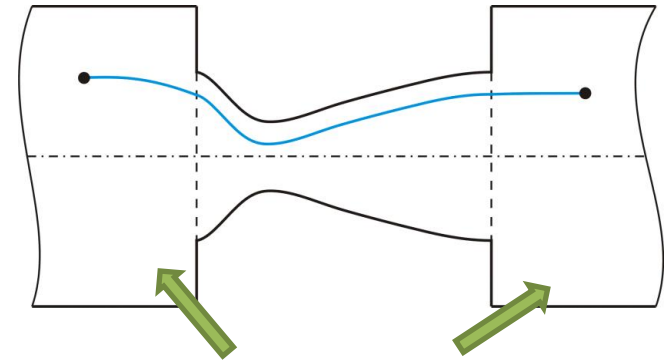
- variation of parameters in nozzle

- $p_{B,5}$ is reduced below $p_{B,4}$

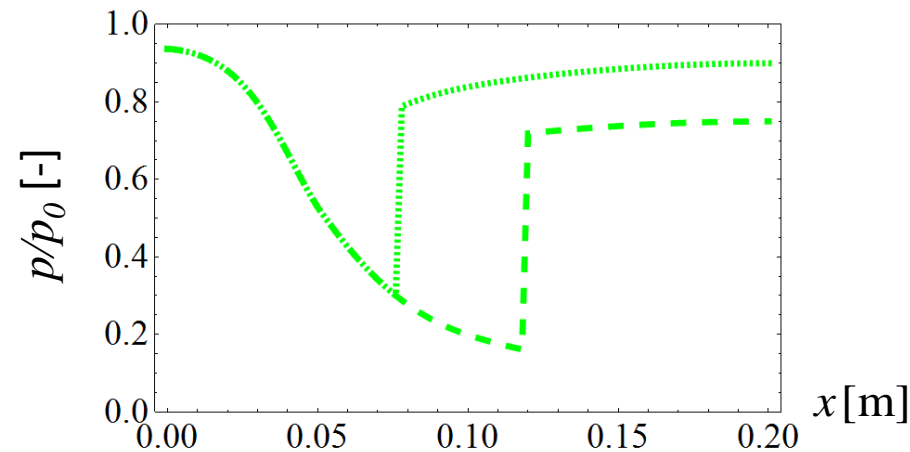
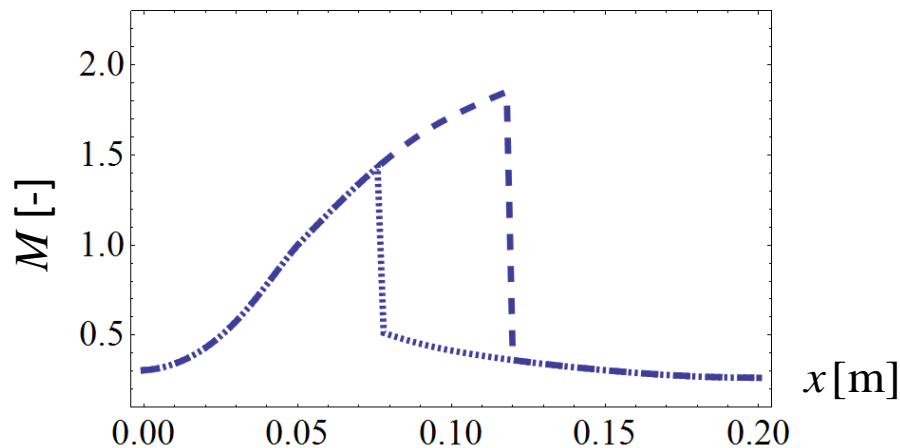


shock wave is moving toward the exit plane,

$$p_{B,4} = p_{e,4}$$



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

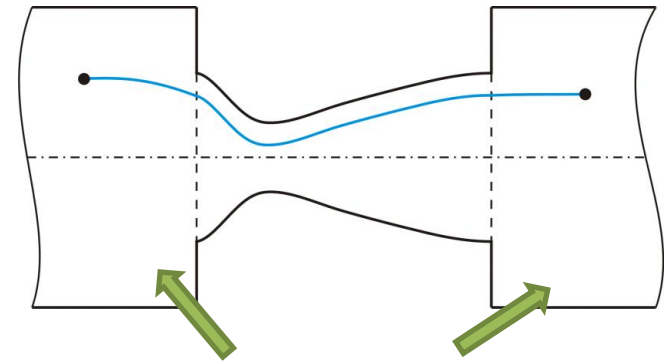
Nozzle fluid flow

- variation of parameters in nozzle

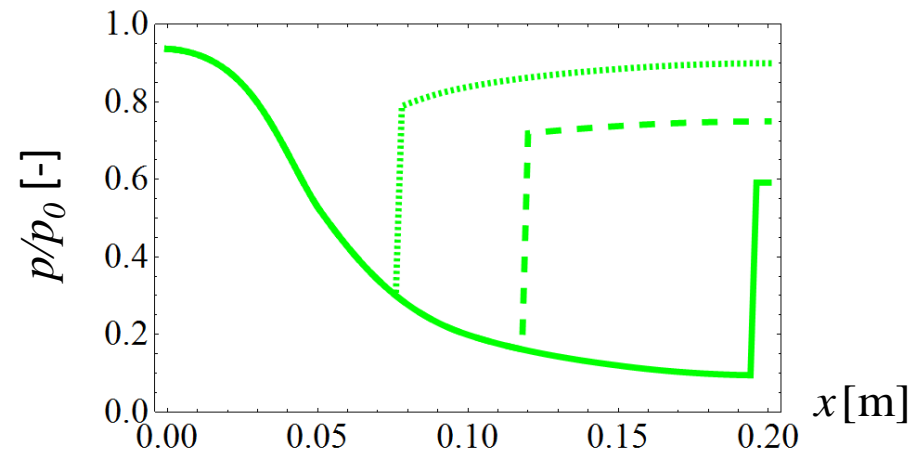
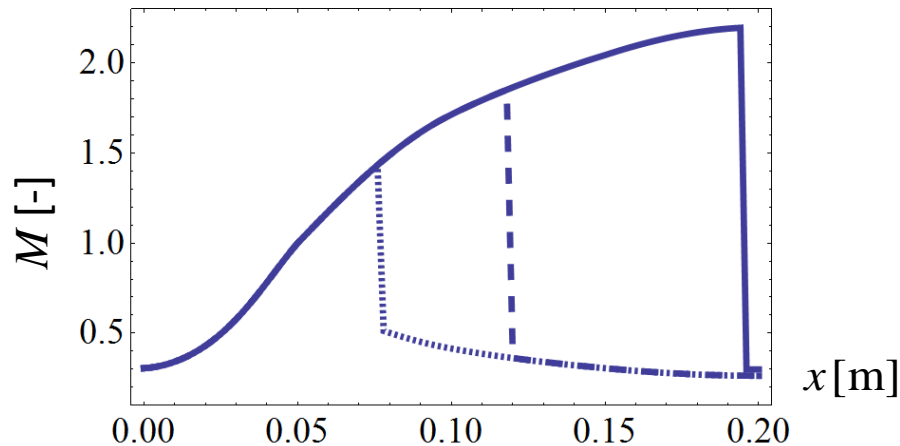
- $p_{B,6}$ is such, that shock wave is on the exit plane



the flow is supersonic in whole nozzle except the exit plane, $p_{B,6} = p_{e,6}$



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- variation of parameters in nozzle

- $p_{B,6}$ is such, that shock wave is on the exit plane

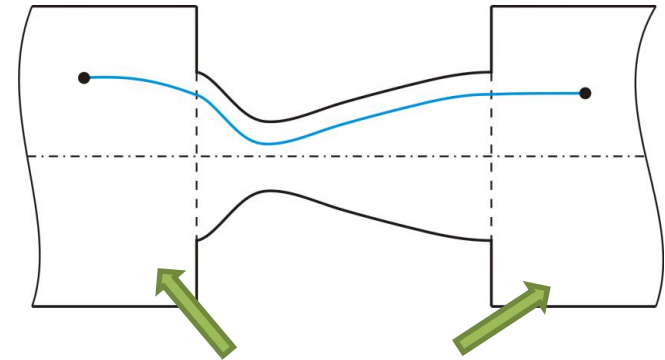


the flow is supersonic in whole nozzle except the exit plane, $p_{B,6} = p_{e,6}$

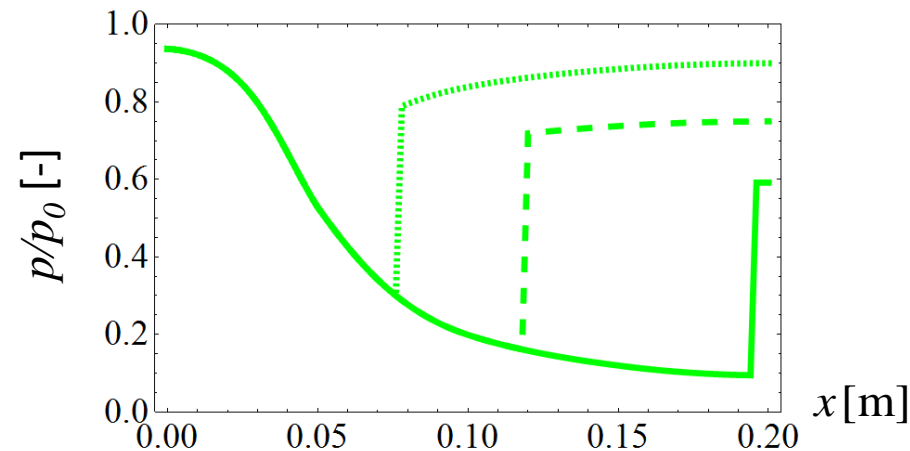


other reduction of back pressure p_B :

- exit pressure is constant p_e
- if $p_B > p_e$ shock waves moves outside nozzle
- if $p_B = p_e$ no shock waves are produced
- if $p_B < p_e$ expansion waves are formed outside the nozzle



pressure difference causes fluid flow



1. Fluid Flow and Thermodynamics

Nozzle fluid flow

- variation of parameters in nozzle

over-expanded flow $p_B > p_e$
(low altitudes)

- $p_{B,6}$ is such, that shock wave is on the exit plane



the flow is supersonic in whole nozzle except the exit plane, $p_{B,6} = p_{e,6}$



other reduction of back pressure p_B :

- exit pressure is constant p_e
- if $p_B > p_e$ shock waves moves outside nozzle
- if $p_B = p_e$ no shock waves are produced
- if $p_B < p_e$ expansion waves are formed outside the nozzle



under-expanded flow $p_B < p_e$
(high altitudes)

source: NASA

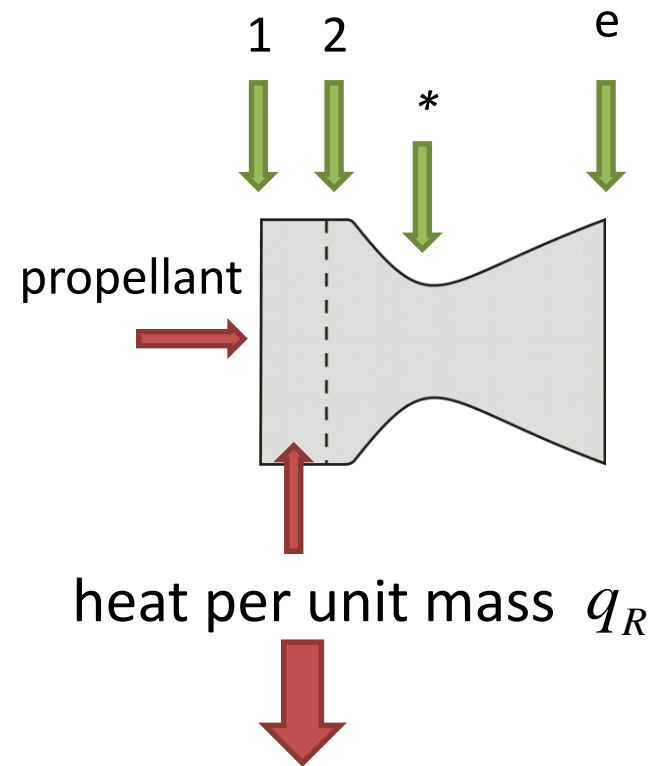
2. Chemical Rocket Propulsion

- Performance Characteristics
- Liquid Propellant Performance
- Feed System

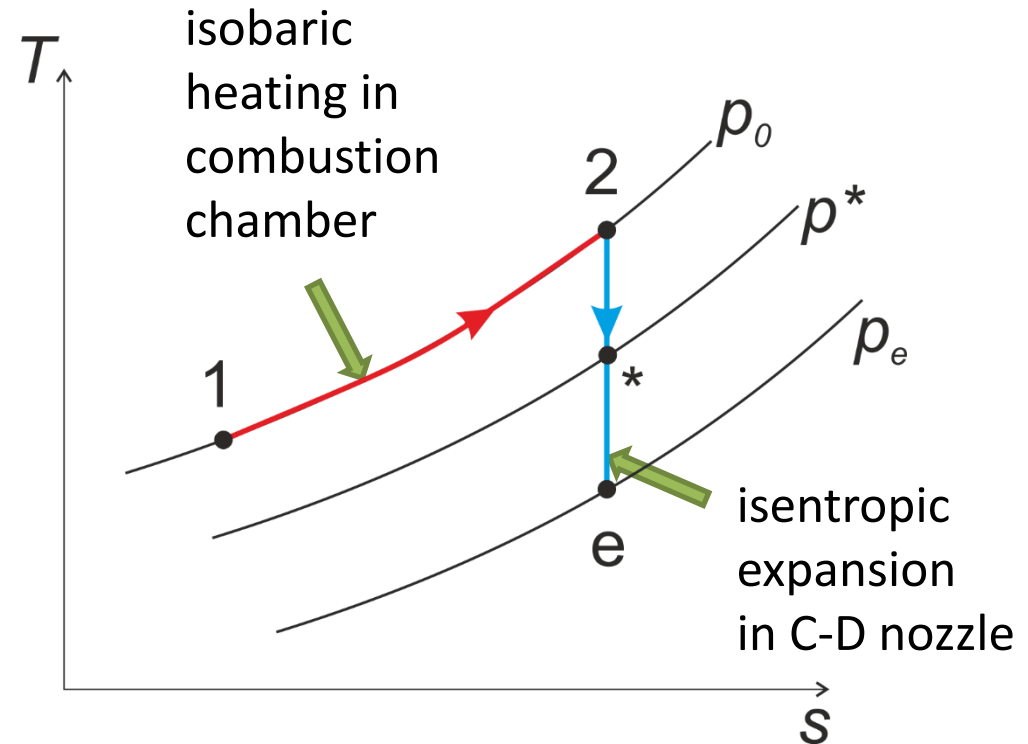
2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – combustion



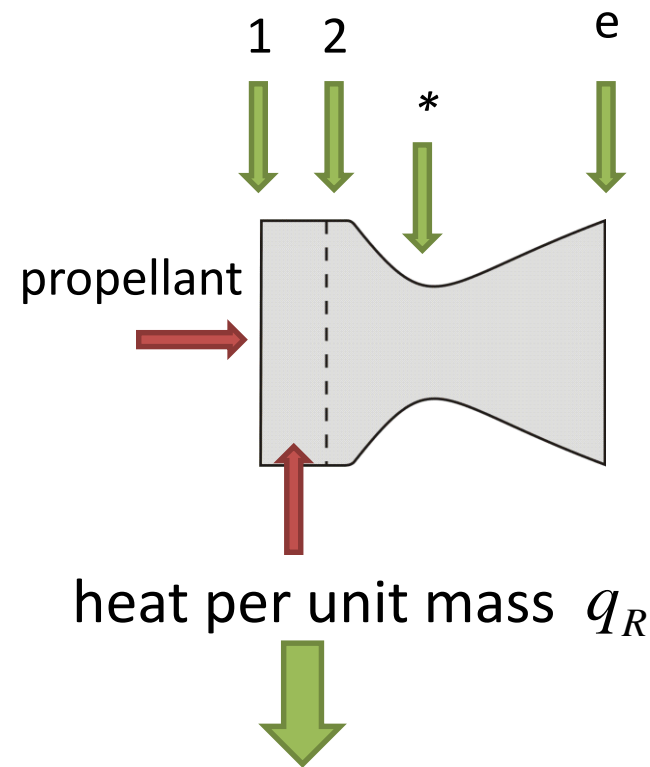
First thermodynamic law – isobaric heating:
 $q_R = \Delta h = c_p \Delta T$



2. Chemical Rocket Propulsion

Performance Characteristics

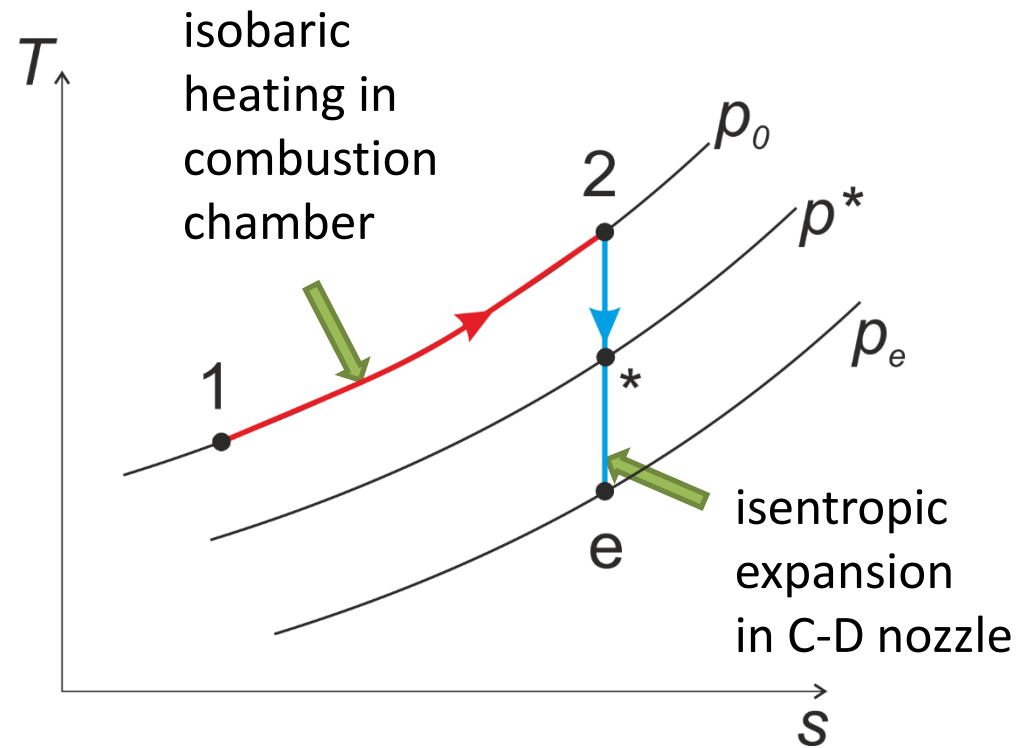
- Rocket thrust chamber – combustion



First thermodynamic law – isobaric heating:

$$q_R = \Delta h = c_p \Delta T \rightarrow$$

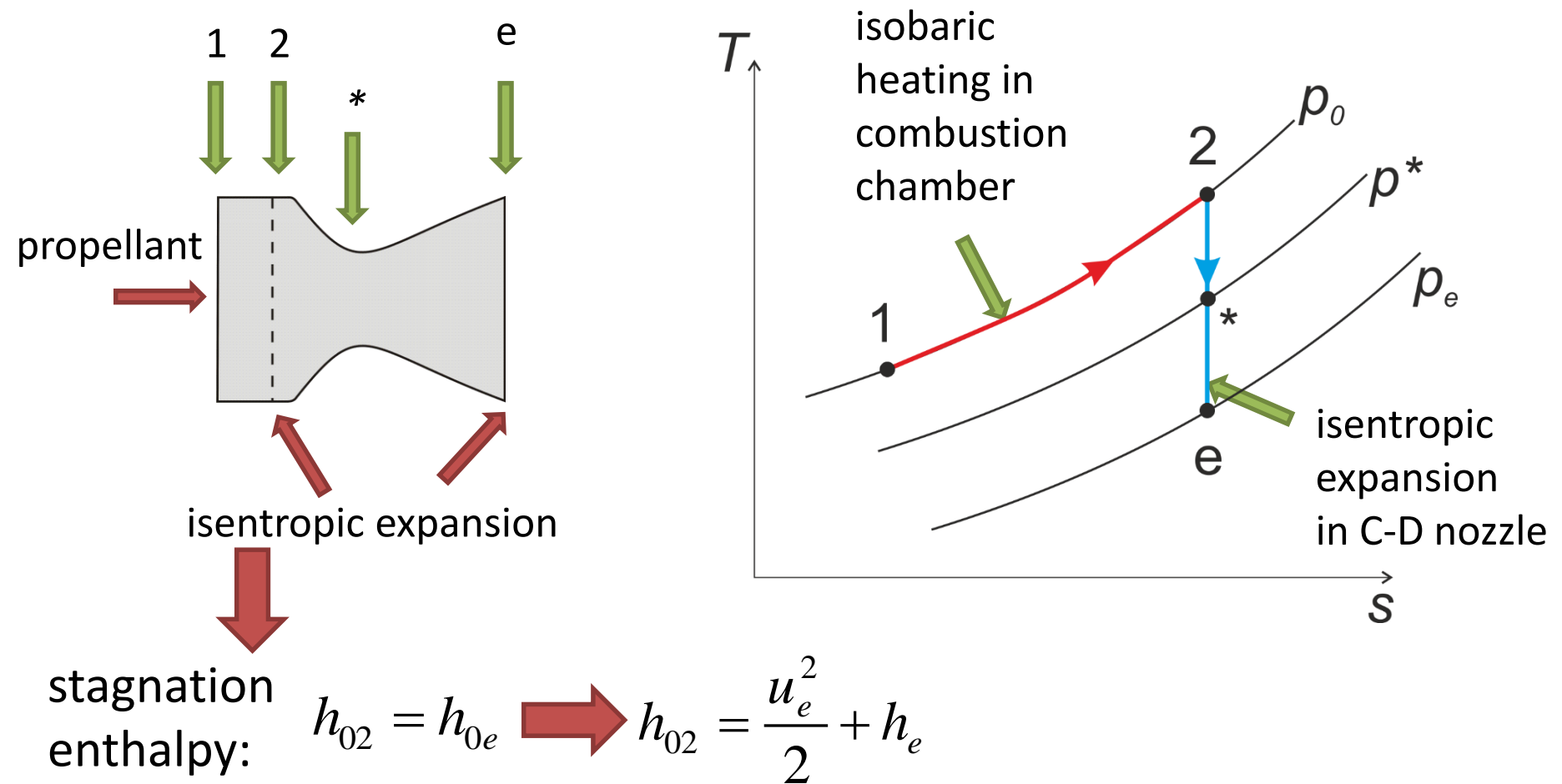
$$T_{02} = T_{01} + \frac{q_R}{c_p}$$



2. Chemical Rocket Propulsion

Performance Characteristics

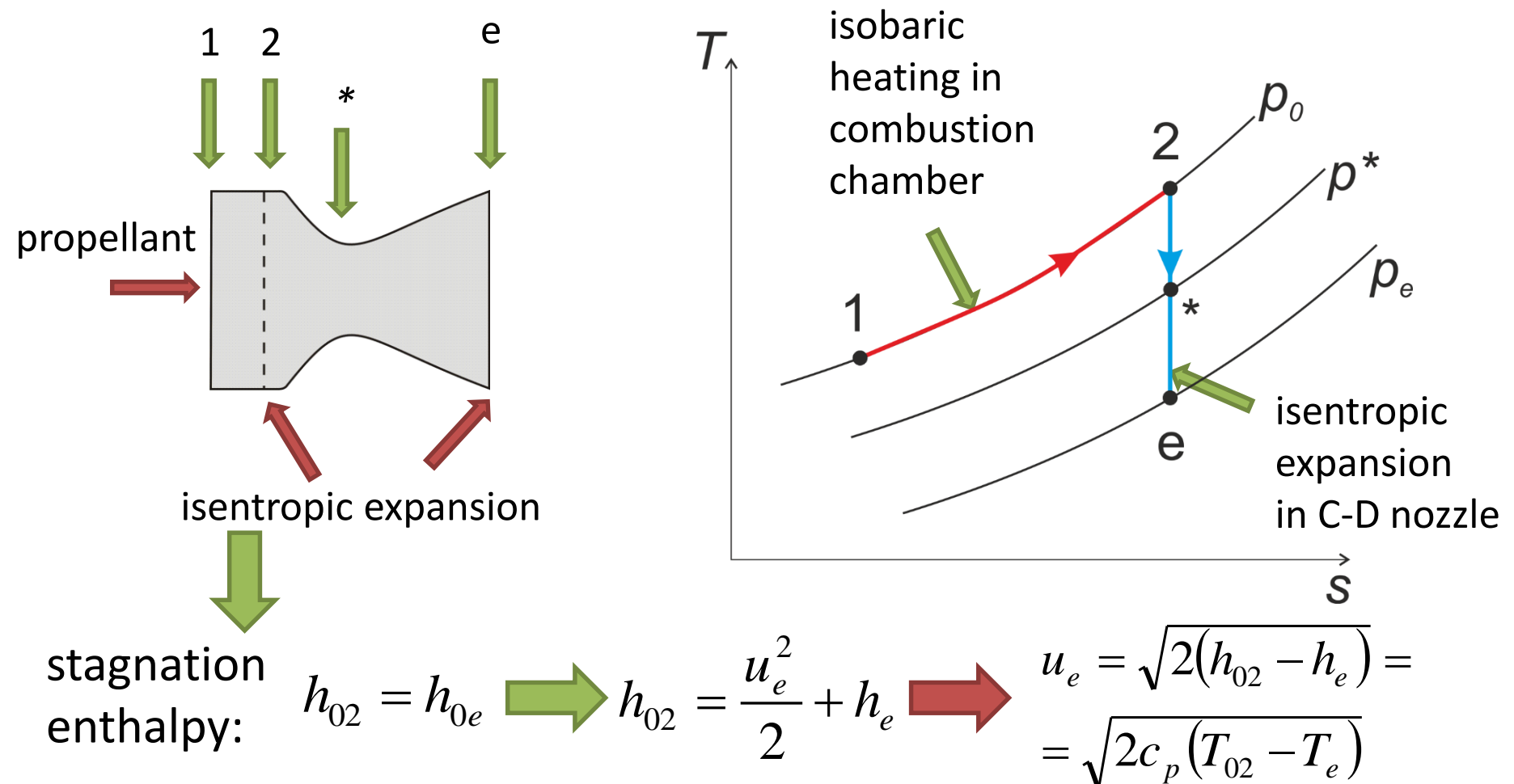
- Rocket thrust chamber – expansion



2. Chemical Rocket Propulsion

Performance Characteristics

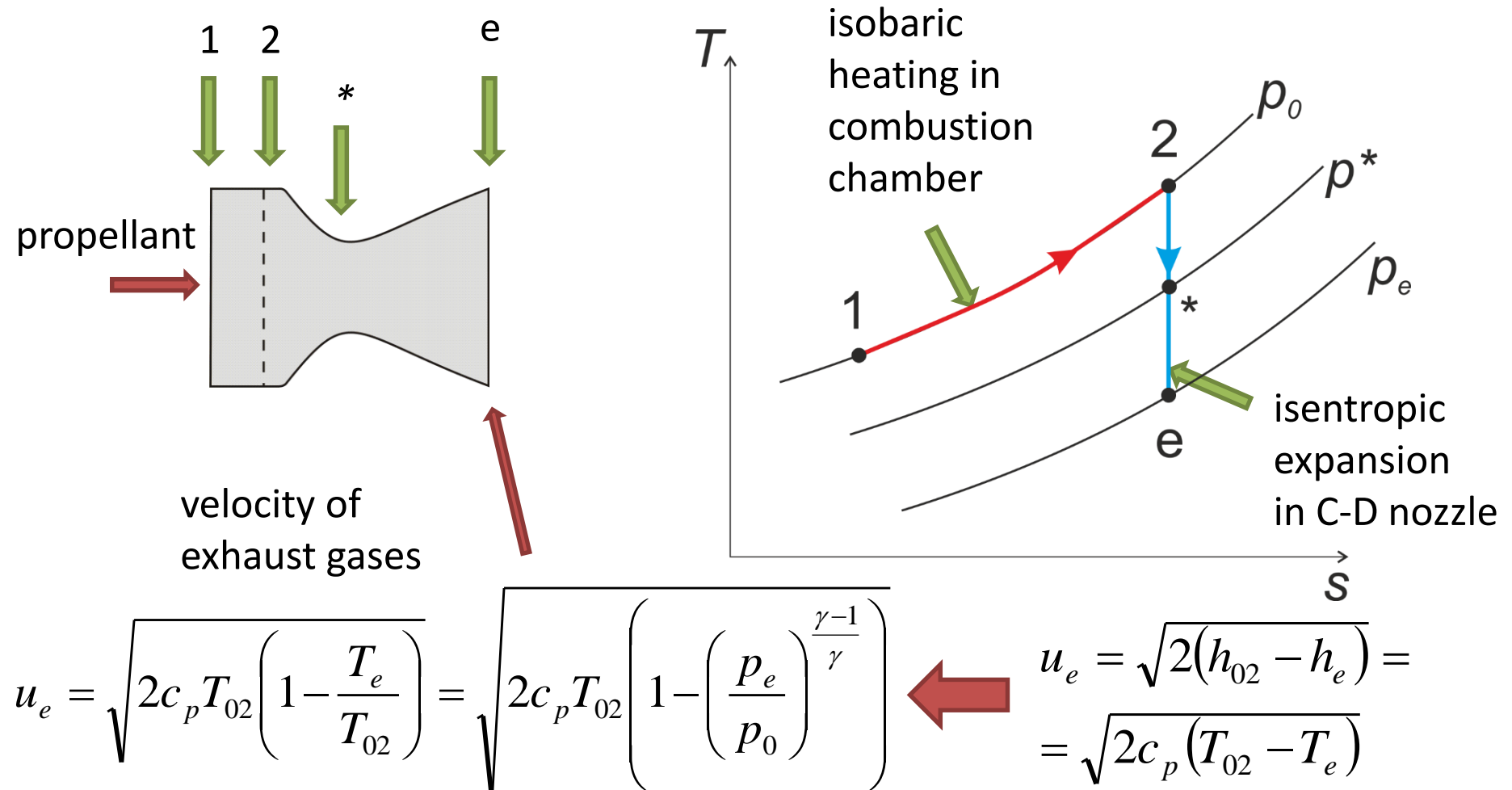
- Rocket thrust chamber – expansion



2. Chemical Rocket Propulsion

Performance Characteristics

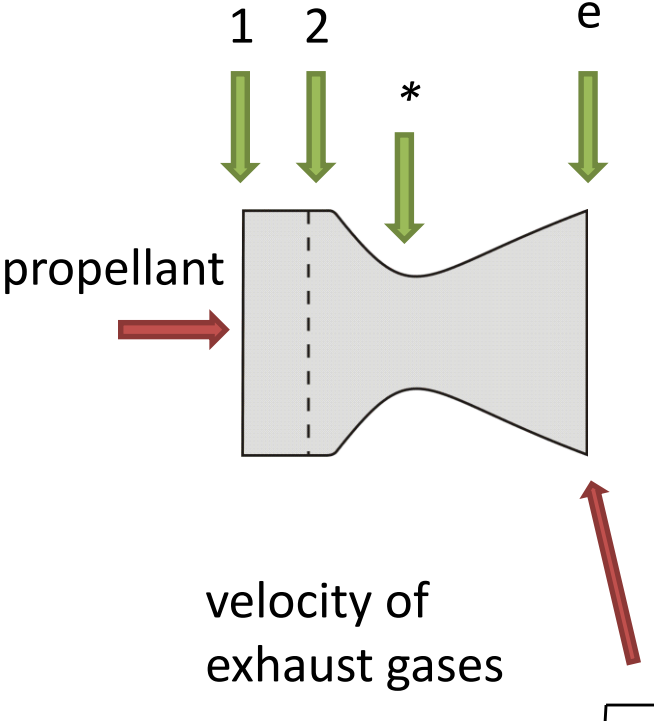
- Rocket thrust chamber – expansion



2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – exit velocity



propellant

1 2 * e

velocity of exhaust gases

$$u_e = \sqrt{2 \frac{\gamma}{\gamma-1} R T_{02} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)}$$

$$c_p = \frac{\gamma R}{\gamma-1}$$

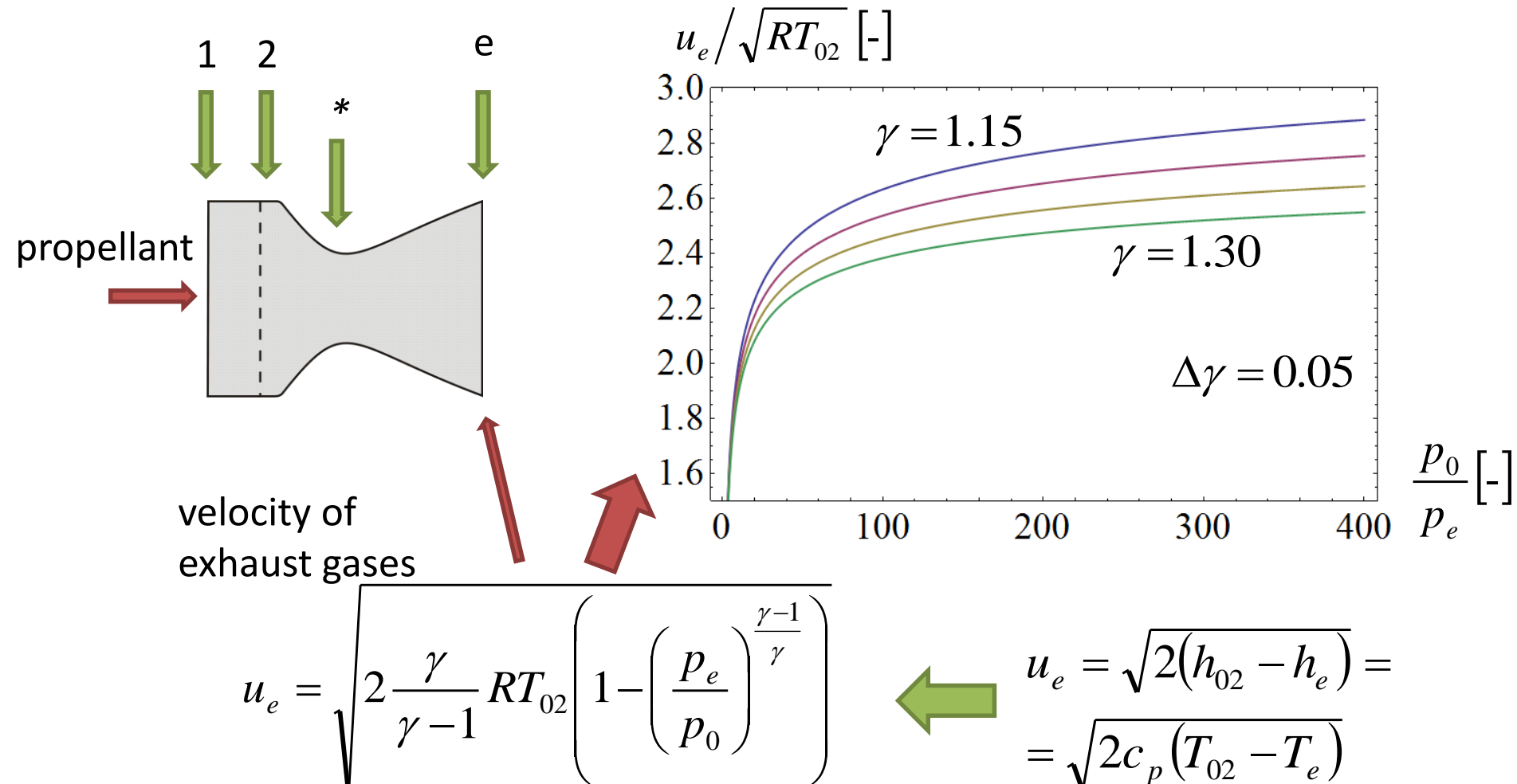
$$u_e = \sqrt{2(h_{02} - h_e)} = \sqrt{2c_p(T_{02} - T_e)}$$

$$u_e = \sqrt{2c_p T_{02} \left(1 - \frac{T_e}{T_{02}} \right)} = \sqrt{2c_p T_{02} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)}$$

2. Chemical Rocket Propulsion

Performance Characteristics

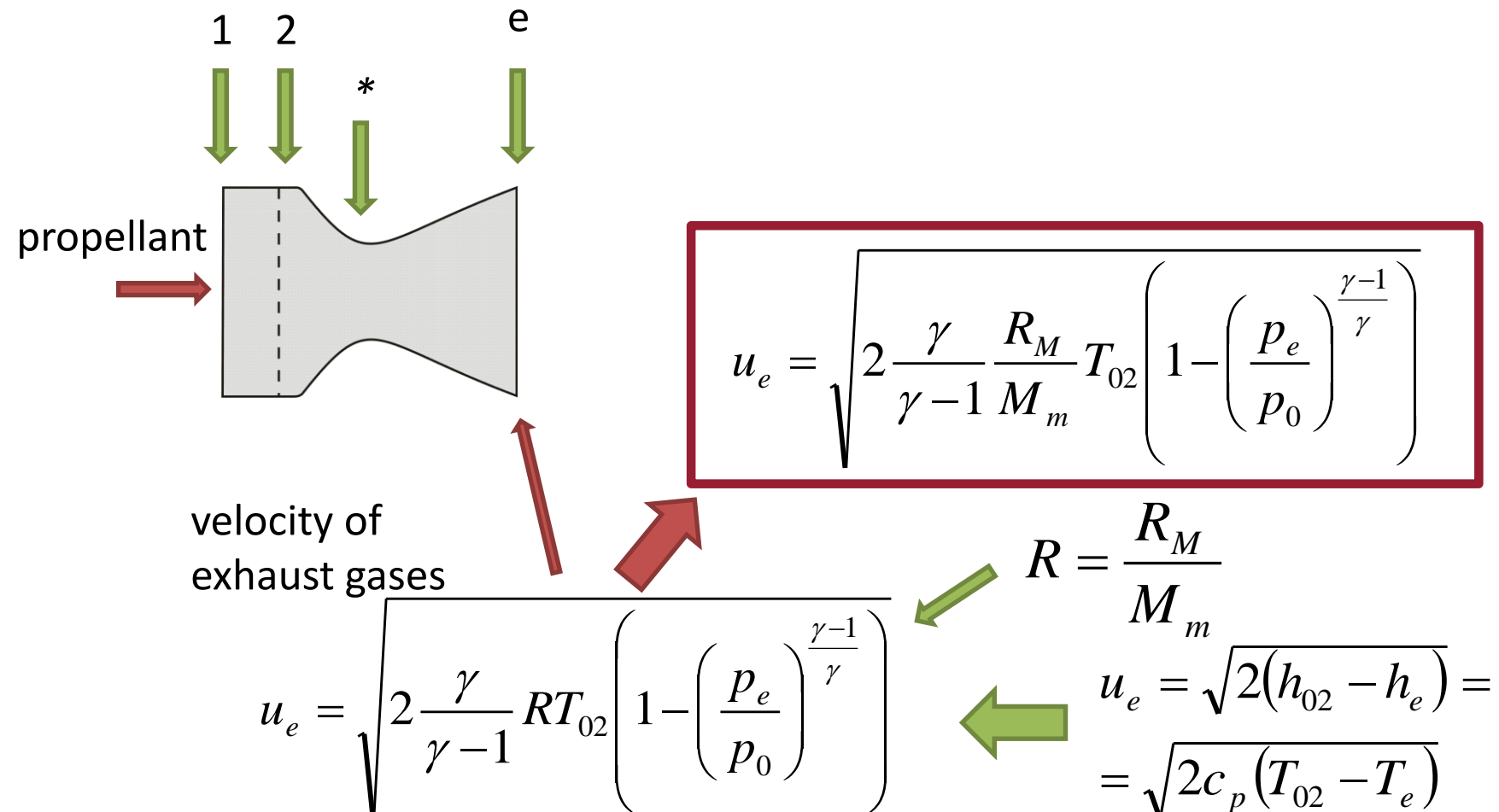
- Rocket thrust chamber – exit velocity



2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – exit velocity



2. Chemical Rocket Propulsion

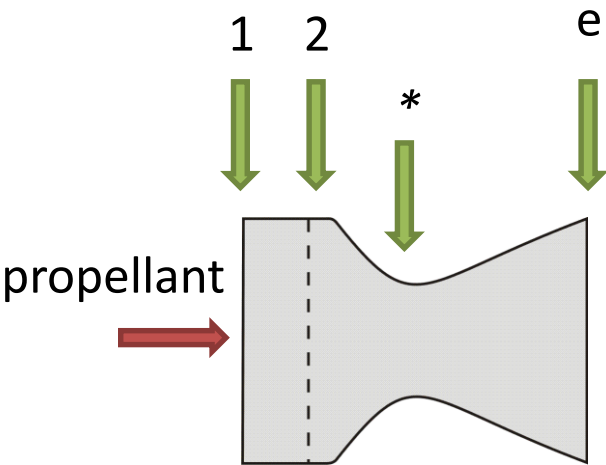
Performance Characteristics

- Rocket thrust chamber – exit velocity

higher stag. temperature → higher exit velocity

low molecular weight → higher exit velocity

higher pressure ratio p_0/p_e → higher exit velocity



$$u_e = \sqrt{2 \frac{\gamma}{\gamma-1} \frac{R_M}{M_m} T_{02} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)}$$

velocity of
exhaust gases

$$u_e = \sqrt{2 \frac{\gamma}{\gamma-1} R T_{02} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)}$$

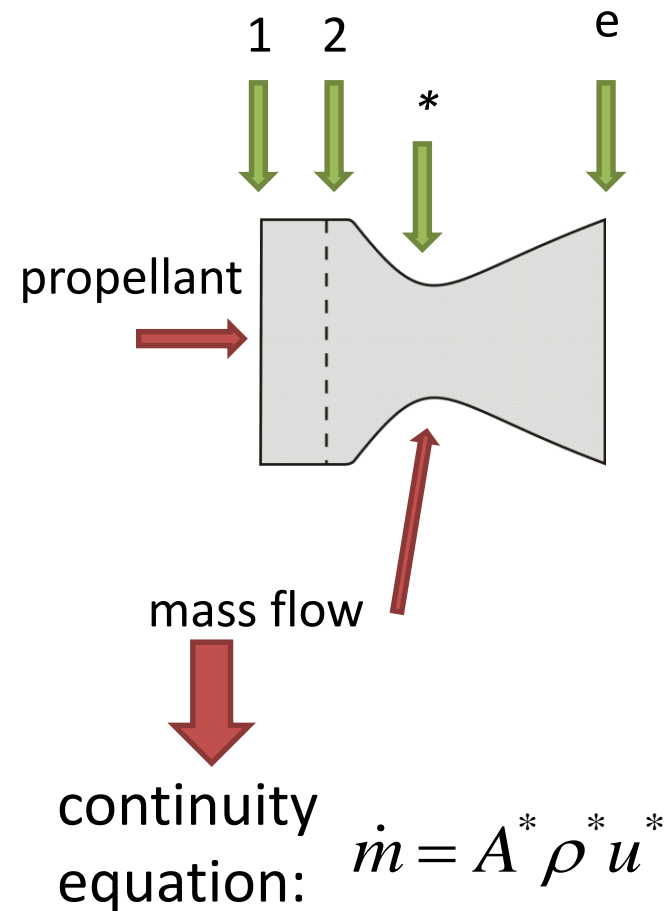
$$R = \frac{R_M}{M_m}$$

$$u_e = \sqrt{2(h_{02} - h_e)} = \sqrt{2c_p(T_{02} - T_e)}$$

2. Chemical Rocket Propulsion

Performance Characteristics

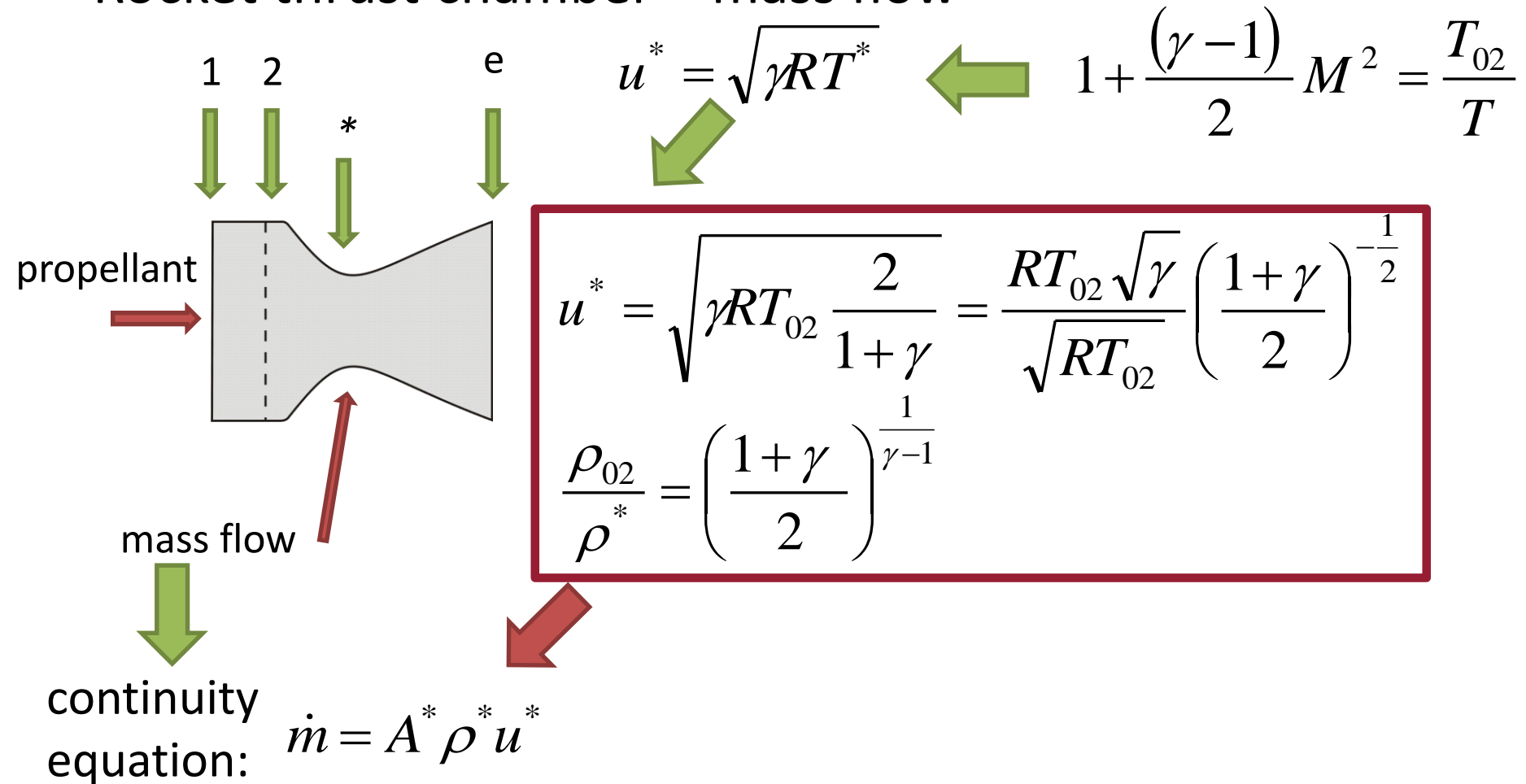
- Rocket thrust chamber – mass flow



2. Chemical Rocket Propulsion

Performance Characteristics

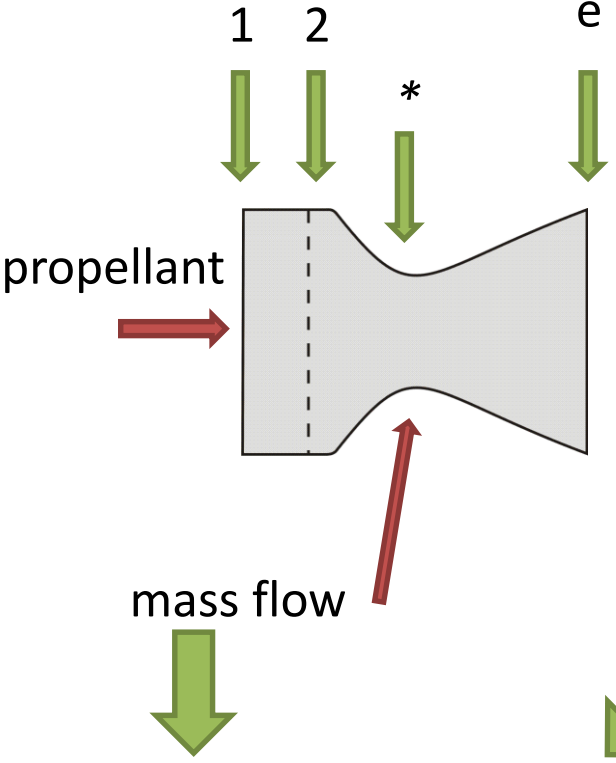
- Rocket thrust chamber – mass flow



2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – mass flow



The diagram illustrates the flow of propellant through a rocket thrust chamber. Propellant enters from the left, indicated by a red arrow labeled 'propellant'. The flow is divided into two sections by a dashed line, labeled 1 and 2. The flow then passes through a nozzle, indicated by a green arrow labeled 'e'. The mass flow is represented by a green arrow labeled 'mass flow' pointing downwards. The continuity equation is shown as $\dot{m} = A^* \rho^* u^*$. The nozzle exit area is labeled A^* . The mass flow rate is labeled \dot{m} . The density is labeled ρ . The velocity is labeled u^* . The stagnation conditions are labeled ρ_{02} and T_{02} . The specific heat ratio is labeled γ . The mass flow rate is also expressed as $\dot{m} = A^* \rho_{02} \left(\frac{1+\gamma}{2} \right)^{\frac{-1}{\gamma-1}} \frac{RT_{02} \sqrt{\gamma}}{\sqrt{RT_{02}}} \left(\frac{1+\gamma}{2} \right)^{-\frac{1}{2}}$.

$$u^* = \sqrt{\gamma RT_{02} \frac{2}{1+\gamma}} = \frac{RT_{02} \sqrt{\gamma}}{\sqrt{RT_{02}}} \left(\frac{1+\gamma}{2} \right)^{-\frac{1}{2}}$$

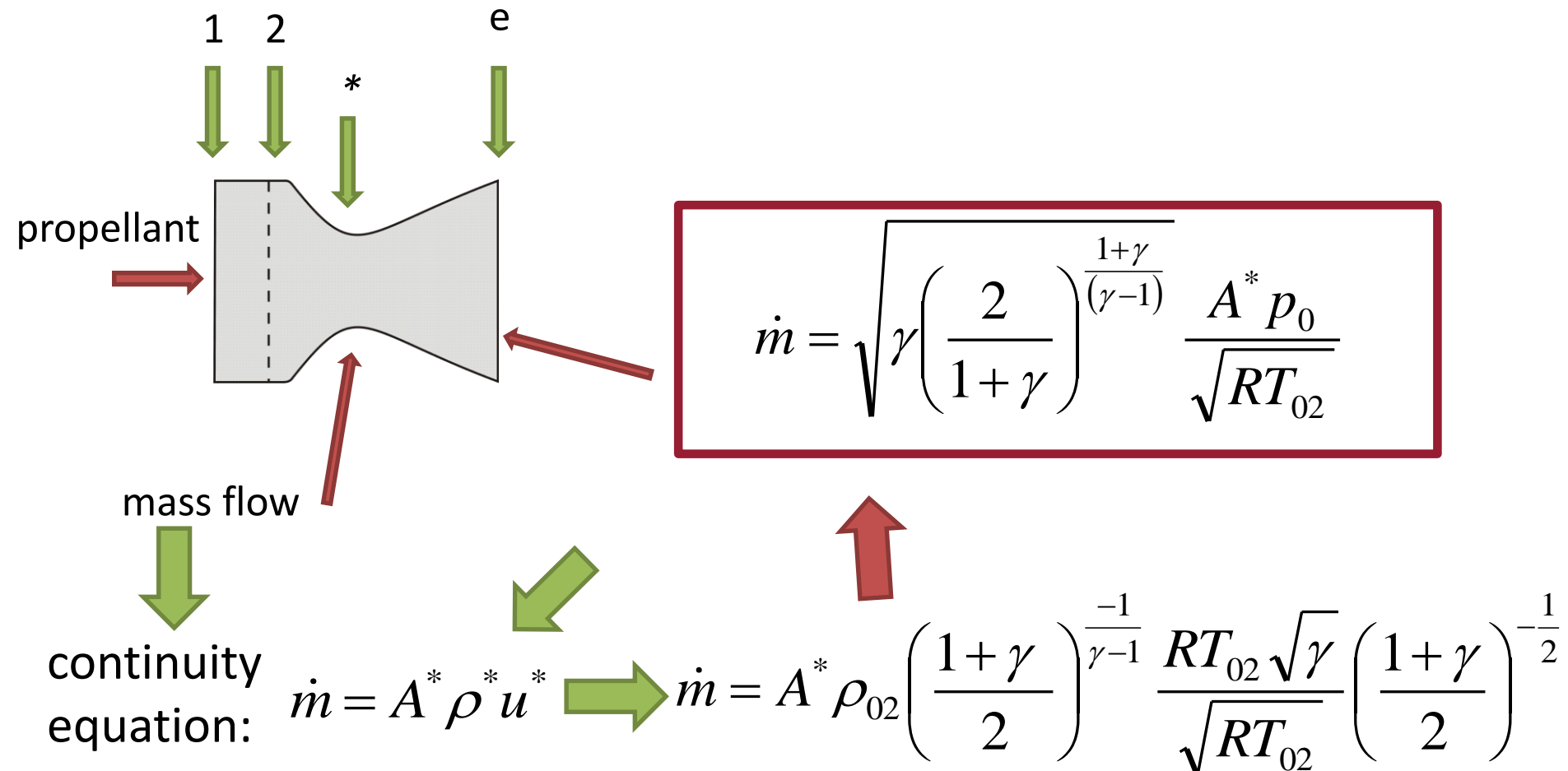
$$\frac{\rho_{02}}{\rho^*} = \left(\frac{1+\gamma}{2} \right)^{\frac{1}{\gamma-1}}$$

$$\dot{m} = A^* \rho^* u^* \Rightarrow \dot{m} = A^* \rho_{02} \left(\frac{1+\gamma}{2} \right)^{\frac{-1}{\gamma-1}} \frac{RT_{02} \sqrt{\gamma}}{\sqrt{RT_{02}}} \left(\frac{1+\gamma}{2} \right)^{-\frac{1}{2}}$$

2. Chemical Rocket Propulsion

Performance Characteristics

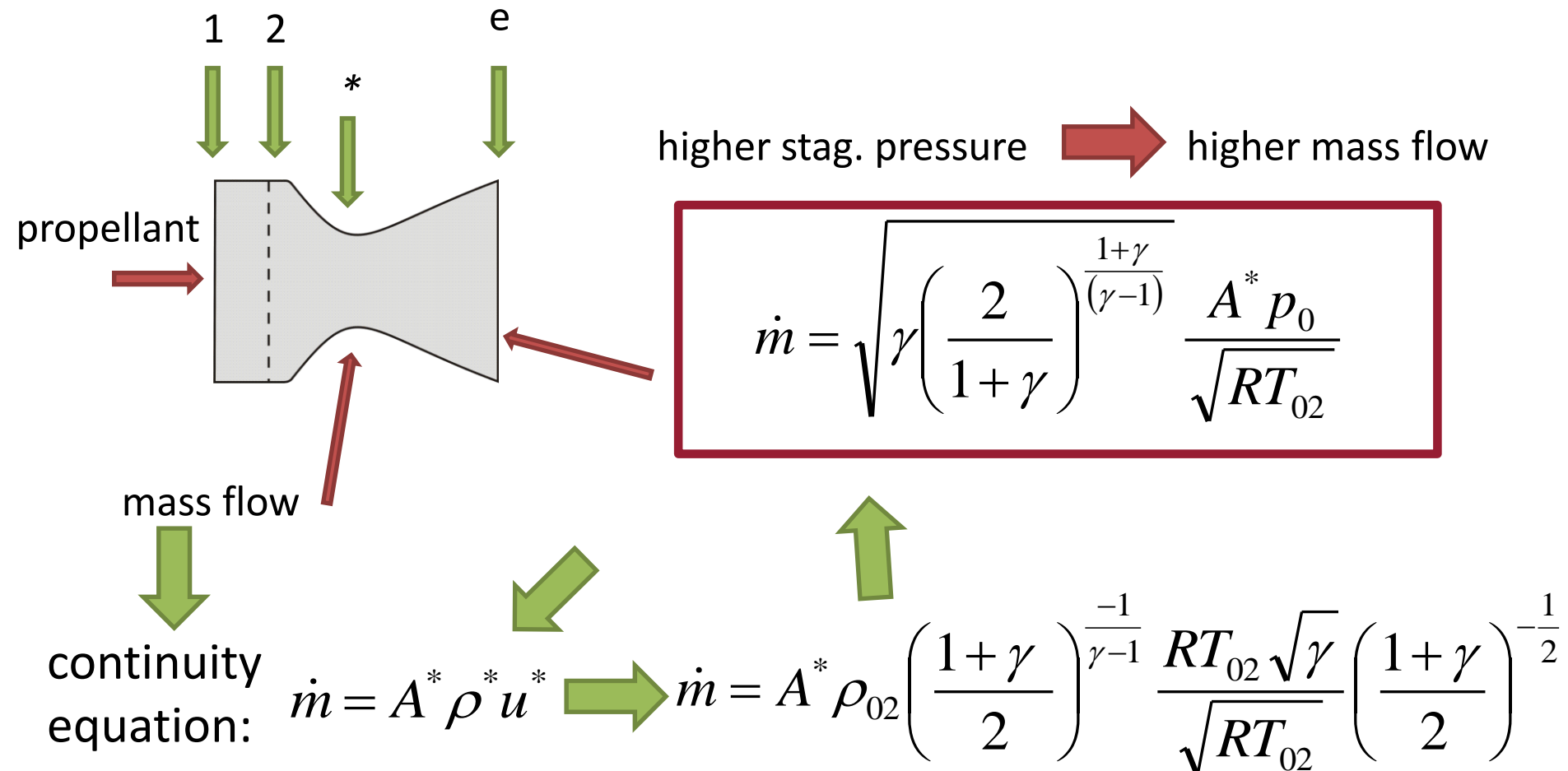
- Rocket thrust chamber – mass flow



2. Chemical Rocket Propulsion

Performance Characteristics

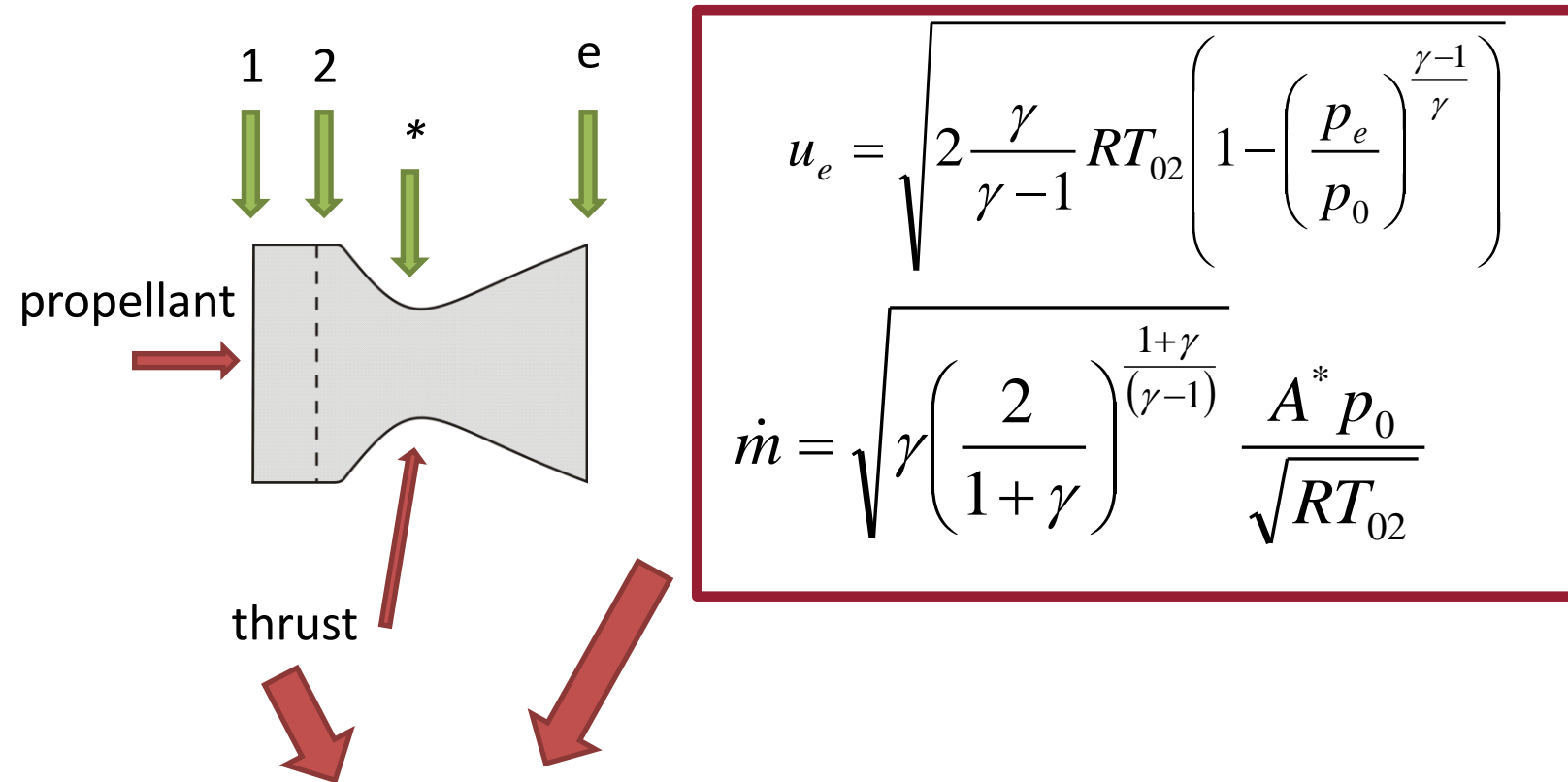
- Rocket thrust chamber – mass flow



2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – thrust

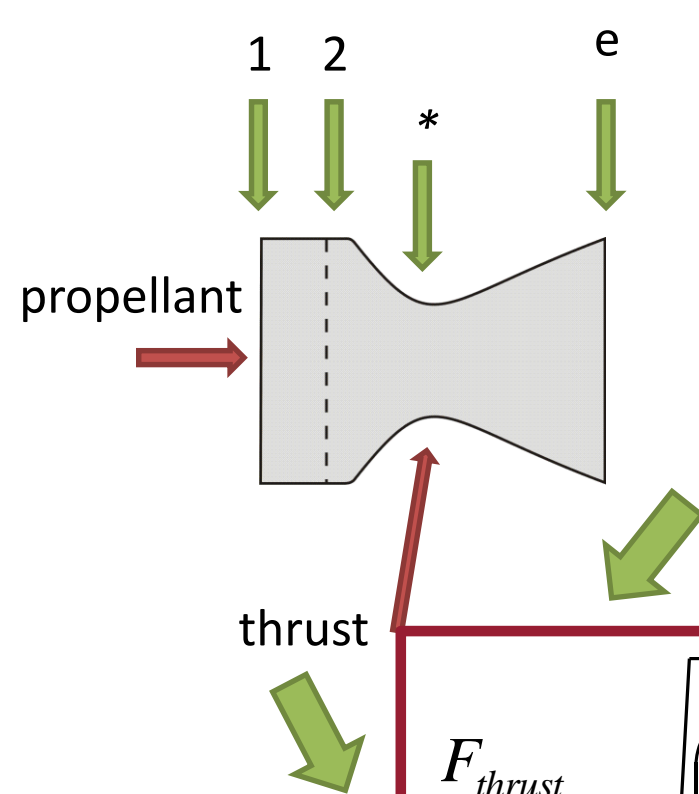


$$F_{thrust} = \dot{m} u_e + A_e (p_e - p_a)$$

2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – thrust



$$u_e = \sqrt{2 \frac{\gamma}{\gamma-1} R T_{02} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)}$$

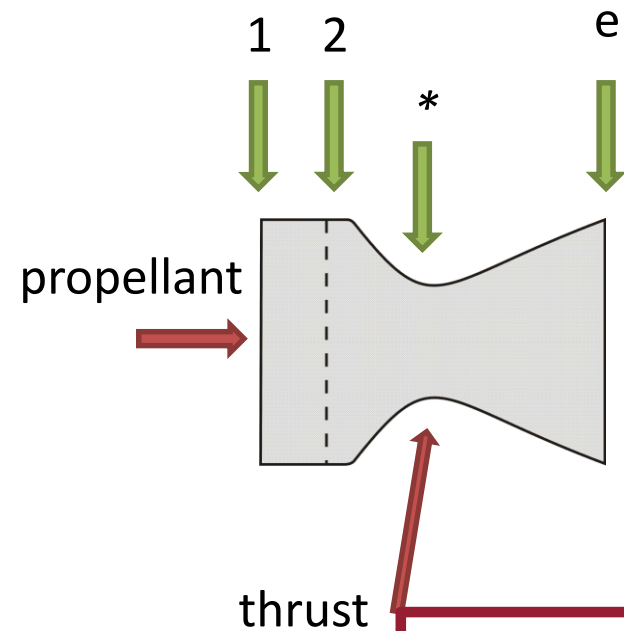
$$\dot{m} = \sqrt{\gamma \left(\frac{2}{1+\gamma} \right)^{\frac{1+\gamma}{(\gamma-1)}} \frac{A^* p_0}{\sqrt{R T_{02}}}}$$

$$\frac{F_{thrust}}{A^* p_0} = \sqrt{\left(\frac{2}{1+\gamma} \right)^{\frac{1+\gamma}{(\gamma-1)}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)} + \frac{A_e (p_e - p_a)}{A^* p_0}$$

2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – thrust



$$F_{thrust} = \dot{m} \frac{p_0 A^*}{\dot{m}} \frac{F_{thrust}}{p_0 A^*} =$$

$$= \dot{m} \underset{\substack{\text{characteristic} \\ \text{velocity}}}{c^*} \underset{\substack{\text{thrust} \\ \text{coefficient}}}{C_F}$$

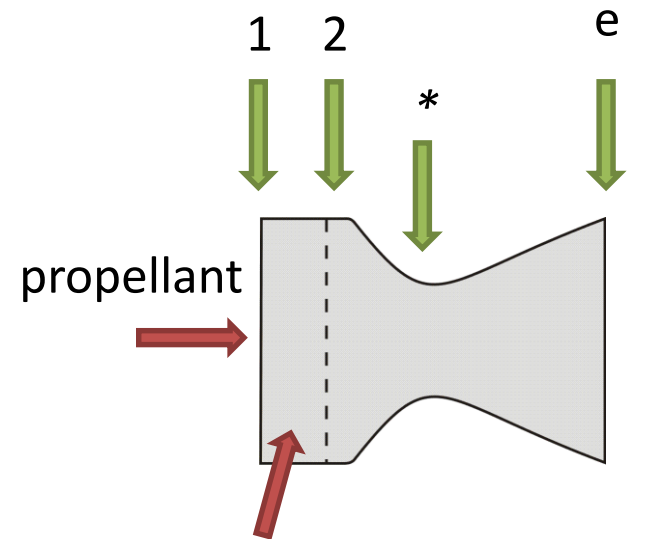
thrust depends only on stagnation pressure in combustion chamber

$$\frac{F_{thrust}}{A^* p_0} = \sqrt{\left(\frac{2}{1+\gamma} \right)^{\frac{1+\gamma}{\gamma-1}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \right)} + \frac{A_e (p_e - p_a)}{A^* p_0}$$

2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – characteristic velocity



characteristic velocity – c^*
specify combustion chamber

$$c^* = \frac{p_0 A^*}{\dot{m}}$$

characteristic velocity is function of
combustion chamber design and
propellant characteristics

$$c^* = \sqrt{\frac{1}{\gamma} \left(\frac{\gamma + 1}{2} \right)^{\frac{\gamma + 1}{\gamma - 1}} \frac{R_M T_{02}}{M_m}}$$

2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – thrust coefficient

thrust coefficient $C_F \rightarrow C_F = \frac{F_{thrust}}{p_0 A^*}$

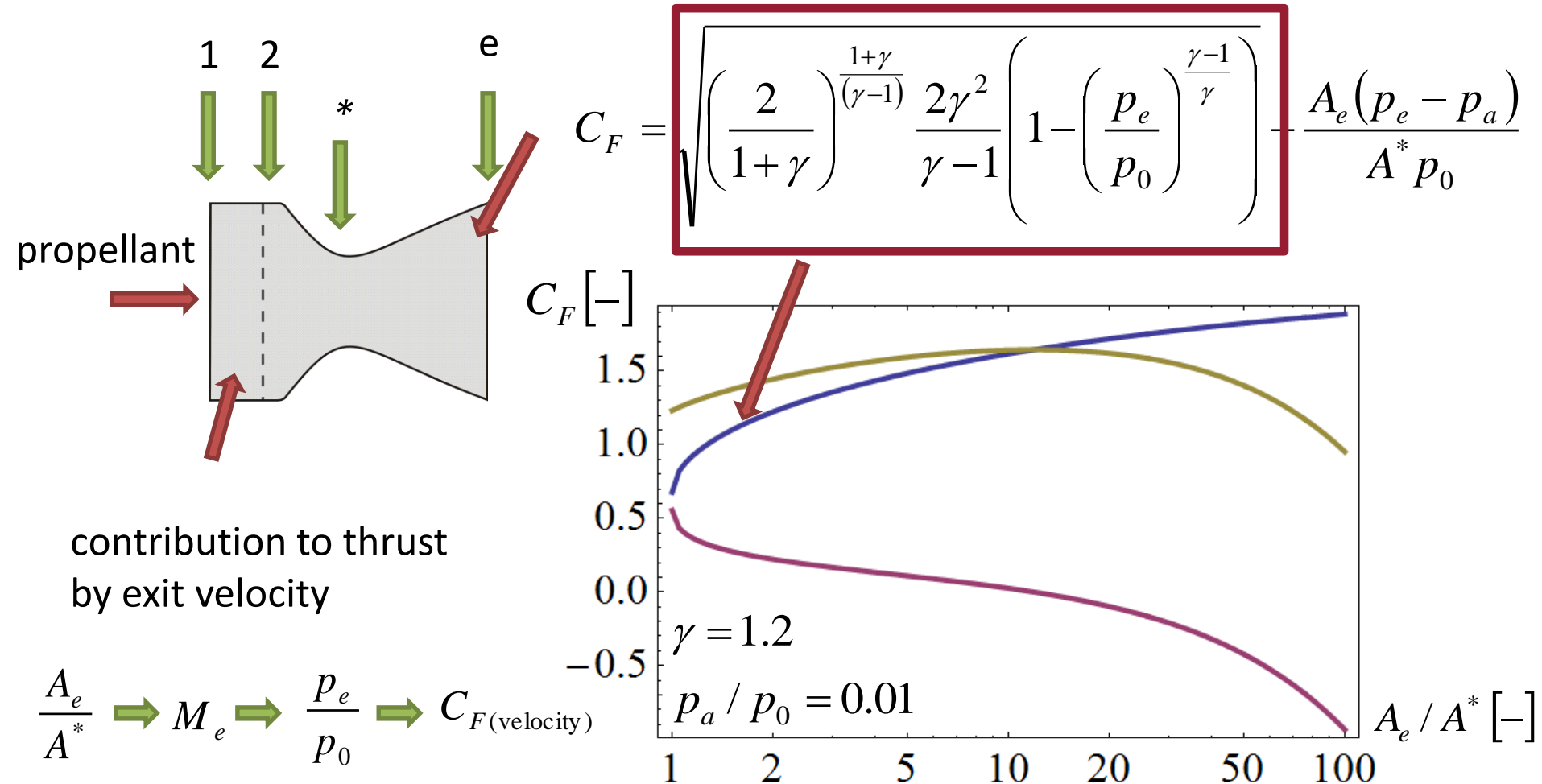
$$C_F = \sqrt{\left(\frac{2}{1+\gamma}\right)^{\frac{1+\gamma}{\gamma-1}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0}\right)^{\frac{\gamma-1}{\gamma}}\right)} + \frac{A_e(p_e - p_a)}{A^* p_0}$$

thrust coefficient depends on gas property (γ) and nozzle parameters (nozzle area ratio and pressure ratio), it is independent on combustion chamber temperature

2. Chemical Rocket Propulsion

Performance Characteristics

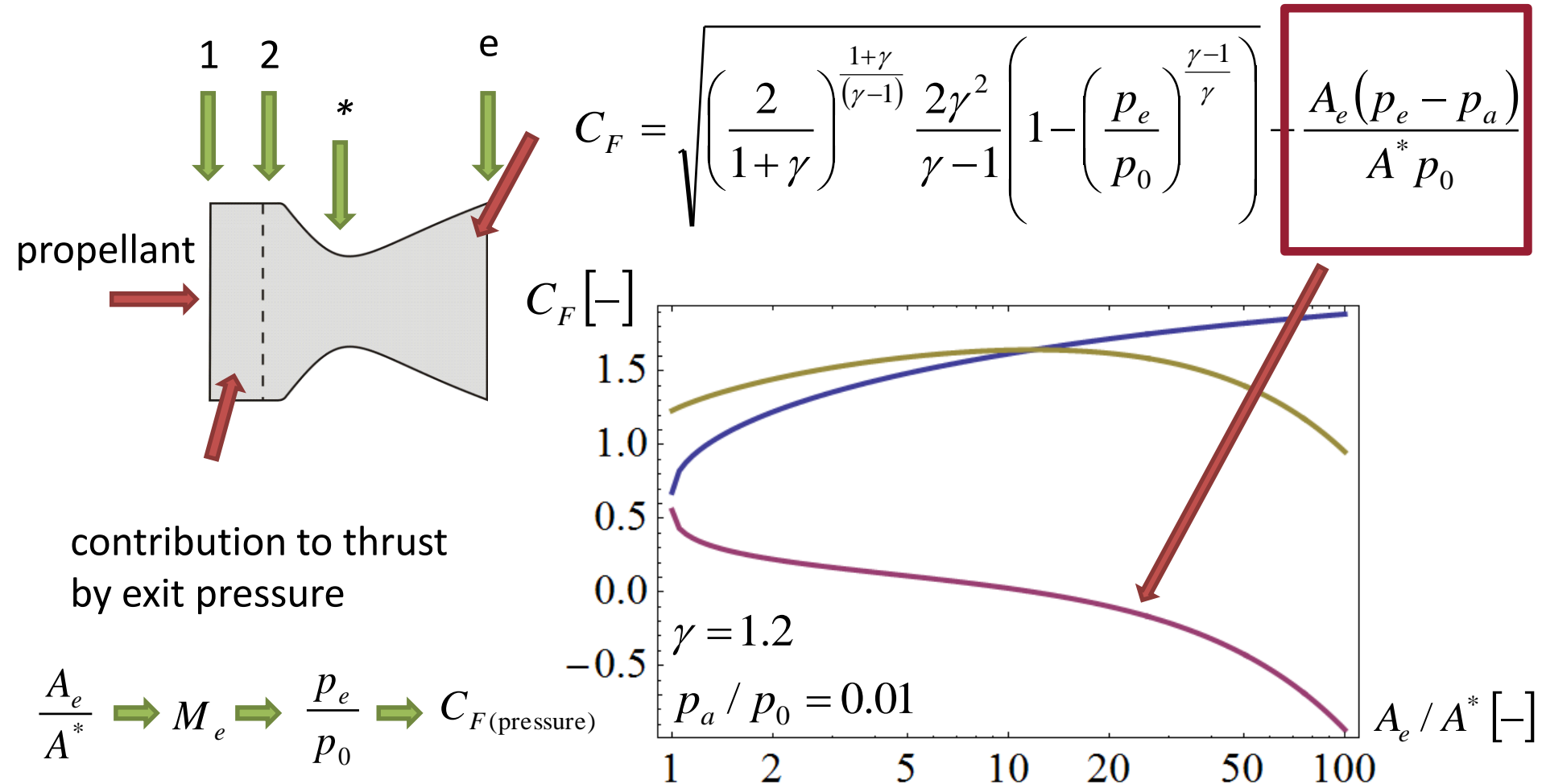
- Rocket thrust chamber – thrust coefficient



2. Chemical Rocket Propulsion

Performance Characteristics

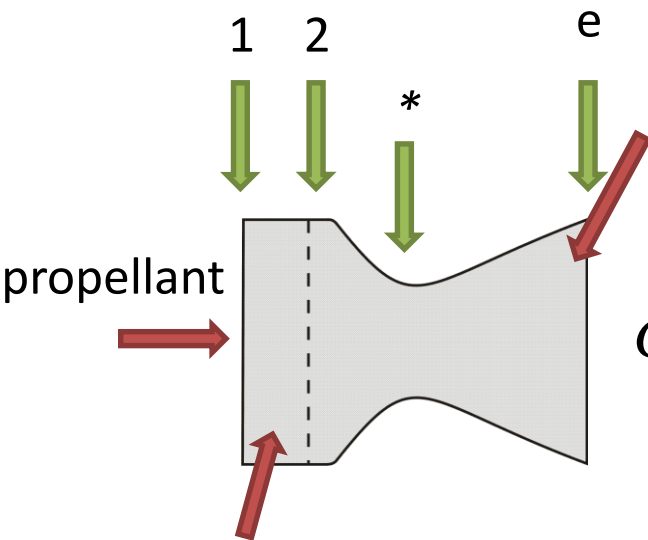
- Rocket thrust chamber – thrust coefficient



2. Chemical Rocket Propulsion

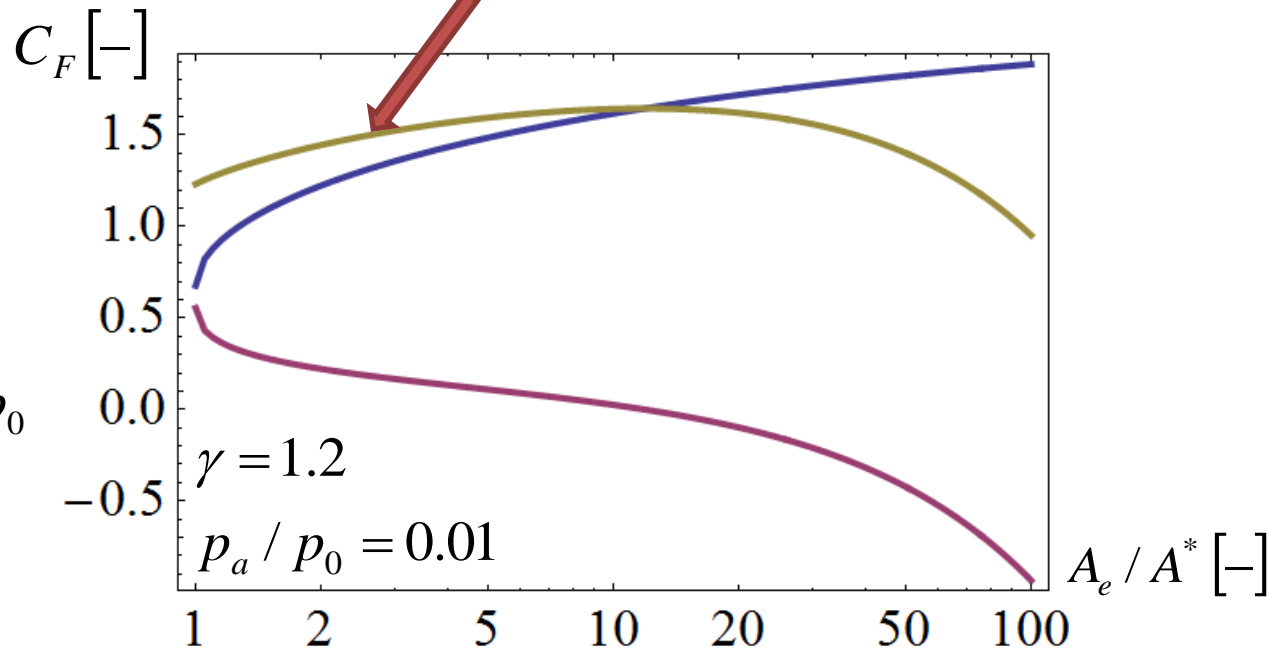
Performance Characteristics

- Rocket thrust chamber – thrust coefficient



$$C_F = \sqrt{\left(\frac{2}{1+\gamma}\right)^{\frac{1+\gamma}{(\gamma-1)}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0}\right)^{\frac{\gamma-1}{\gamma}}\right)} + \frac{A_e(p_e - p_a)}{A^* p_0}$$

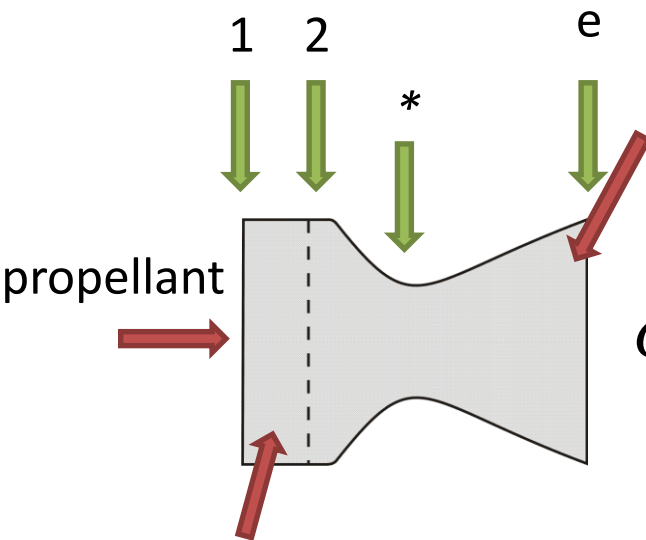
thrust coefficient for defined conditions p_a / p_0 depends on area ratio



2. Chemical Rocket Propulsion

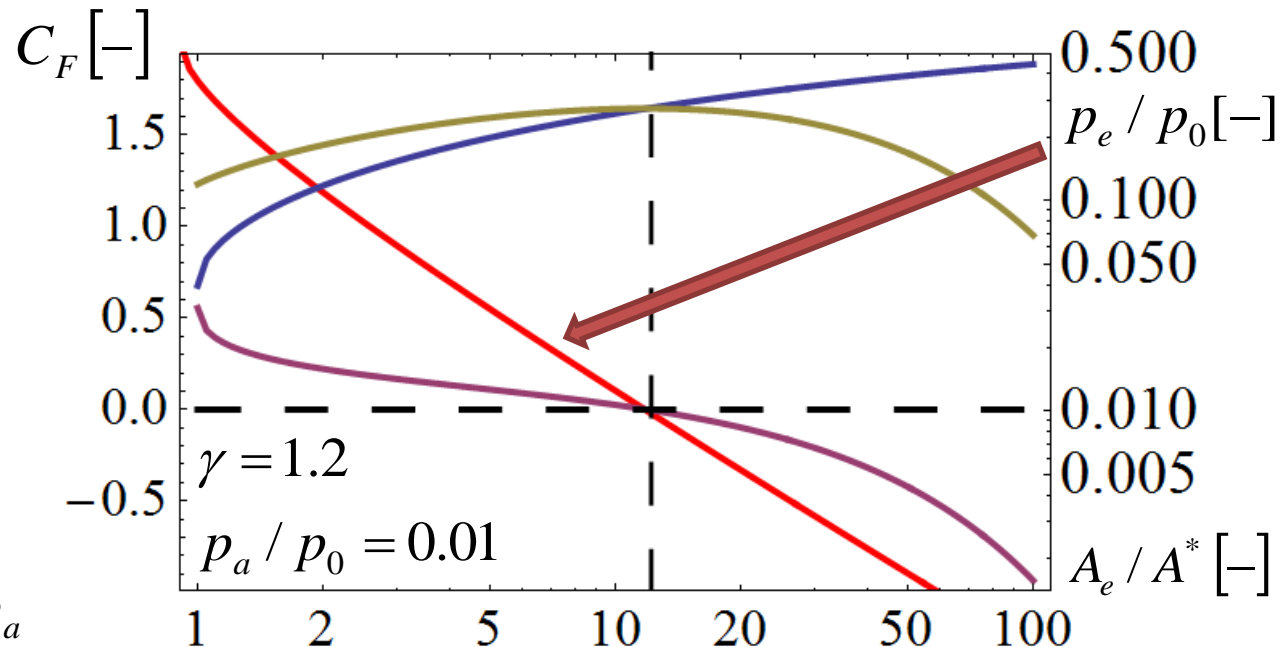
Performance Characteristics

- Rocket thrust chamber – thrust coefficient



$$C_F = \sqrt{\left(\frac{2}{1+\gamma}\right)^{\frac{1+\gamma}{\gamma-1}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0}\right)^{\frac{\gamma-1}{\gamma}}\right)} + \frac{A_e(p_e - p_a)}{A^* p_0}$$

optimal thrust coefficient is defined by area ratio, where exit pressure equals ambient pressure $p_e = p_a$

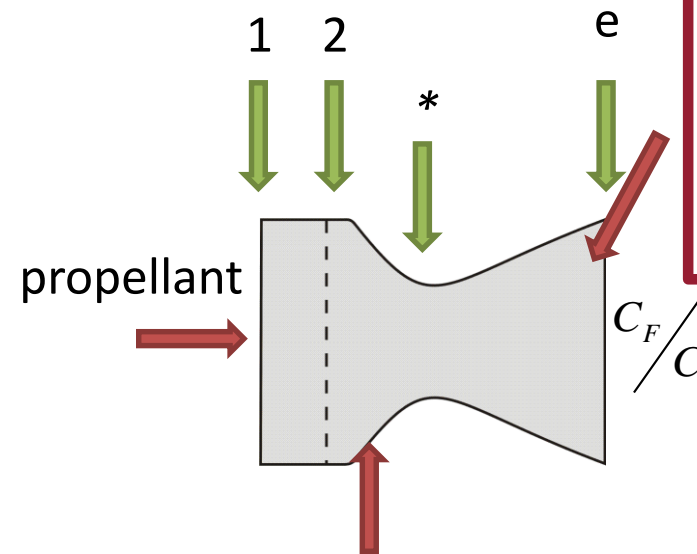


2. Chemical Rocket Propulsion

Performance Characteristics

- Rocket thrust chamber – thrust coefficient

$$C_F = \sqrt{\left(\frac{2}{1+\gamma}\right)^{\frac{1+\gamma}{\gamma-1}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0}\right)^{\frac{\gamma-1}{\gamma}}\right)} + \frac{A_e(p_e - p_a)}{A^* p_0}$$

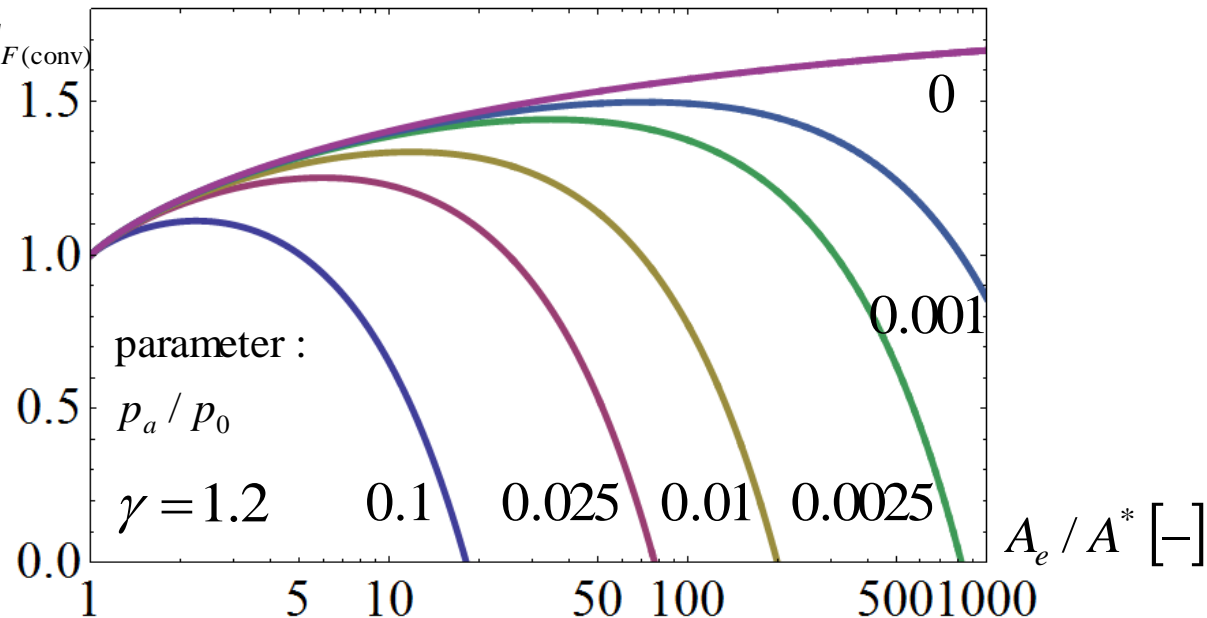


convergent part of nozzle

$$\left(1 + \frac{(\gamma-1)}{2} M^2\right)^{\frac{\gamma}{\gamma-1}} = \frac{p_0}{p_e}$$

$M=1$

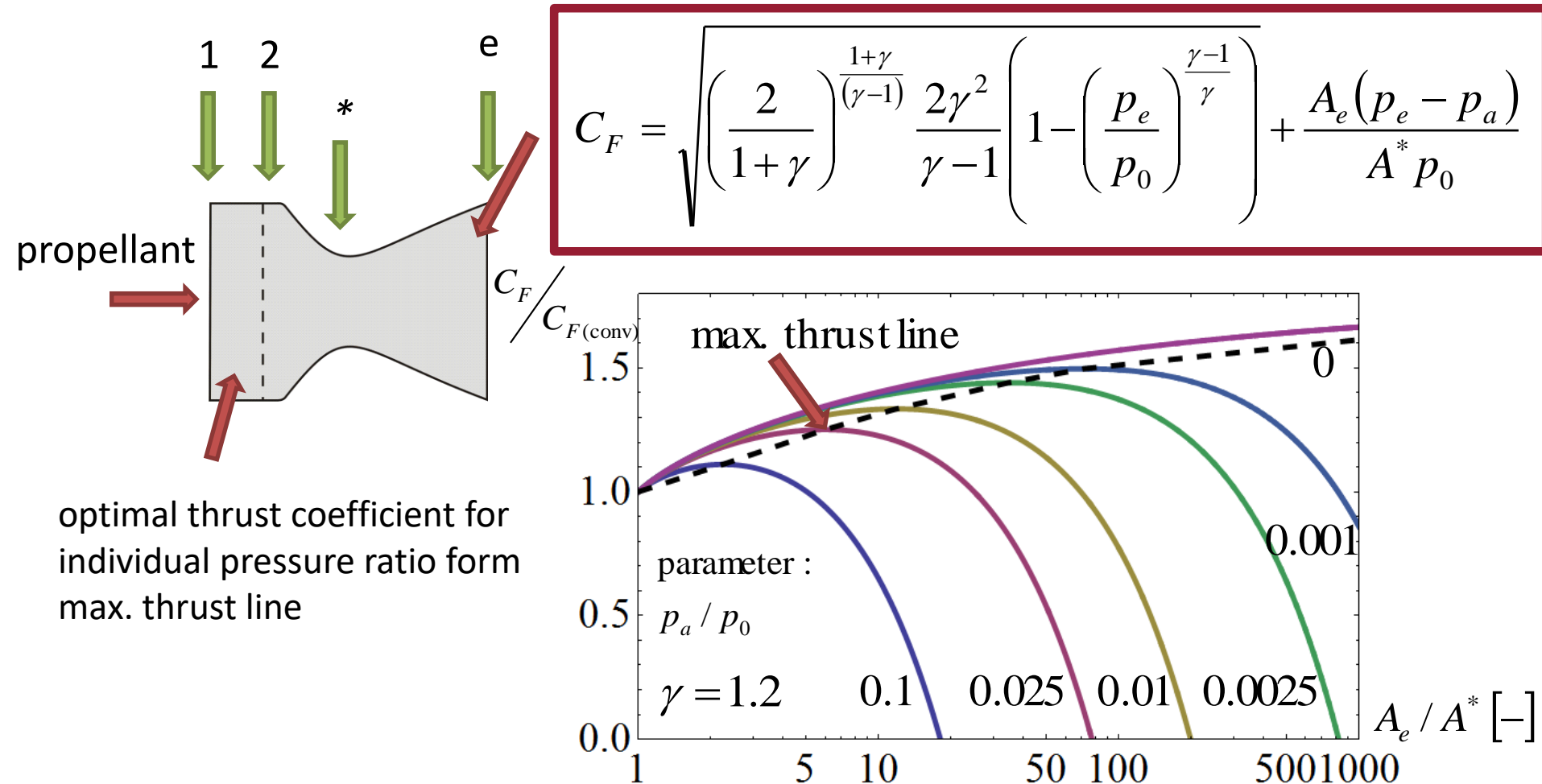
$$C_{F(\text{conv})} \leftarrow C_F$$



2. Chemical Rocket Propulsion

Performance Characteristics

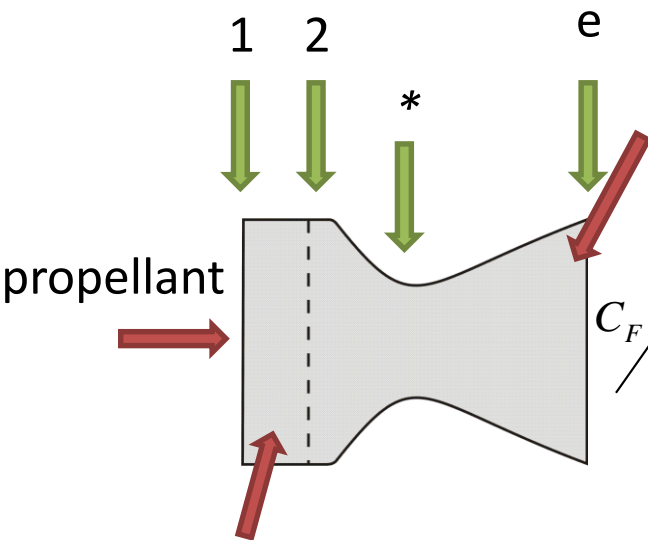
- Rocket thrust chamber – thrust coefficient



2. Chemical Rocket Propulsion

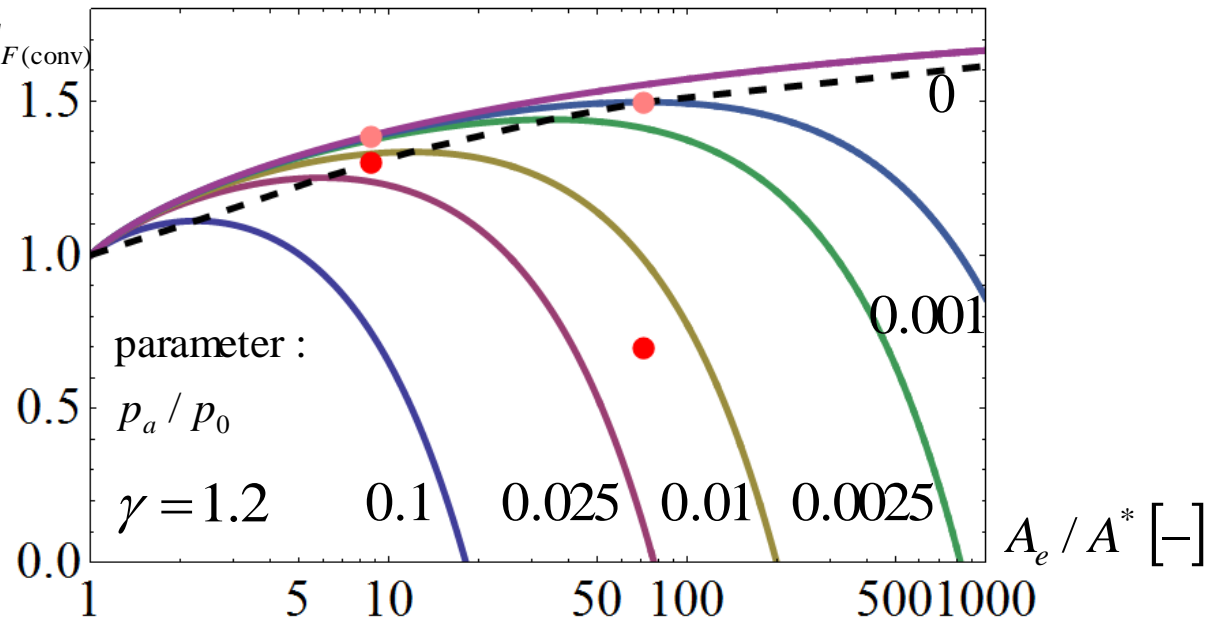
Performance Characteristics

- Rocket thrust chamber – thrust coefficient



optimal thrust coefficient for sea level and pressure ratio 0.001 and the change of the coefficient

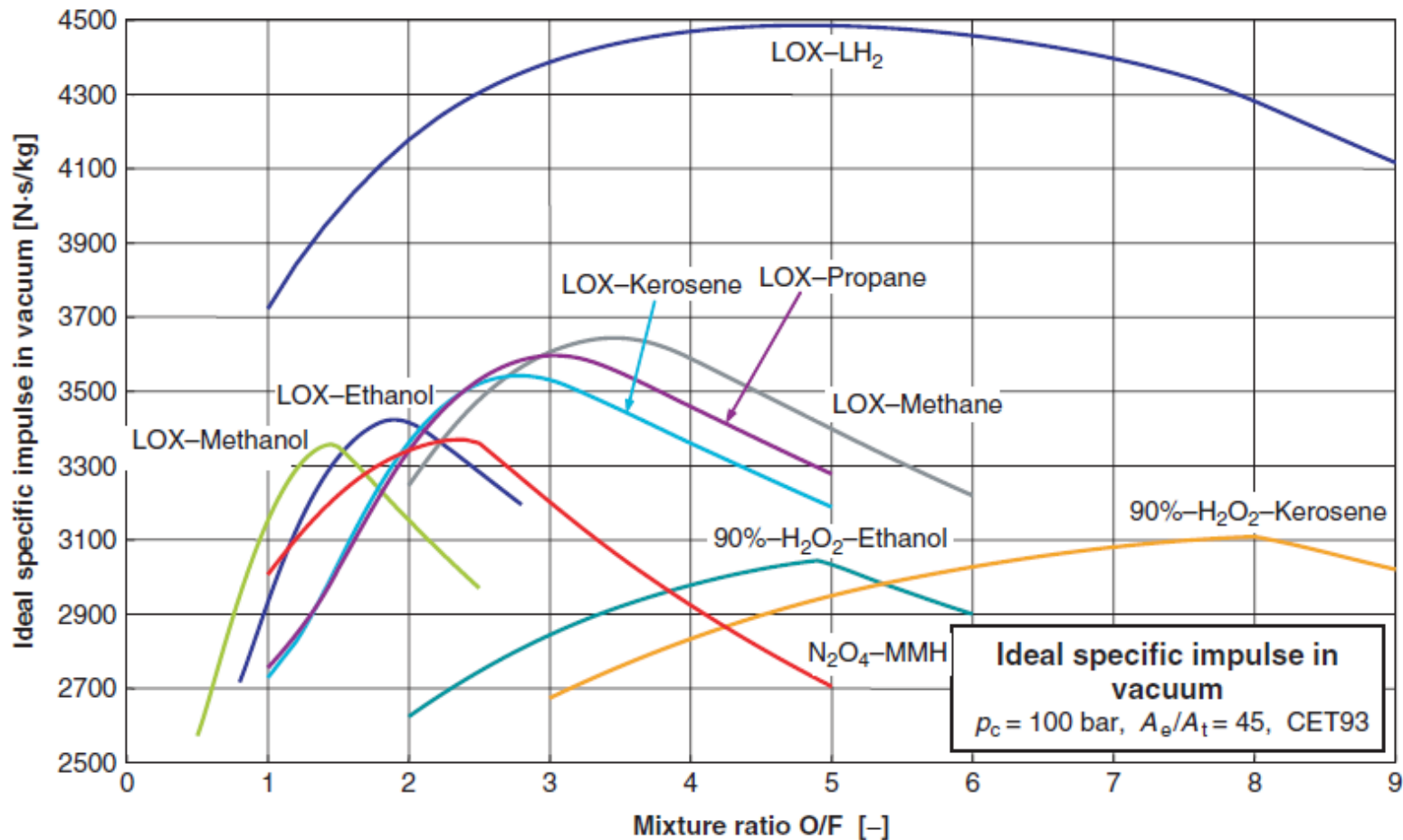
$$C_F = \sqrt{\left(\frac{2}{1+\gamma}\right)^{\frac{1+\gamma}{(\gamma-1)}} \frac{2\gamma^2}{\gamma-1} \left(1 - \left(\frac{p_e}{p_0}\right)^{\frac{\gamma-1}{\gamma}}\right)} + \frac{A_e(p_e - p_a)}{A^* p_0}$$



2. Chemical Rocket Propulsion

Liquid Propellant Performance

- performance of individual liquid propellants

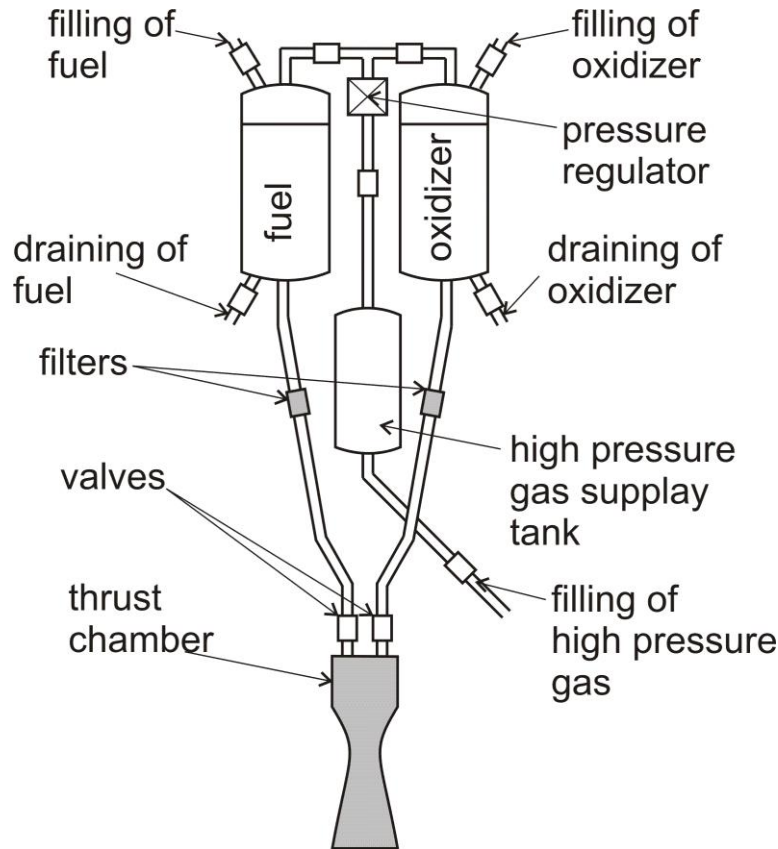


source: Ley, Wittmann, Hallmann:
Handbook of space technology

2. Chemical Rocket Propulsion

Feed System

- There are 2 main feed systems for liquid propellant:
pressurized systems

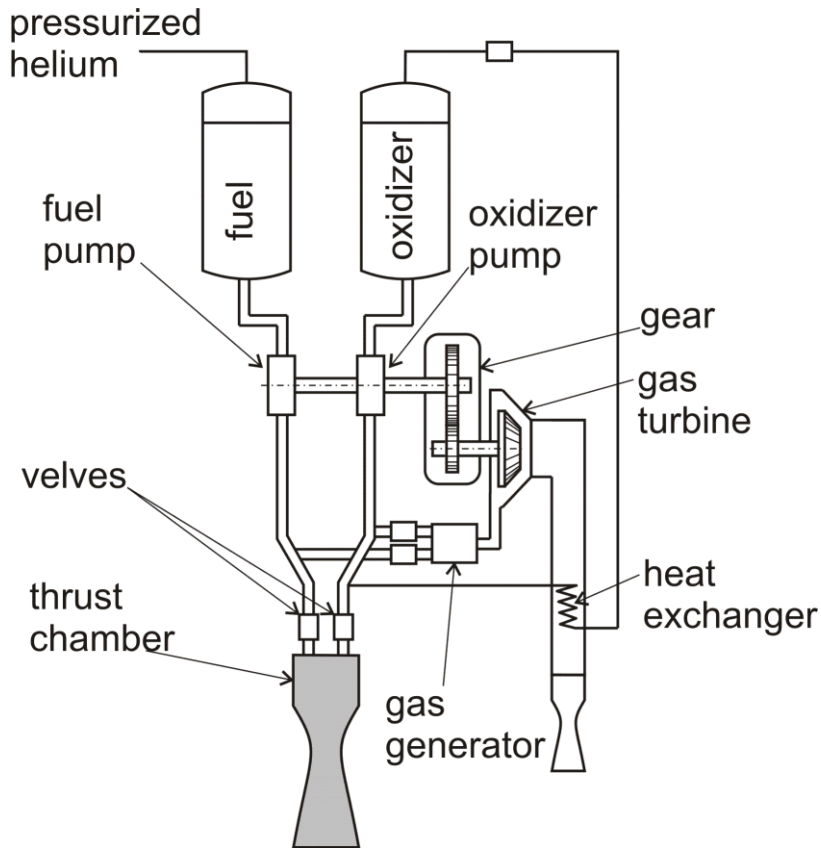


- They are usually used when:
 - total impulse is small
 - pressure in combustion chamber is small
- Disadvantages:
 - walls of tanks are thicker – system is heavier
- Usage:
 - control of attitude and change of orbit

2. Chemical Rocket Propulsion

Feed System

- There are 2 main feed systems for liquid propellant:
turbopump systems



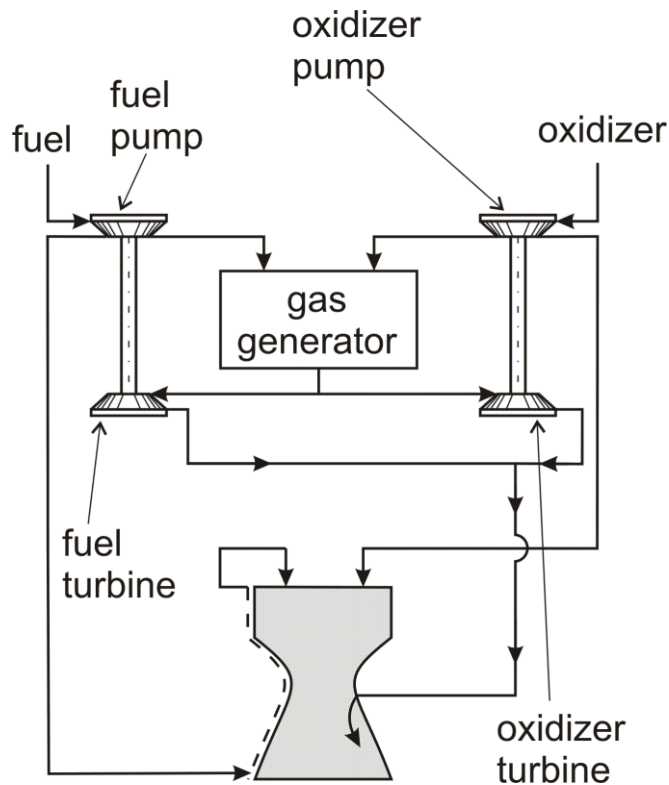
- They are usually used when:
 - total impulse is large
 - pressure in combustion chamber is large
- Positive characteristics of system:
 - pressure in tanks is lower than pressure in tanks when gas pressure feed system is used so the thickness of walls of tank is smaller
- Usage:
 - dominantly for boosters

2. Chemical Rocket Propulsion

Feed System

- turbopump systems – 3 basic cycles

Gas generator cycle - open cycle



Description:

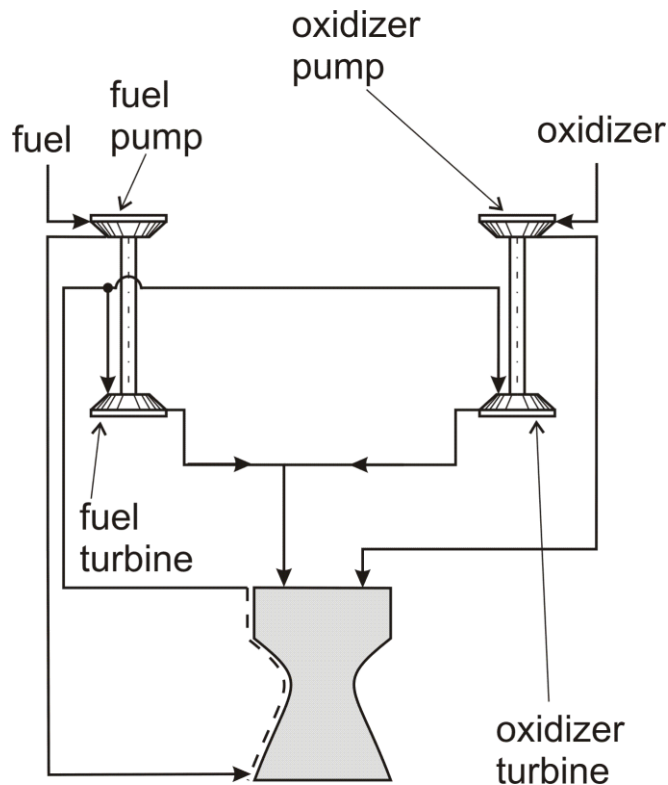
- It is the most common cycle
- It is relatively simple cycle
- The cycle efficiency is smaller than efficiency of closed cycle
- Small part of the propellant is consumed in small combustion chamber for generating gas for a turbine, which drives the pump
- Gas from turbine flows to separate nozzle or to the end part of the main nozzle, where it operates as cooler of nozzle
- Engines: F-1 (Saturn V) , 2 Vulcain (Ariane 5)

2. Chemical Rocket Propulsion

Feed System

- turbopump systems – 3 basic cycles

Expander cycle - closed cycle



Description:

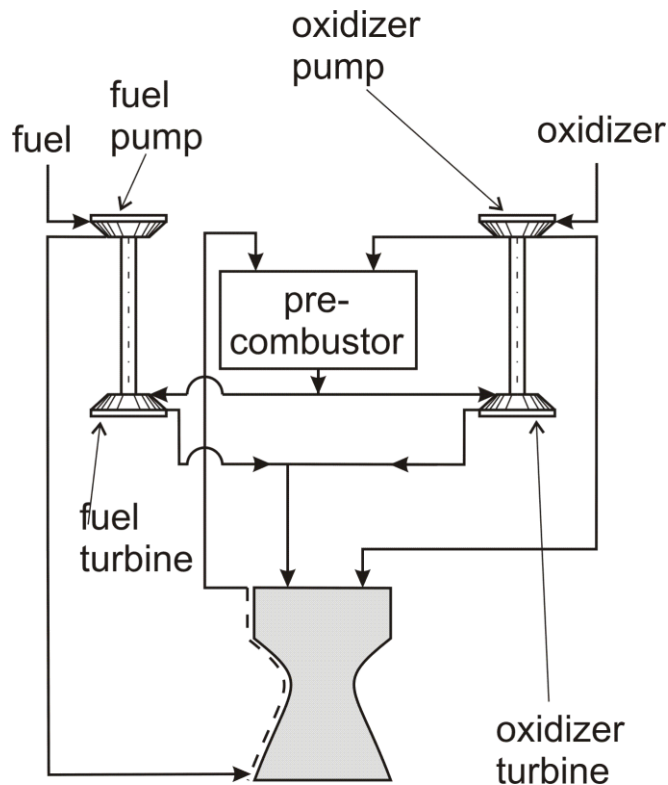
- The fuel passed through the cooling jacket of nozzle where it picked up energy and the fuel works as coolant of nozzle
- The fuel is evaporated, heated, and then fed to low pressure-ratio turbines
- at the outlet of the turbine fuel enters the combustion chamber where it is mixed with an oxidizer
- in that cycle all the fuel is burnt in combustion chamber and the efficiency of engine is increased
- Engines: RL10 (the second stage of the Delta IV) , Vinci (ESA)

2. Chemical Rocket Propulsion

Feed System

- turbopump systems – 3 basic cycles

Staged-combustion cycle - closed cycle



Description:

- The fuel passed through the cooling jacket of nozzle as in expander cycle
- Then the fuel flows into the precombustor where all the fuel is burnt with a part of the oxidizer, forming a high-energy gas to drive
- The turbines that drive the pumps all the gas at the outlet of the turbine flows into the combustion chamber where is mixed with remaining oxidizer
- pressure in combustion chamber: up to 40 MPa
- Engines: Space Shuttle Main Engine – SSME, RD-170 (Energija)

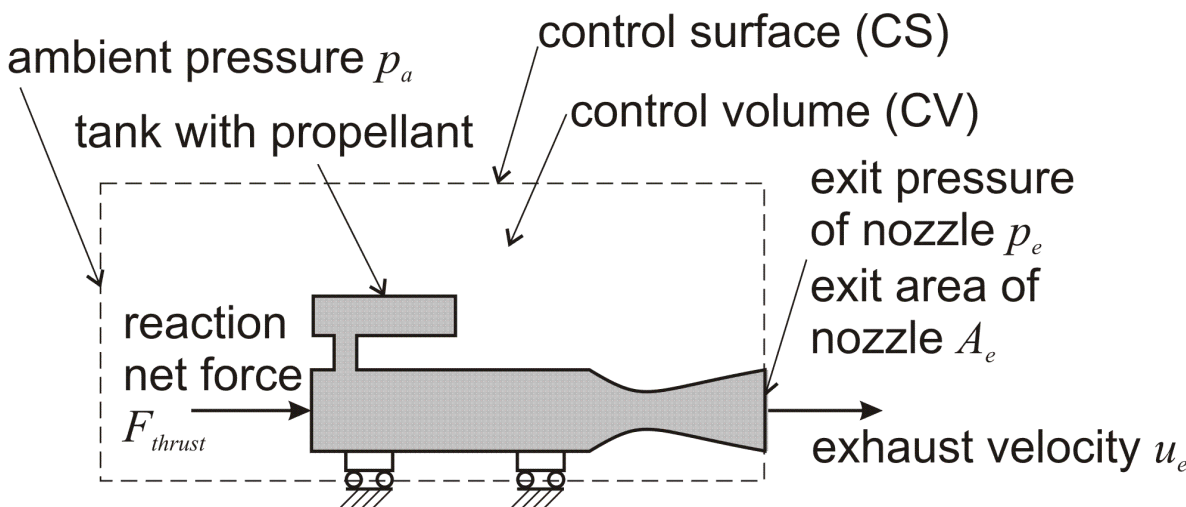
3. Performance of Rocket Vehicle

- Static Performance
- Force-Free Motion
- Motion with Gravity
- Launch Flight Mechanics

3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:
- simplifications:
 - quasi-one dimensional flow
 - steady-state flow

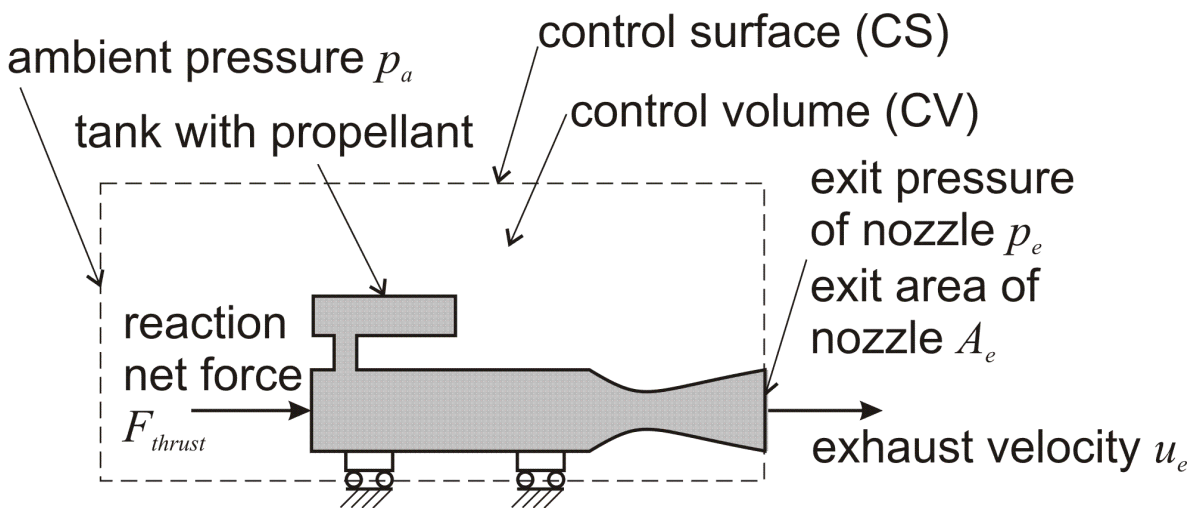


3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:

$$\left[\begin{array}{l} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right]$$

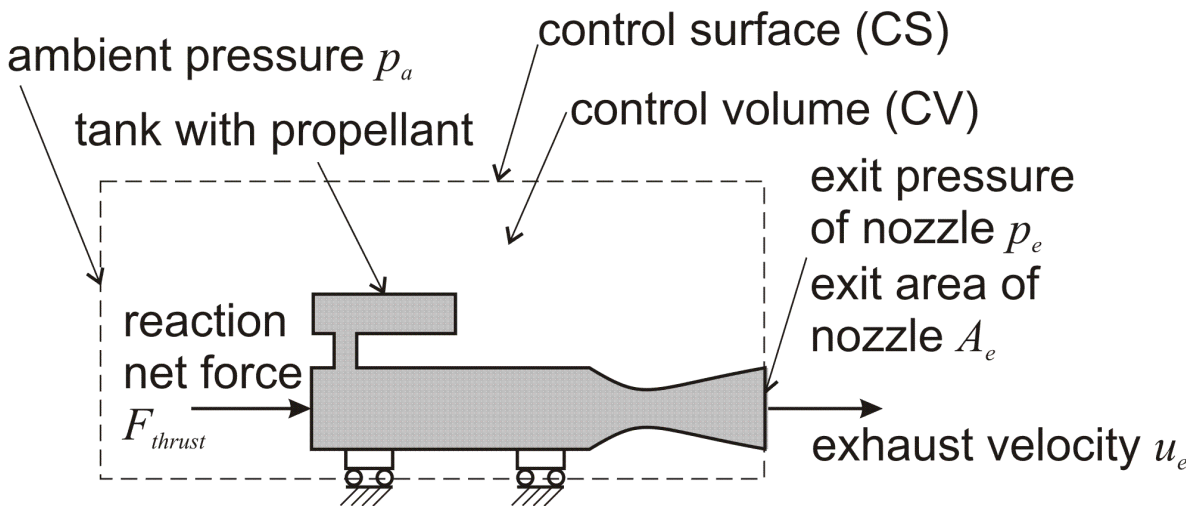


3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:

$$\left[\begin{array}{l} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right] \Rightarrow F_{thrust} + A_e(p_a - p_e) = \dot{m}u_e$$



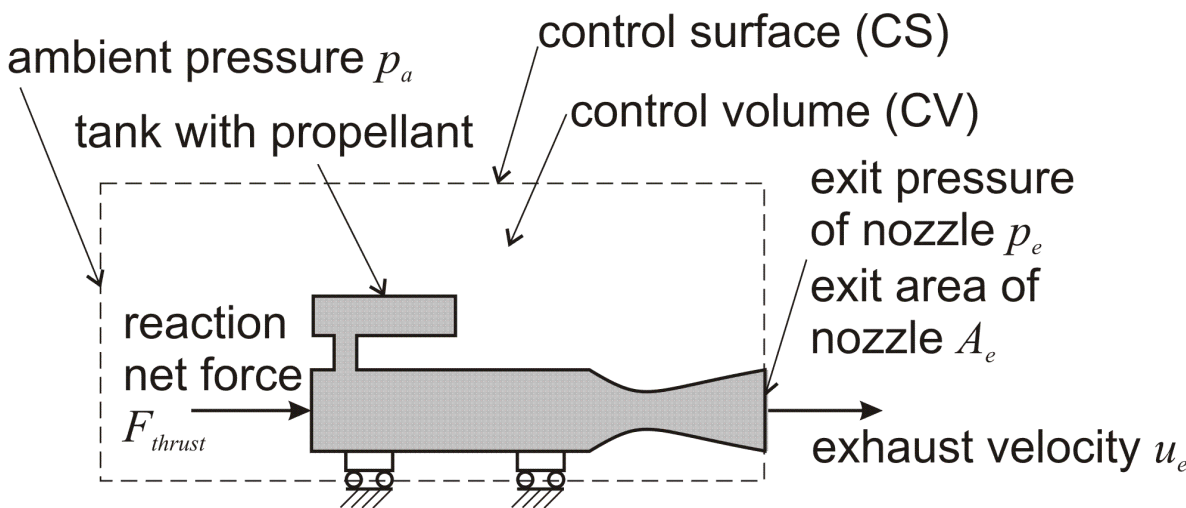
3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:

$$\left[\begin{array}{l} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{l} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right] \quad \Rightarrow \quad F_{thrust} + A_e(p_a - p_e) = \dot{m}u_e$$

$$\begin{aligned} F_{thrust} &= \dot{m}u_e + A_e(p_e - p_a) = \\ &= \dot{m} \left(u_e + A_e \left(\frac{p_e - p_a}{\dot{m}} \right) \right) \end{aligned}$$



3. Performance of Rocket Vehicle

Static Performance

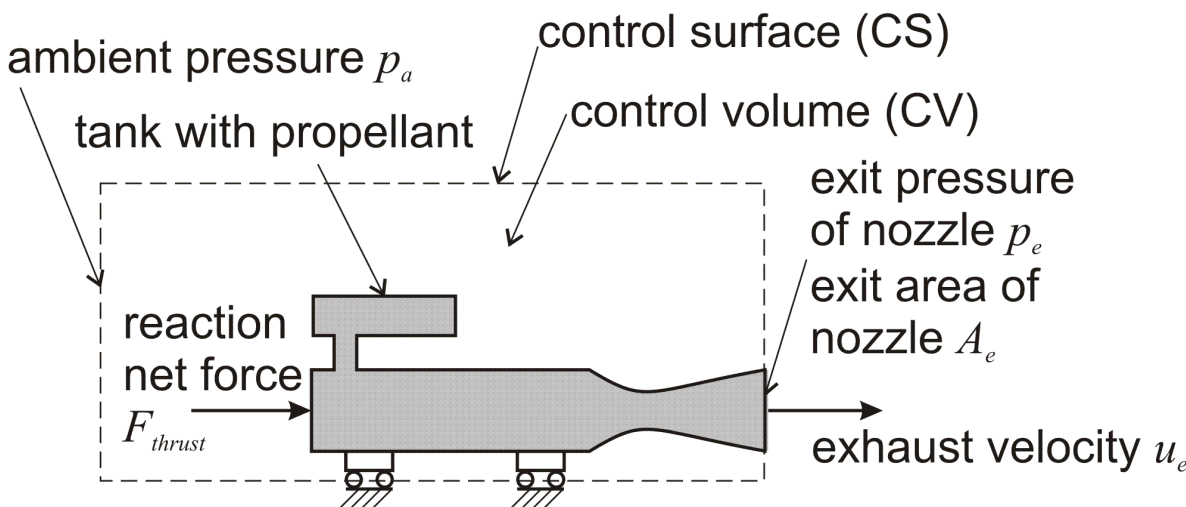
- Momentum equation – written for CV:

$$\left[\begin{array}{c} \text{Forces on} \\ \text{gas in CV in} \\ \text{direction } x \end{array} \right] = \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{leaves CV} \end{array} \right] - \left[\begin{array}{c} \text{Rate} \\ \text{momentum} \\ \text{enters CV} \end{array} \right] \Rightarrow F_{thrust} + A_e(p_a - p_e) = \dot{m}u_e$$

$$\begin{aligned} F_{thrust} &= \dot{m}u_e + A_e(p_e - p_a) = \\ &= \dot{m} \left(u_e + A_e \left(\frac{p_e - p_a}{\dot{m}} \right) \right) \end{aligned}$$

$$F_{thrust} = \dot{m}u_{ef}$$

Thrust

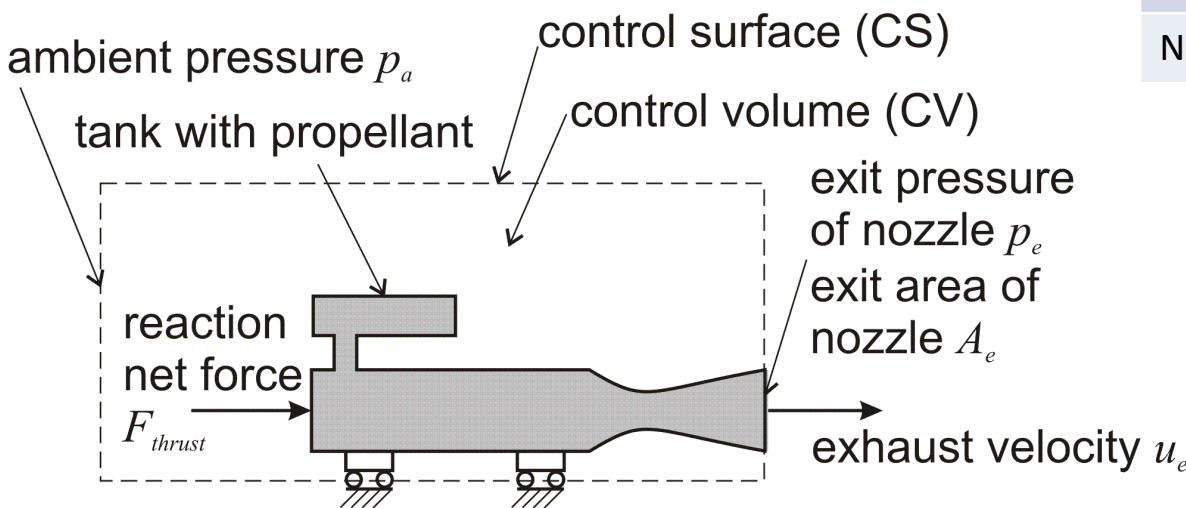


3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:

Engine	Thrust [MN]
F1	7.77 (vacuum)
Vulcain 2	1.35
J2	1.03
NK33	1.51



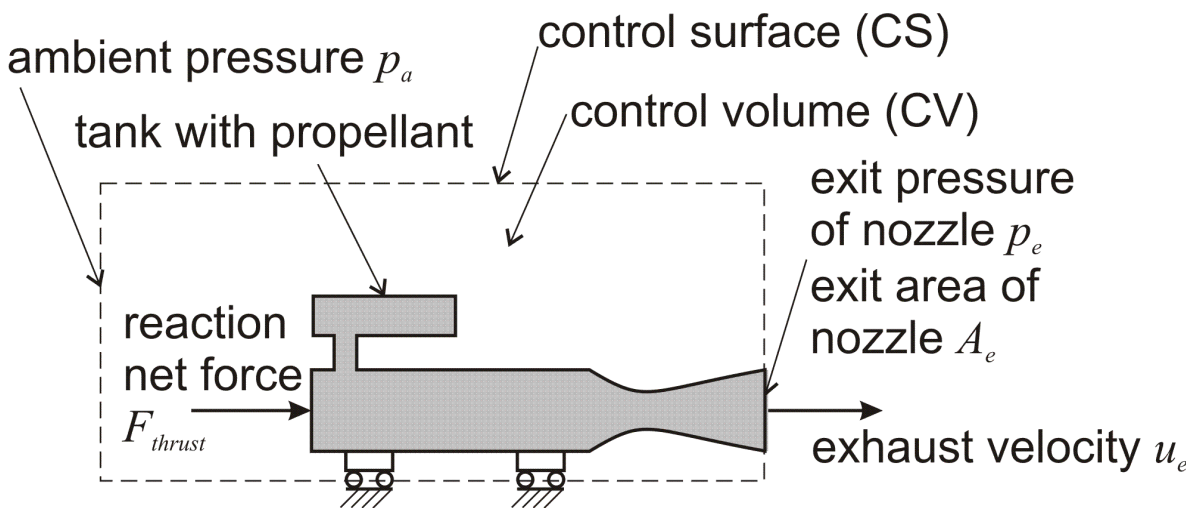
$$F_{thrust} = \dot{m}u_{ef}$$

Thrust

3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:



Total impulse

$$I_t = F_{thrust} t$$

$$F_{thrust} = \dot{m} u_{ef}$$

Thrust

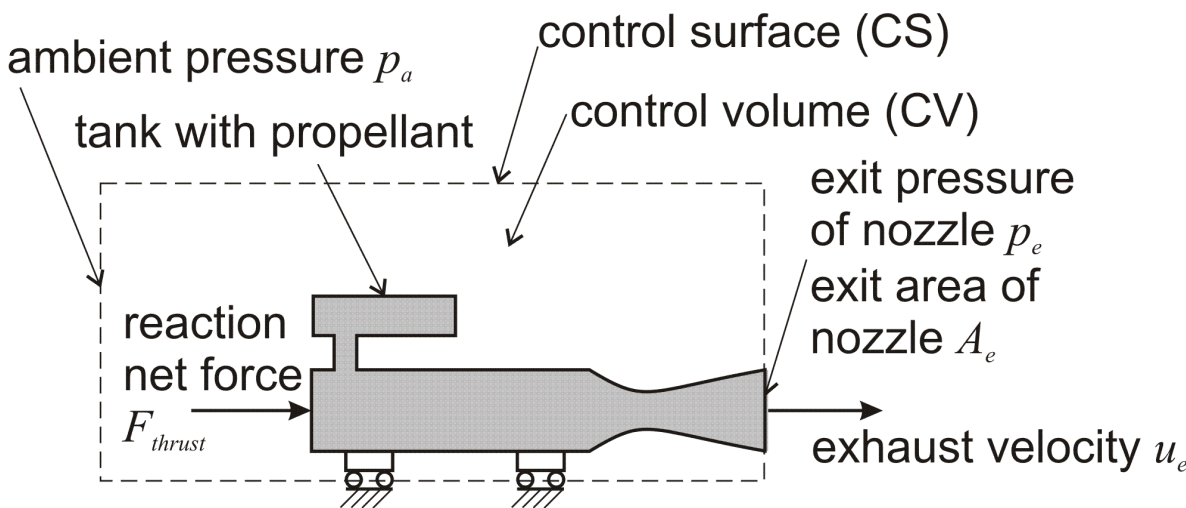
3. Performance of Rocket Vehicle

Static Performance

- Momentum equation – written for CV:

Specific impulse

$$I_s = I_t / mg = F_{thrust} / \dot{m}g = \dot{m}u_{ef} / \dot{m}g = u_{ef} / g$$



Total impulse

$$I_t = F_{thrust} t$$

$$F_{thrust} = \dot{m}u_{ef}$$

Thrust

3. Performance of Rocket Vehicle

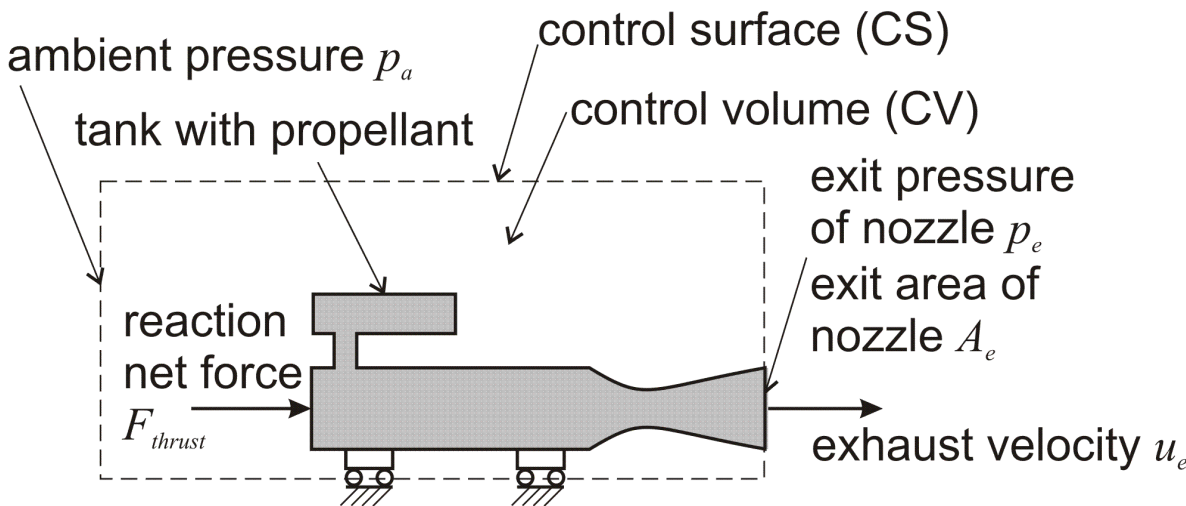
Static Performance

- Momentum equation – written for CV:

Propellant	Specific impulse I_s [s]
cold gas	50
Monopropellant hydrazine	230
LOX/LH2	455
Ion propulsion	>3000

Specific impulse

$$I_s = I_t / mg = F_{thrust} / \dot{m}g = \dot{m}u_{ef} / \dot{m}g = u_{ef} / g$$



Total impulse

$$I_t = F_{thrust} t$$

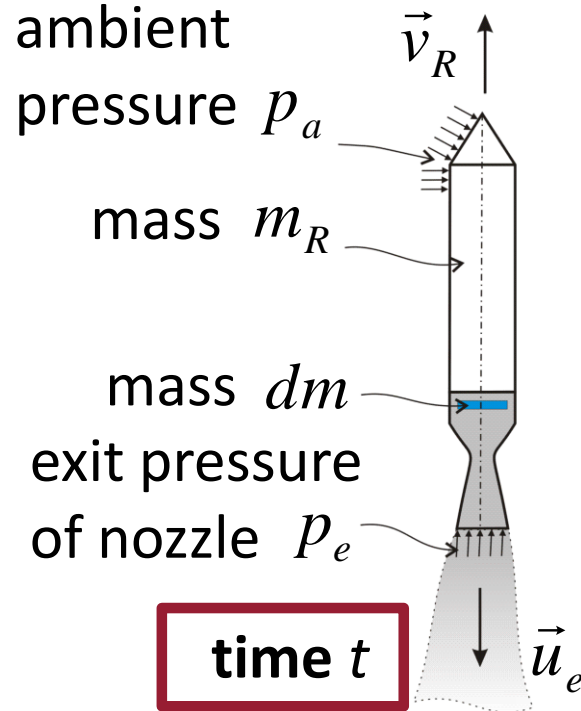
$$F_{thrust} = \dot{m}u_{ef}$$

Thrust

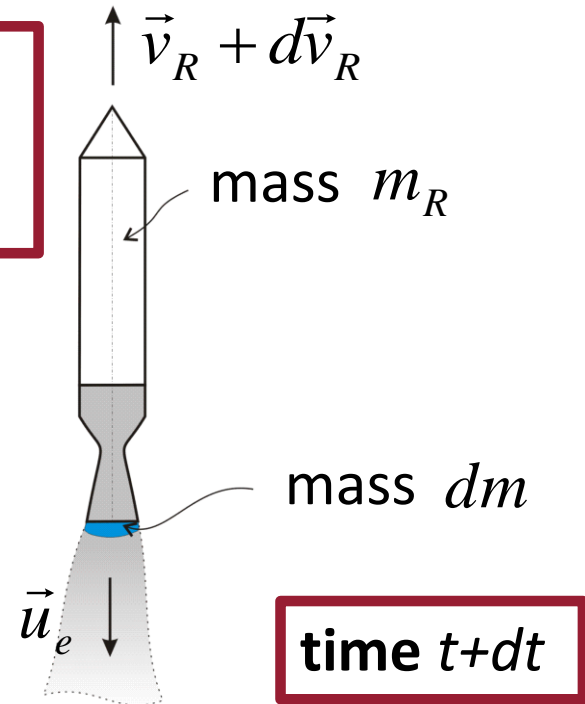
3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion \rightarrow absence of external forces



only ambient pressure is considered



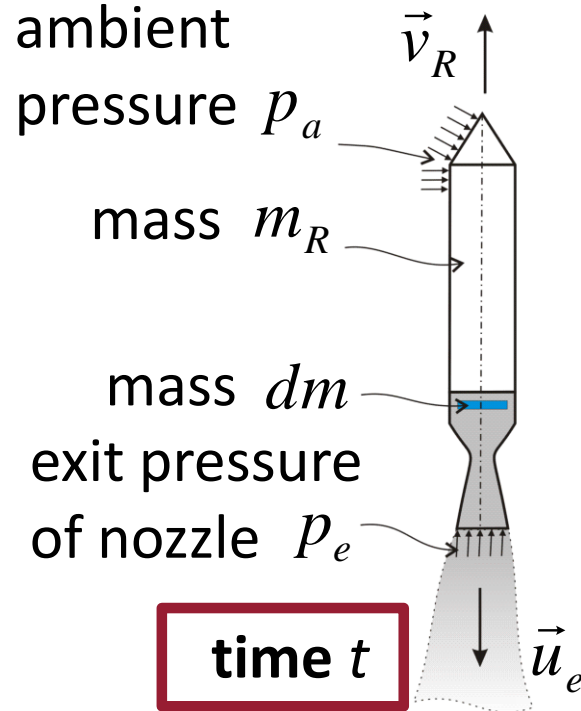
Momentum: $(m_R + dm)\vec{v}_R \rightarrow m_R(\vec{v}_R + d\vec{v}_R) + dm(\vec{v}_R + \vec{u}_e)$

Change of momentum in dt $m_R d\vec{v}_R + dm\vec{u}_e$

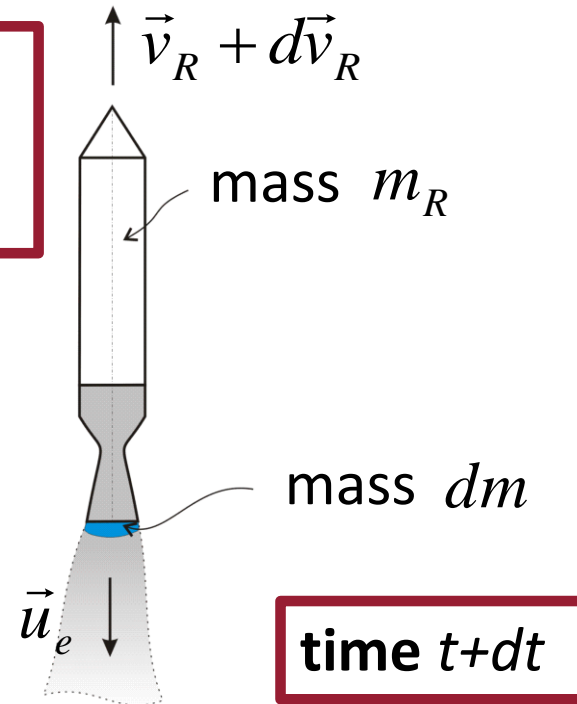
3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion \rightarrow absence of external forces



only ambient pressure is considered



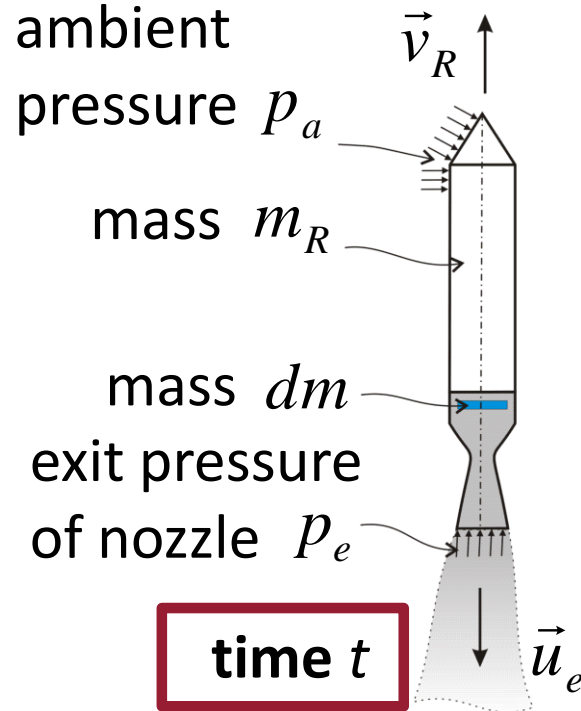
Pressure force: $(p_e - p_a)A_e \vec{i}_R \rightarrow (p_e - p_a)A_e \vec{i}_R$

Total impulse in dt : $(p_e - p_a)A_e \vec{i}_R dt$

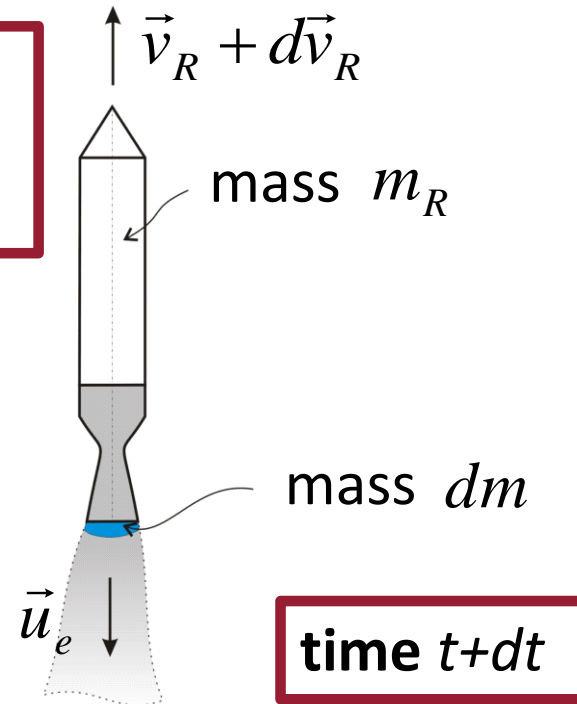
3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion  absence of external forces



only ambient pressure is considered



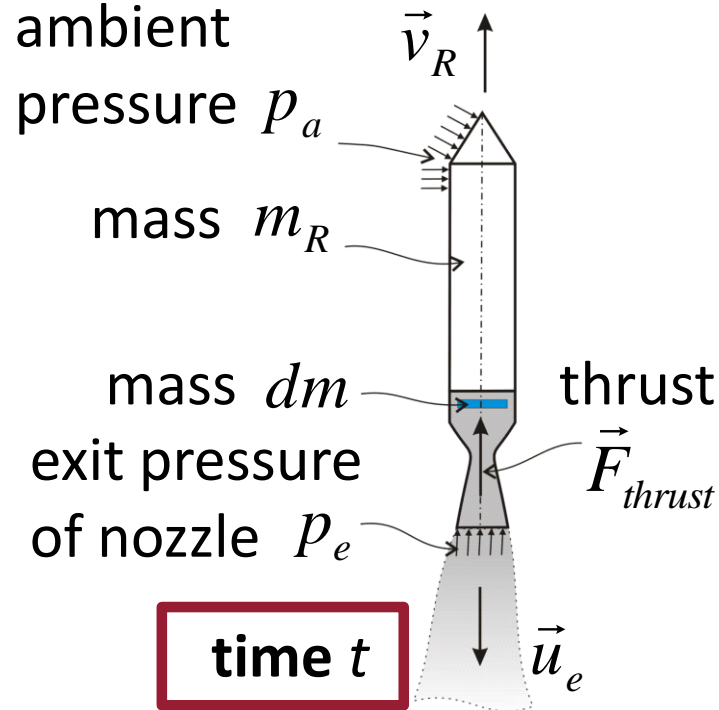
Momentum equation: $m_R d\vec{v}_R + dm\vec{u}_e = (p_e - p_a)A_e \vec{i}_R dt$

Momentum equation in \vec{i}_R : $m_R dv_R = dm u_e + (p_e - p_a)A_e dt$

3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion → absence of external forces



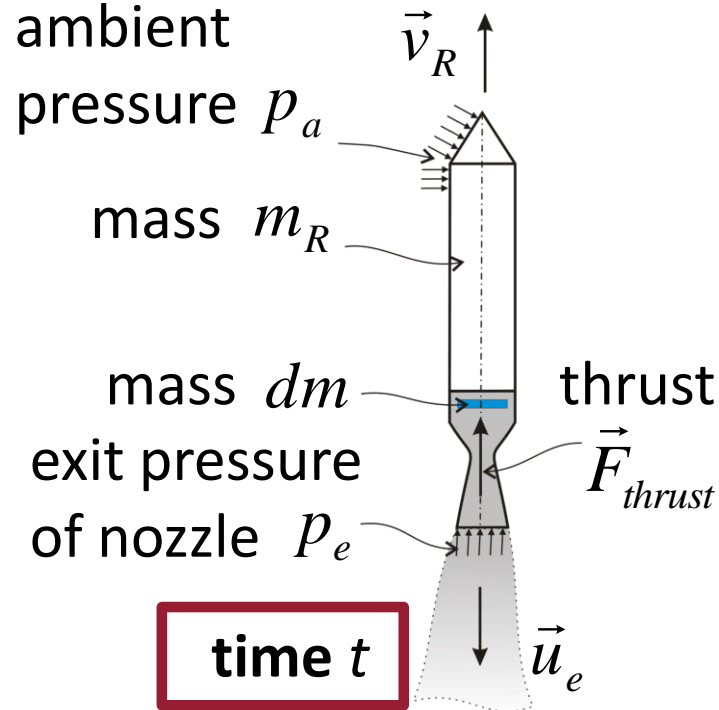
$$dm = \dot{m}dt$$

Momentum equation in \vec{i}_R : $m_R dv_R = dm u_e + (p_e - p_a) A_e dt$

3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion → absence of external forces



$$m_R dv_R = (\dot{m}u_e + (p_e - p_a)A_e)dt$$

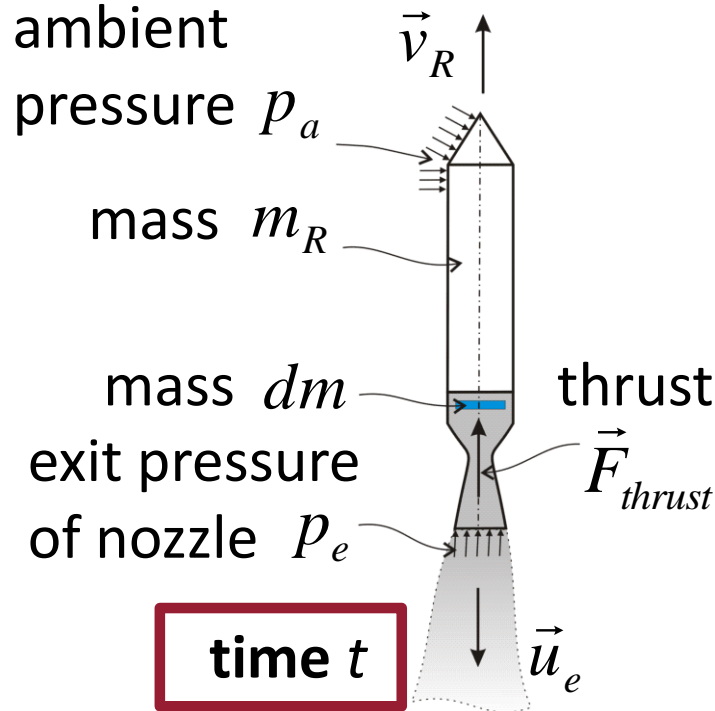
$$dm = \dot{m}dt$$

Momentum equation in \vec{i}_R : $m_R dv_R = dm u_e + (p_e - p_a)A_e dt$

3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion → absence of external forces



$$dm = \dot{m}dt$$

$$m_R \frac{dv_R}{dt} = F_{thrust}$$

$$m_R dv_R = \dot{m}u_{ef}dt$$

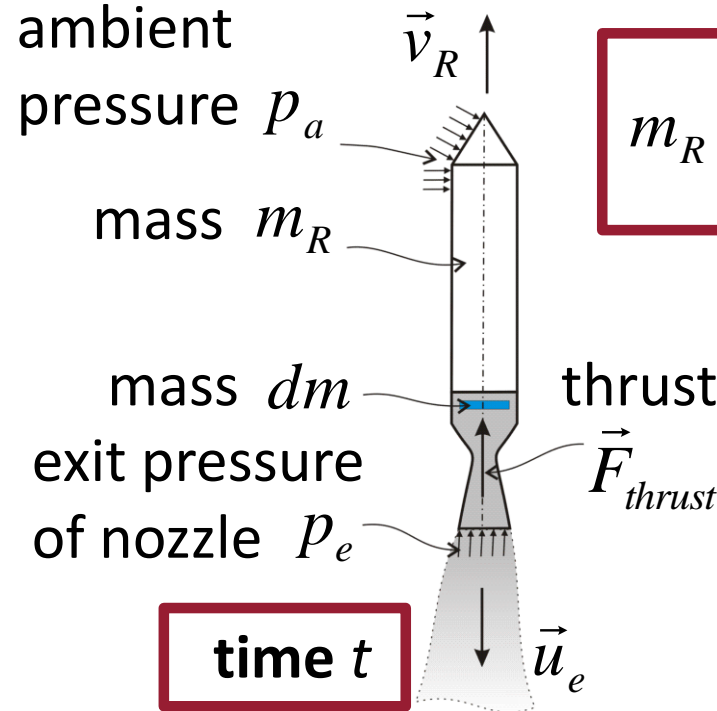
$$m_R dv_R = (\dot{m}u_e + (p_e - p_a)A_e)dt$$

Momentum equation in \vec{i}_R : $m_R dv_R = dm u_e + (p_e - p_a)A_e dt$

3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion → absence of external forces



$$m_R \frac{d\vec{v}_R}{dt} = \vec{F}_{thrust}$$

$$m_R \frac{dv_R}{dt} = F_{thrust}$$

$$m_R dv_R = \dot{m} u_{ef} dt$$

$$m_R dv_R = (\dot{m} u_e + (p_e - p_a) A_e) dt$$

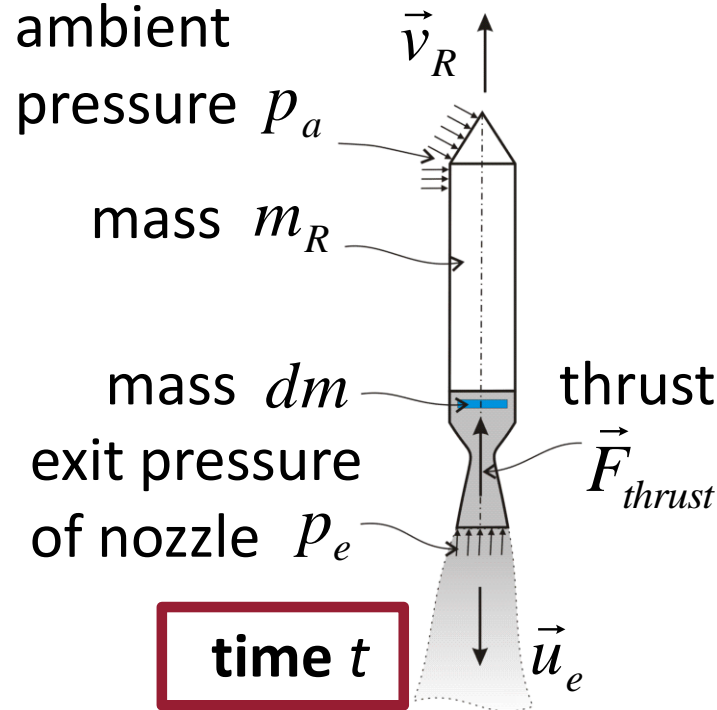
$$dm = \dot{m} dt$$

Momentum equation in \vec{i}_R : $m_R dv_R = dm u_e + (p_e - p_a) A_e dt$

3. Performance of Rocket Vehicle

Force-Free Motion

- Force-free motion → absence of external forces



$$dv_R = -\frac{dm_R}{m_R} u_{ef}$$

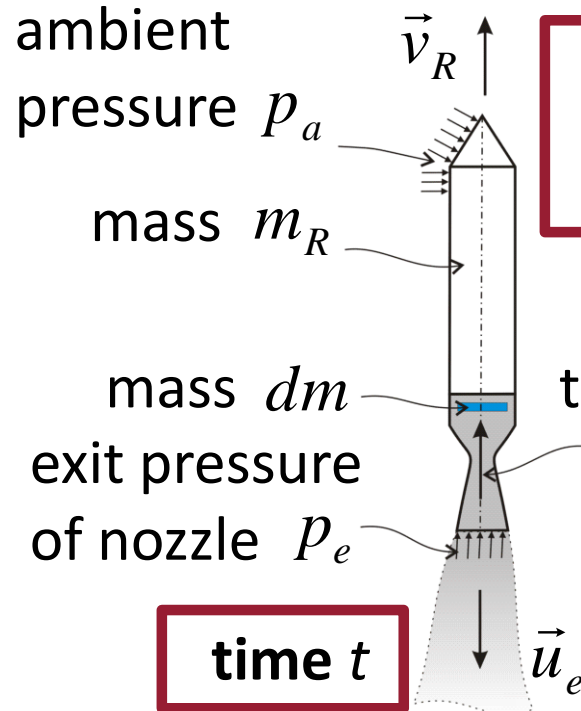
$$m_R dv_R = \dot{m} u_{ef} dt$$

$$\frac{dm_R}{dt} = -\dot{m}$$

3. Performance of Rocket Vehicle

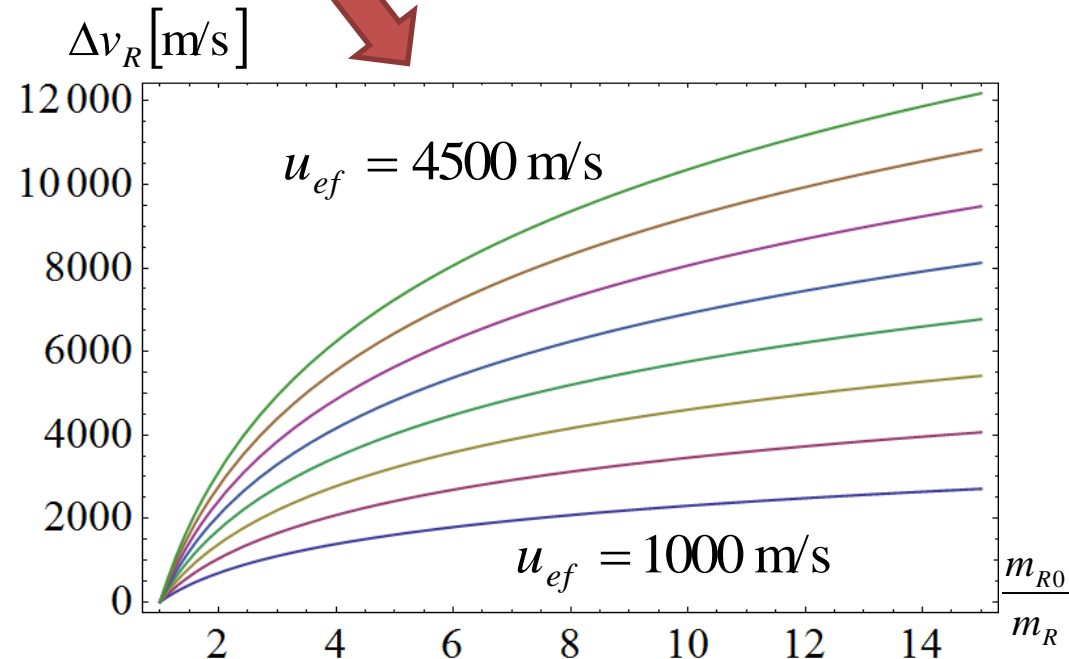
Force-Free Motion

- Force-free motion → absence of external forces



$$\Delta v_R = u_{ef} \ln \left(\frac{m_{R0}}{m_R} \right)$$

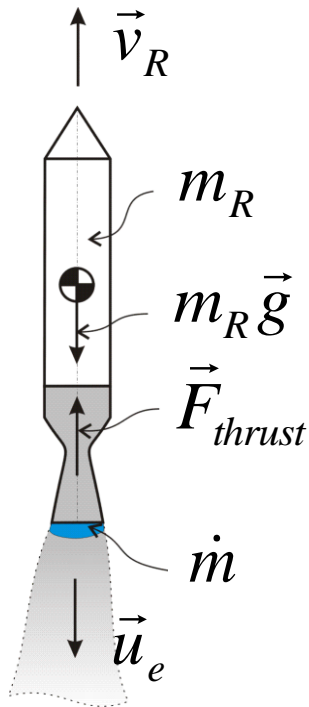
$$dv_R = - \frac{dm_R}{m_R} u_{ef}$$



3. Performance of Rocket Vehicle

Motion with Gravity

- Motion with gravity → vertical motion in gravity field



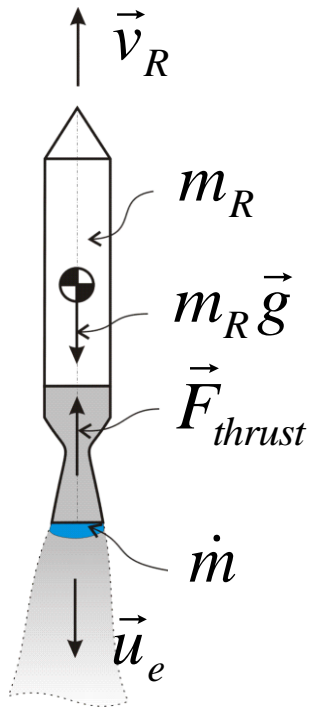
$$m_R \frac{dv_R}{dt} = - \frac{dm_R}{dt} u_{ef} - m_R g$$

$$m_R \frac{d\vec{v}_R}{dt} = \frac{dm_R}{dt} \vec{u}_{ef} + m_R \vec{g}$$

3. Performance of Rocket Vehicle

Motion with Gravity

- Motion with gravity \rightarrow vertical motion in gravity field



$$dv_R = -\frac{dm_R}{m_R} u_{ef} - g dt$$

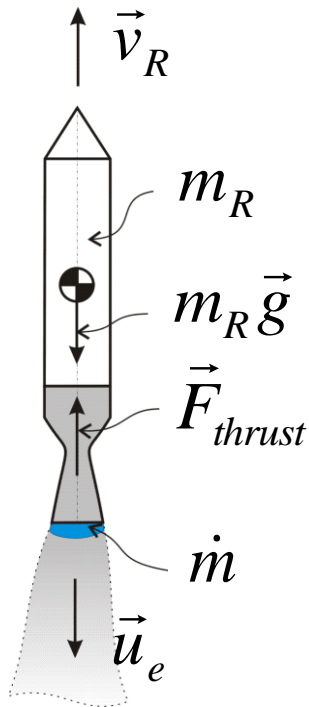
$$m_R \frac{dv_R}{dt} = -\frac{dm_R}{dt} u_{ef} - m_R g$$

$$m_R \frac{d\vec{v}_R}{dt} = \frac{dm_R}{dt} \vec{u}_{ef} + m_R \vec{g}$$

3. Performance of Rocket Vehicle

Motion with Gravity

- Motion with gravity \rightarrow vertical motion in gravity field



$$dv_R = -\frac{dm_R}{m_R} u_{ef} - g dt$$

$$\Delta v_R = u_{ef} \text{Ln} \left(\frac{m_{R0}}{m_R} \right) - gt$$

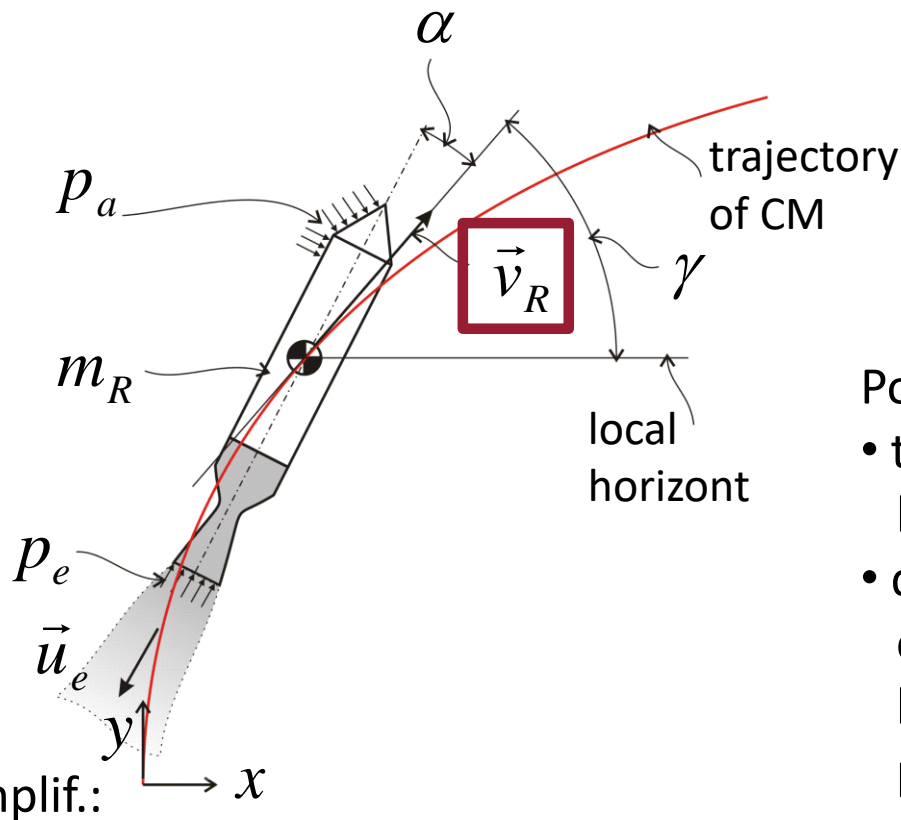
$$m_R \frac{dv_R}{dt} = -\frac{dm_R}{dt} u_{ef} - m_R g$$

$$m_R \frac{d\vec{v}_R}{dt} = \frac{dm_R}{dt} \vec{u}_{ef} + m_R \vec{g}$$

3. Performance of Rocket Vehicle

Launch Flight Mechanics

- Forces act on rocket



The flight of rocket has 2 main phases:

1. powered phase
2. unpowered phase

Powered phase:

- trajectory of vehicle from launch pad to burnout point
- during the phase, guidance system control the trajectory – vehicle at burnout point should have prescribed position and velocity

- all forces act on the same plane
- Earth is inertial frame of reference

Launch Flight Mechanics

- [illegible]

flight path angle – angle between
local horizont and velocity vector

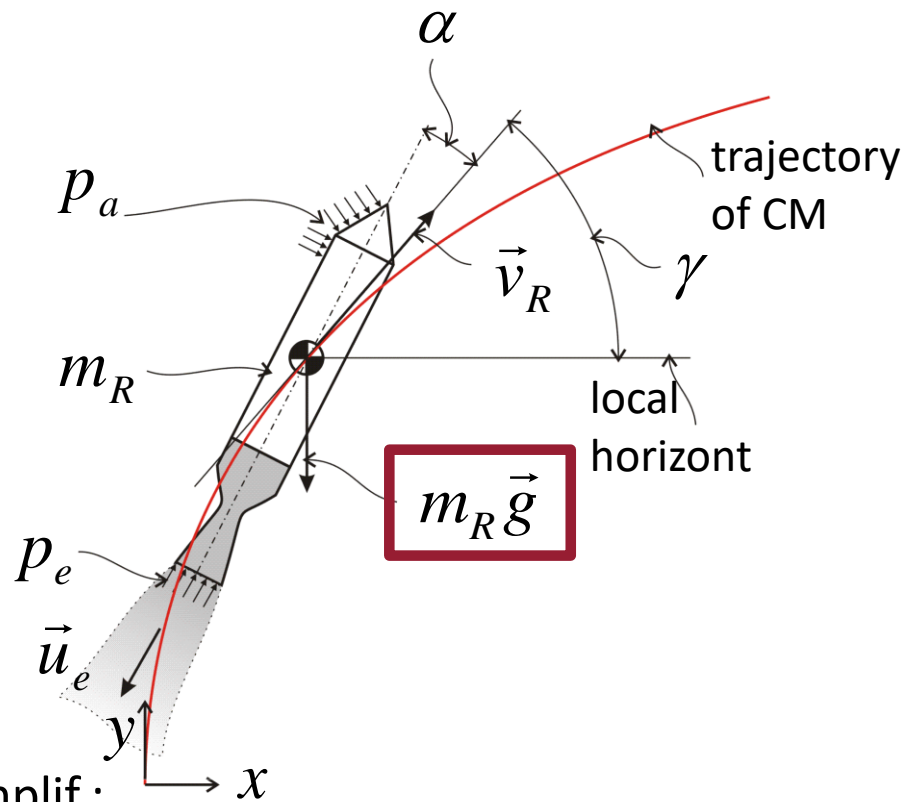
Simplif.:

- all forces act on the same plane
- Earth is inertial frame of reference

3. Performance of Rocket Vehicle

Launch Flight Mechanics

- Forces act on rocket



Three forces act on rocket at each instant:

1. gravitational force – applied at the CM $m_R \vec{g}$



it is function of vertical location of rocket

mass of rocket is function of propellant mass flow



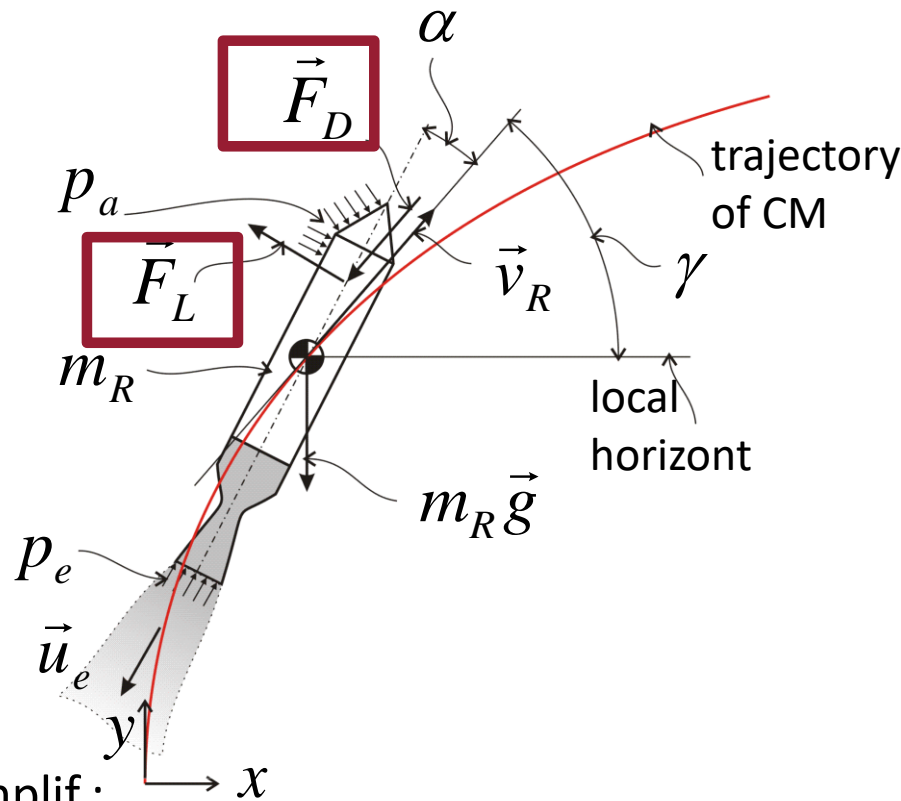
equation of propellant mass flow

- all forces act on the same plane
- Earth is inertial frame of reference

3. Performance of Rocket Vehicle

Launch Flight Mechanics

- Forces act on rocket



Three forces act on rocket at each instant:

- gravitational force – applied at the CM $m_R \vec{g}$
- aerodynamic force – applied at aerodynamic center and can be decomposed into:
 - drag force \vec{F}_D
 - lift force \vec{F}_L



they are function of vertical location and attitude of rocket

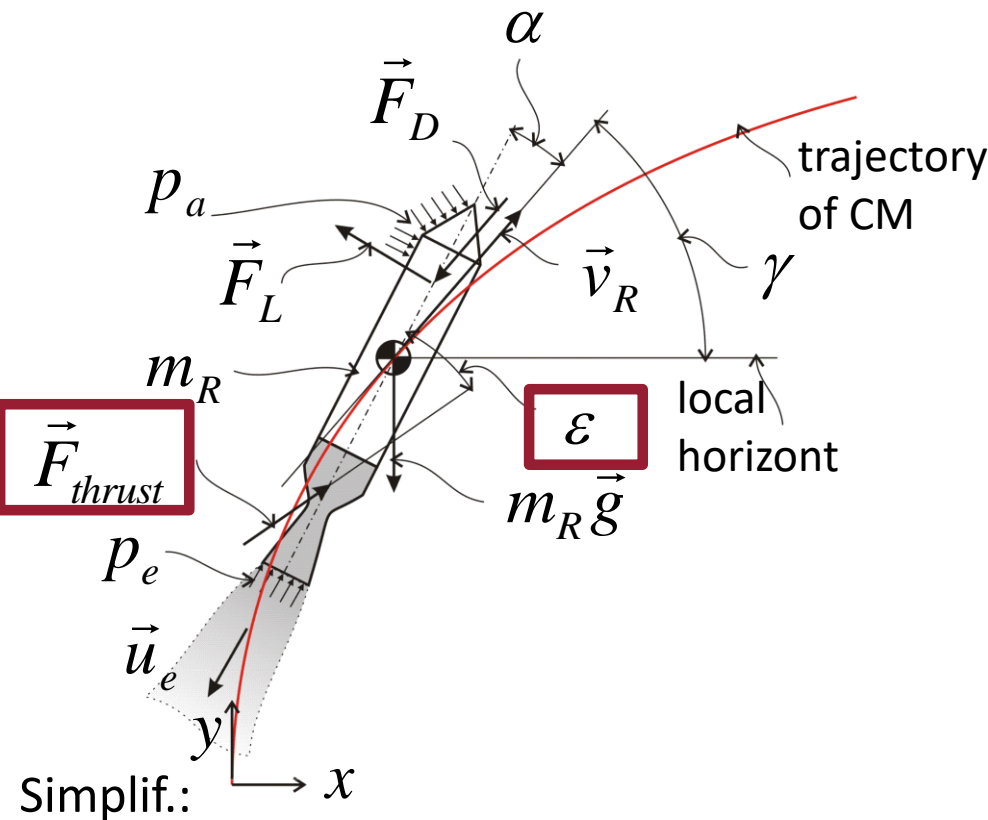
Simplif.:

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 - lift force \vec{F}_L
- thrust force \vec{F}_{thrust}



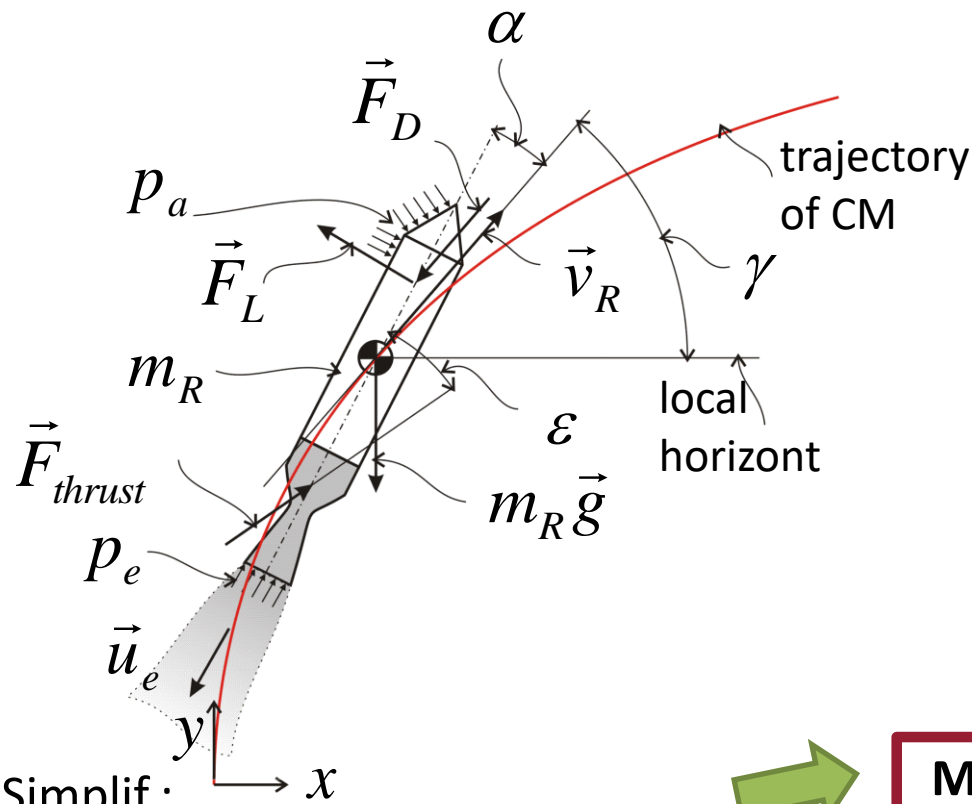
magnitude and direction can be controlled

- Simplif.:
- all forces act on the same plane
 - Earth is inertial frame of reference

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 - lift force \vec{F}_L
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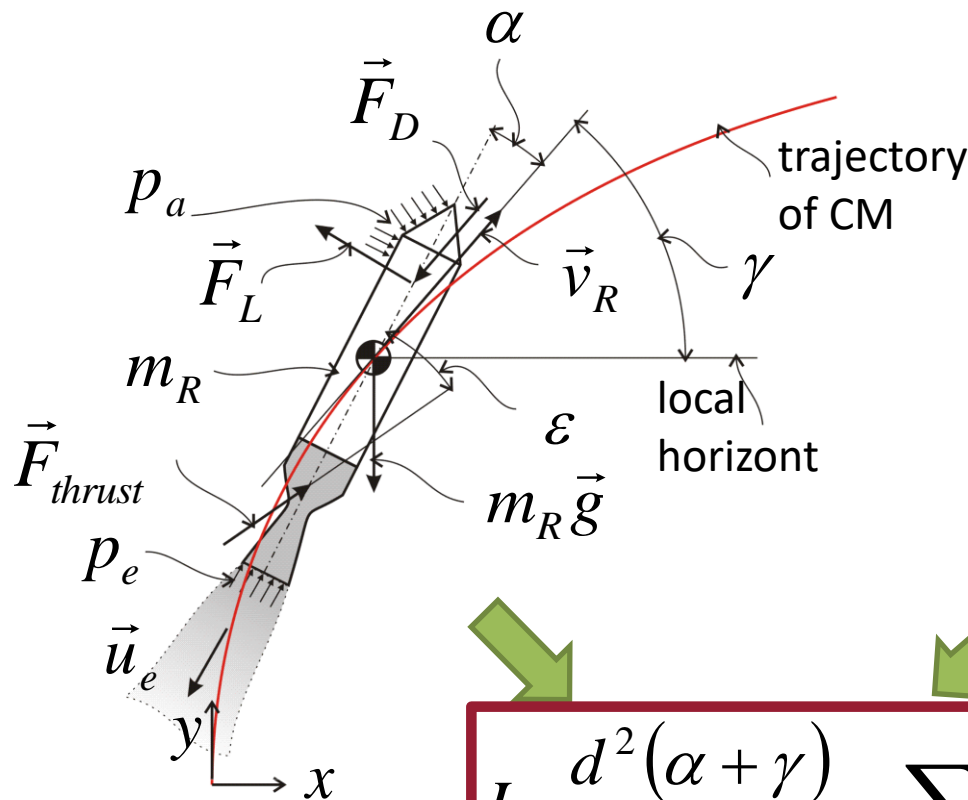
Motion of vehicle in 2D:

- translation motion of Center of Mass
- relative rotation motion around the CM

3. Performance of Rocket Vehicle

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Three forces act on rocket at each instant:

1. gravitational force – applied at the CM $m_R \vec{g}$
2. aerodynamic force – applied at aerodynamic center and can be decomposed into:
 - drag force \vec{F}_D
 - lift force \vec{F}_L
3. thrust force \vec{F}_{thrust}

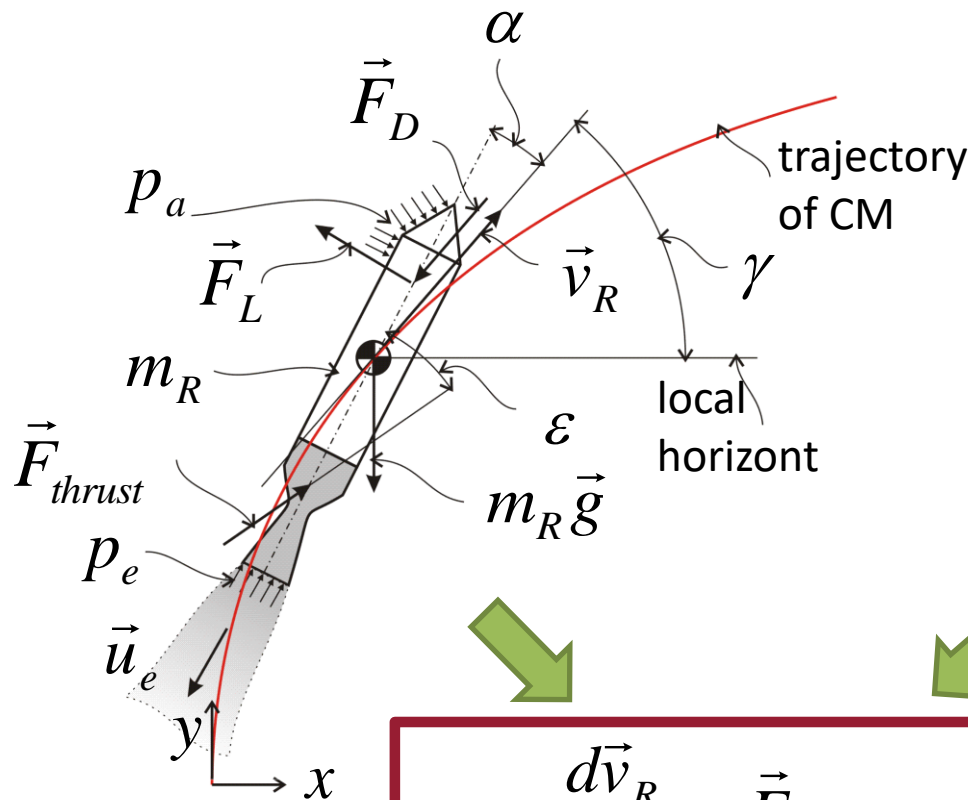
$$I_R \frac{d^2(\alpha + \gamma)}{dt^2} = \sum_i M_{Fi}$$

dynamic equations:
relative rotation motion
around the CM

3. Performance of Rocket Vehicle

Launch Flight Mechanics

- Forces act on rocket



Three forces act on rocket at each instant:

1. gravitational force – applied at the CM $m_R \vec{g}$
2. aerodynamic force – applied at aerodynamic center and can be decomposed into:
 - drag force \vec{F}_D
 - lift force \vec{F}_L
3. thrust force \vec{F}_{thrust}

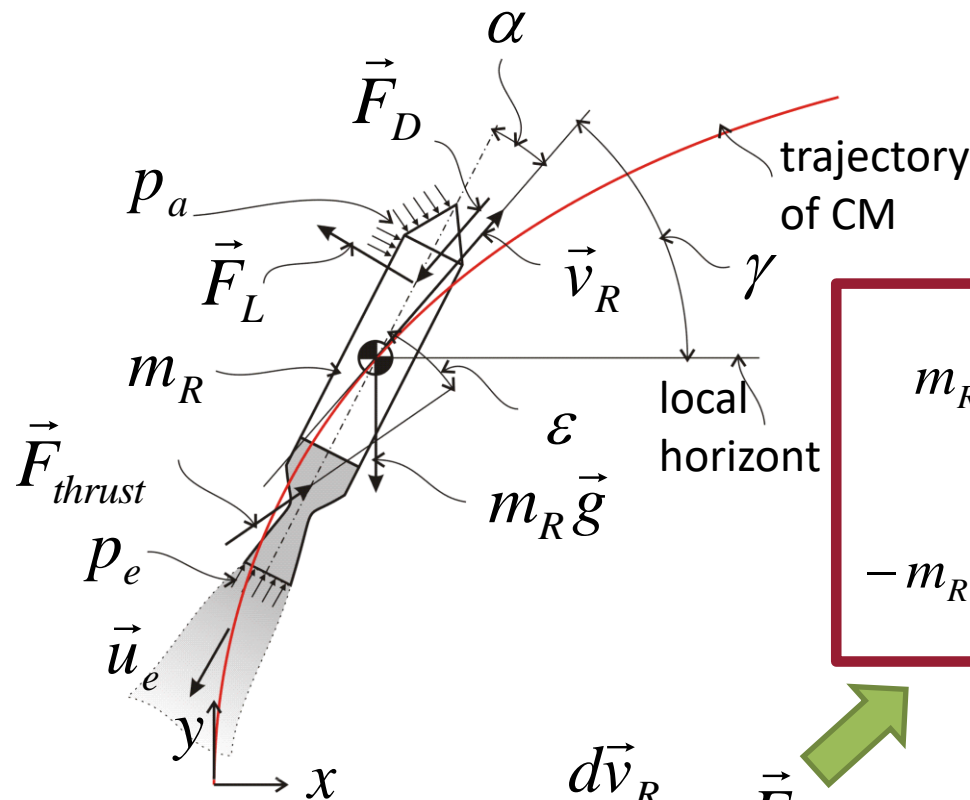
$$m_R \frac{d\vec{v}_R}{dt} = \vec{F}_{thrust} + m_R \vec{g} + \vec{F}_D + \vec{F}_L$$

dynamic equations:
translation motion
of the CM

3. Performance of Rocket Vehicle

Launch Flight Mechanics

- Forces act on rocket



tangent and normal decomposition

$$m_R \frac{dv_R}{dt} = F_{thrust} \cos(\varepsilon - \alpha) - m_R g \sin \gamma - F_D$$

$$-m_R v_R \frac{d\gamma}{dt} = F_{thrust} \sin(\varepsilon - \alpha) + m_R g \cos \gamma - F_L$$

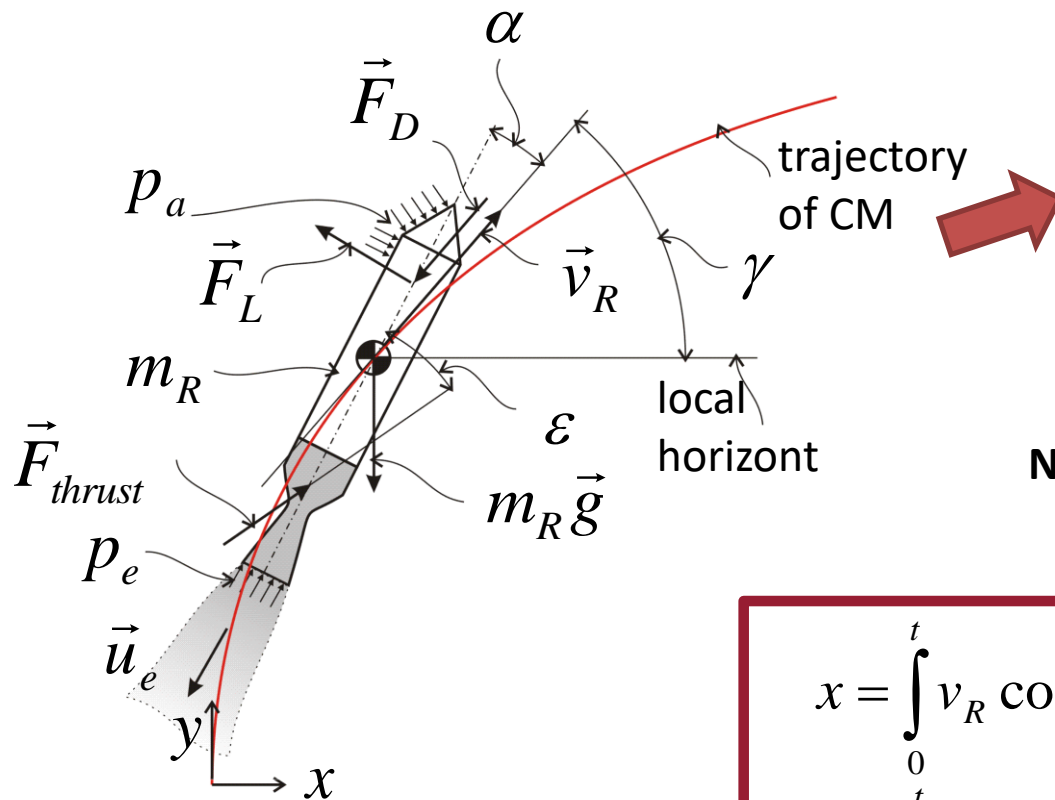
$$m_R \frac{d\vec{v}_R}{dt} = \vec{F}_{thrust} + m_R \vec{g} + \vec{F}_D + \vec{F}_L$$

dynamic equations:
translation motion
of the CM

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Equations of rocket motion:

- dynamic equations
- kinematic equations
- equation of propellant mass flow

Numerical solution of system of ODE

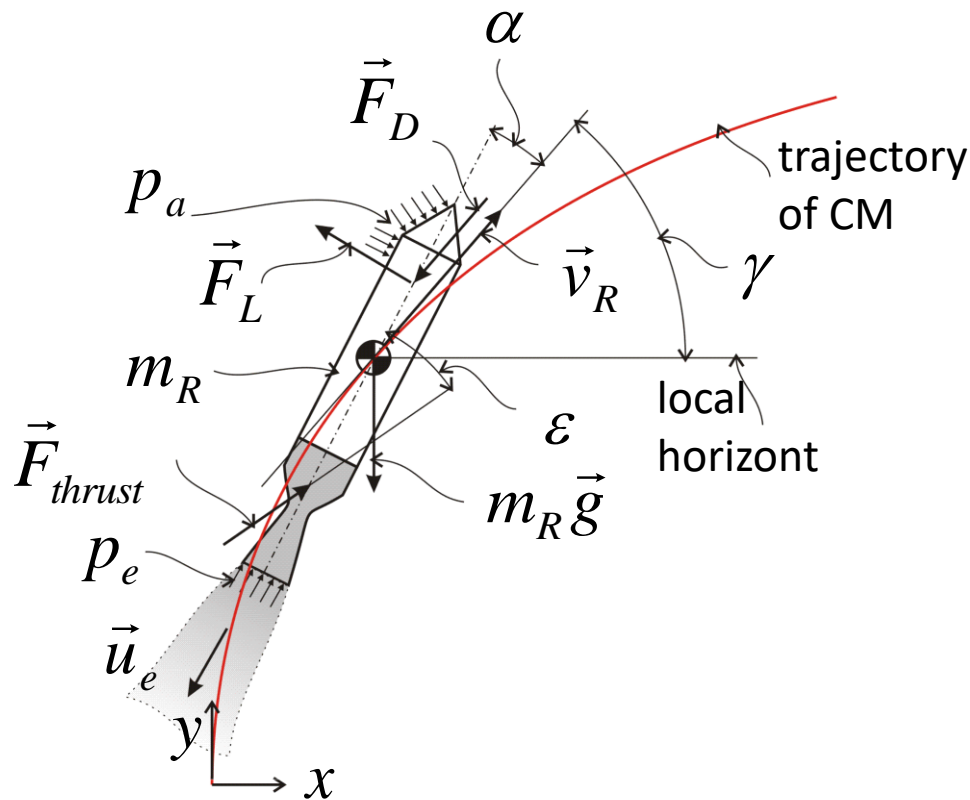
$$x = \int_0^t v_R \cos \gamma dt$$
$$y = \int_0^t v_R \sin \gamma dt$$

kinematic equations:
vertical and horizontal
distance

3. Performance of Rocket Vehicle

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Gravity turn:

- gravity turn trajectory – change of flight angle due to gravity
- only thrust and gravity is considered
- angle of attack is zero
- thrust is in axis of rocket



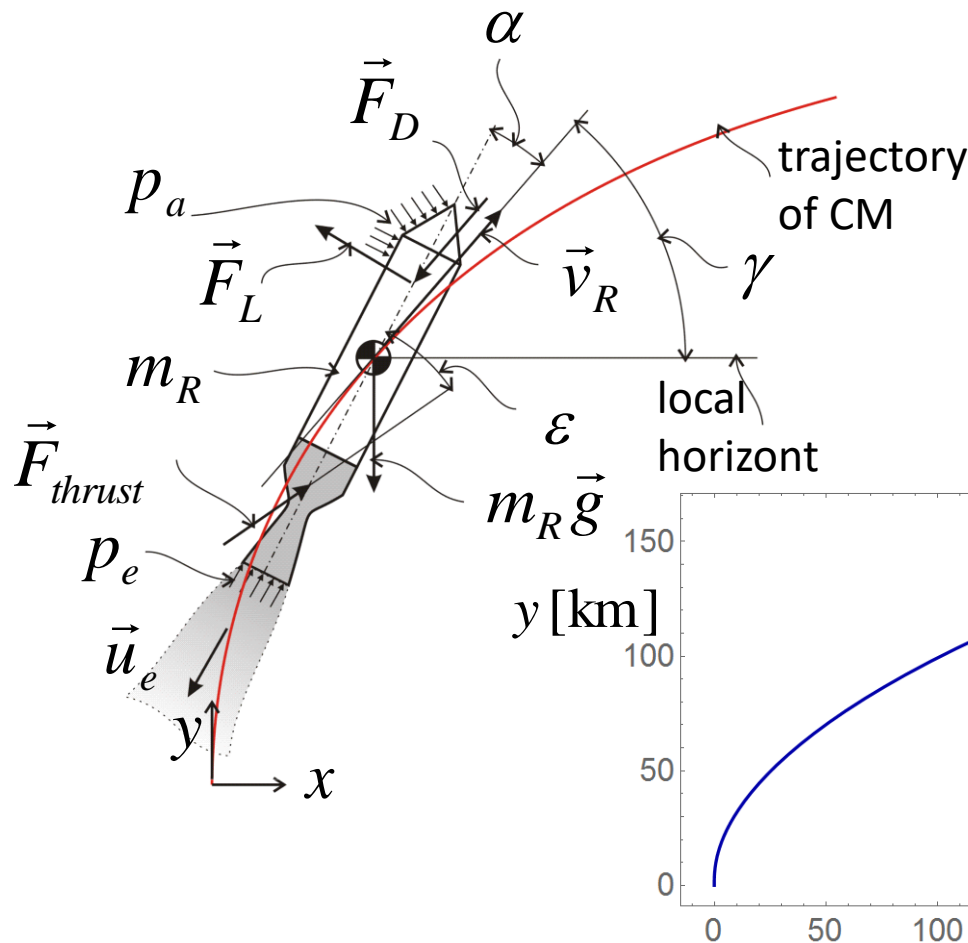
$$m_R \frac{dv_R}{dt} = F_{thrust} - m_R g \sin \gamma$$

$$v_R \frac{d\gamma}{dt} = -g \cos \gamma$$

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Gravity turn:

- initial mass 90 t, propellant is 80% of mass with flow 250 kg/s effective velocity 4000 m/s
- in altitude 1 km, flight angle is changed to 89.85°

$$m_R \frac{dv_R}{dt} = F_{thrust} - m_R g \sin \gamma$$

$$v_R \frac{d\gamma}{dt} = -g \cos \gamma$$

$$\frac{dx}{dt} = v_R \cos \gamma$$

$$\frac{dy}{dt} = v_R \sin \gamma$$

